Electrical control of the chemical bonding of fluorine on graphene

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We study the electronic structure of diluted F atoms chemisorbed on graphene using density functional theory calculations. We show that the nature of the chemical bonding of a F atom adsorbed on top of a C atom in graphene strongly depends on carrier doping. In neutral samples the F impurities induce a sp^3 -like bonding of the C atom below, generating a local distortion of the hexagonal lattice. As the graphene is electron-doped, the C atom retracts back to the graphene plane and for high doping (10^{14} cm⁻²) its electronic structure corresponds to a nearly pure sp^2 configuration. We interpret this sp^3 - sp^2 doping-induced crossover in terms of a simple tight-binding model and discuss the physical consequences of this change.

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The unusual electronic and mechanical properties of graphene, which make it an excellent candidate to build nanoscale electronic devices, have generated intense and exciting activity during the last few years. 1-3 This two-dimensional (2D) honeycomb lattice of C atoms is a prototype of the C-C sp^2 binding. The cohesion and structural stability of the lattice are mainly due to the σ bonds, while the p_{τ} orbitals determine its low-energy electronic properties. The anomalous charge transport in pure graphene is a consequence of the chiral nature of the low-energy excitations, which correspond to massless Dirac fermions,³ and its linear dispersion relation close to the Fermi energy. This leads to a strong suppression of the backscattering processes which, together with the high purity inherent to the samples, is responsible for the long electron mean-free path. Clean graphene samples can be grown because substitutional impurities, vacancies, and other structural defects are quite rare due to the stability of the strong σ bonds. Therefore, one of the main sources of electron scattering is the presence of adsorbed impurities. For this reason, the control and manipulation of adatoms is viewed as a possible route to modify the electronic properties and to create the appropriate conditions for charge and spin transport.

The problem of adatoms on graphene has been theoretically and experimentally studied by several groups. 4-13 It is known that impurities like H or F are adsorbed on top of a C atom—from hereon denoted as C₀—and induce a distortion of the flat honeycomb lattice. This distortion consists of a puckering of the C₀ carbon as its electronic configuration changes from sp^2 to another with a more sp^3 character (see Fig. 1). The behavior of H and F impurities on neutral graphene presents some quantitative and qualitative differences: On the one hand, the charge transfer is larger and the graphene distortion smaller for F impurities. On the other hand, there are strong indications that H creates a magnetic defect while no magnetism is obtained with F. However, spin-related effects are to be expected even in the case of F impurities as the local distortion of the 2D plane leads, like in carbon nanotubes, 14,15 to spin-orbit coupling."

Despite this intense activity, little is known about the effect of graphene's doping on the character of the bonding of atoms and molecules. Here we show that carrier doping

determines the structure of the defect induced by F impurities and the nature of the chemical bond. With electron doping, the distortion of the lattice decreases and the F-C₀ distance increases. Notably, for large-enough doping, the F-C bonding acquires an ionic character. This effect provides an opportunity to study, in a controllable and continuous way, the evolution from ioniclike (sp^2 configuration of C₀) to covalentlike (sp^3 configuration of C₀) bonding of adatoms on graphene. Hole doping, as opposed to electron doping, results in a moderate increase of the lattice distortion. These changes have important consequences on the local spectroscopic properties of the adatom and its surrounding atoms and will certainly modify the charge and spin transport properties of the system.

We consider a graphene layer with a unit cell of 60 C atoms and one F atom (a 1.6% concentration of F atoms). A schematic side view of the distortion induced by the F atom is shown in Fig. 1(b). Figures 1(c) and 1(d) show the partial density of states (PDOS) of an undoped graphene layer obtained using density functional theory (DFT). ¹⁶ These results make evident that there is a large hybridization between the F and the C_0 atom—the strong broadening of the F's atomic levels is consistent with a covalent bonding. We also note that there is a narrow resonance close to the Fermi energy, similar to what is obtained for H. However, in the case of F, as opposed to the H case, this vacancylike state does not lead to the appearance of magnetic order. ²¹

For a quantitative analysis of the distortion, we follow Ref. 11 and relate the angles between the bonds with the sp character of the C_0 orbitals. In terms of its s and p_z atomic orbitals, the hybrid π orbital can be written as $|\pi\rangle = A|s\rangle + \sqrt{1-A^2}|p_z\rangle$, where A=0 and $A=\frac{1}{2}$ correspond to the sp^2 and sp^3 configurations, respectively. A local basis includes the σ orbitals that are also parameterized by the same constant A.¹¹ The angle θ between the σ bonds and the z axis, perpendicular to the graphene plane, is given by $\cos\theta = -A/\sqrt{2+A^2}$. This expression interpolates between $\theta=90^\circ$ for the sp^2 and $\theta=109.47^\circ$ for the sp^3 configurations. Our results for the case of diluted F atoms on neutral graphene give $\theta=102.5^\circ$ and A=0.314. These values indicate a dominant sp^3 character of the bonds of the C_0 carbon atom. They are in qualitative agreement with previous results obtained with a

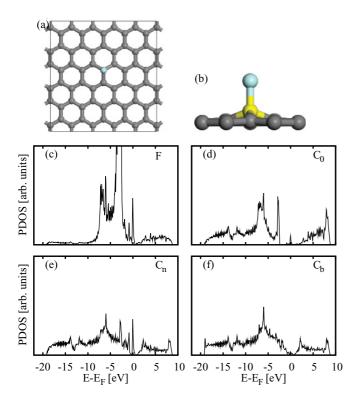


FIG. 1. (Color online) (a) Unit cell used in our DFT calculations (60 C atoms). (b) schematic lateral view of the defect showing the local puckering of the graphene sheet caused by the adatom. Partial density of states (PDOS) for the undoped case projected on the F atom (c), the C_0 atom (d), a C_0 -nearest neighbor C atom, C_n (e), and a "bulk" C_b atom (f).

high F concentration but are smaller than the distortion found in F-graphene. ²²

To simulate electron and hole doping we add or remove electrons from the unit cell. The extra charge is compensated with a uniform charged background. The results obtained by adding one and two electrons per DFT unit cell, corresponding to dopings of the order of $\sim 10^{14}$ cm⁻², are shown in Fig. 2. The F's PDOS shows an increasing narrowing of the resonances as the doping is increased while, at the same time, the C_0 's PDOS becomes more similar to the one corresponding to a "bulk" carbon atom (C_b). This indicates an important reduction of the hybridization parameter V as a consequence of the new structure of the defect in the doped system. Namely, with electron-doping the C_0 carbon atom retracts back to the graphene plane and the F-C₀ distance increases. This change is accompanied by an increase of the charge transfer to the F atom. For the largest doping studied here, two added electrons per DFT unit cell, we obtain $\theta = 91.9^{\circ}$, which gives A = 0.047. These values correspond to an almost pure sp^2 configuration of C_0 . On the contrary, with hole-doping, the distortion increases and the F-C₀ distance decreases, indicating an increase of the sp^3 character of the bonding of the C_0 carbon atom (see Table I for details).

Hence, our results indicate that the effect of the F adatom on graphene can be strongly modified by external gates, a simple way to change doping, offering a new route to control graphene's transport properties. For instance, as mentioned above, the local distortion of the lattice intro-

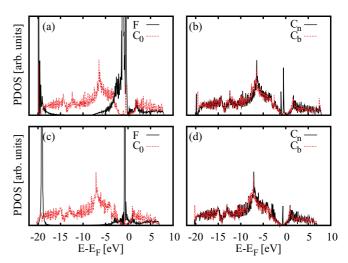


FIG. 2. (Color online) PDOS for electron-doped graphene with one (a),(b) and two (c),(d) additional electrons per DFT unit cell. Labels are the same as in Fig. 1. Notice the reduction of the hybridization between the F atom and the graphene sheet as the doping increases.

duces a spin-orbit coupling. As discussed in Ref. 11, the effective spin-orbit coupling is roughly given by $\Delta_{so} = -\Delta_{so}^{at}A\sqrt{3(1-A^2)}$, where Δ_{so}^{at} is the atomic spin-orbit coupling of the C atoms. Then, our results show that, as the system is doped with electrons, Δ_{so} changes by a factor \sim 6. Such a change should be clearly detected in spin-dependent transport measurements.

In order to understand the physics underlying this sp^3 - sp^2 crossover we use a minimal tight-binding (TB) model that captures the relevant features. The π bands of the graphene plane are described by a Hubbard Hamiltonian $H_{\text{graph}} = -\sum_{\langle i,j\rangle\sigma} t_{ij}\,c_{i\sigma}^{\dagger}c_{j\sigma} + U_c\sum_i(\hat{n}_{i\uparrow}-\frac{1}{2})(\hat{n}_{i\downarrow}-\frac{1}{2})$, where $c_{i\sigma}^{\dagger}$ creates an electron with spin σ at site i of the graphene lattice, $\hat{n}_{i\sigma} = c_{i\sigma}^{\dagger}c_{i\sigma}$, and the first sum runs over nearest neighbors. The F impurity is bounded to the C_0 atom, located at site i=0 and represented by the relevant $|p_z\rangle$ orbital, and is described by $H_f = \sum_{\sigma} \varepsilon_f \, f_{\sigma}^{\dagger} f_{\sigma} + U_f \, f_{\uparrow}^{\dagger} f_{\uparrow} f_{\uparrow}^{\dagger} f_{\downarrow}$, where f_{σ}^{\dagger} creates an electron with spin σ and energy ε_f at the relevant orbital of the impurity and U_f is the intra-atomic Coulomb repulsion. The impurity-graphene interaction includes a one-body hybridization V, a distortion-induced shift of the C_0

TABLE I. Impurity-induced graphene distortion for different numbers of electrons added (or removed) from the unit cell. Here ϕ is the angle formed by the C_0 atom and two of its nearest-neighbors (C_n) ; θ is the angle of the C_0 - C_n σ bonds and the normal to the graphene plane; d_{F-C_0} and $d_{C_0-C_n}$ are the F- C_0 and C_0 - C_n bond lengths, respectively.

| Electrons | ϕ | θ | $d_{\text{F-C}_0}$ (Å) | $d_{\mathcal{C}_0\text{-}\mathcal{C}_n}$ (Å) |
|-----------|--------|----------|------------------------|--|
| -2 | 115.1 | 103.1 | 1.47 | 1.49 |
| Neutral | 115.5 | 102.5 | 1.57 | 1.48 |
| +1 | 118.8 | 96.3 | 2.00 | 1.43 |
| +2 | 119.9 | 91.9 | 2.50 | 1.42 |

energy Δ , the interaction with the image charge $V_{\rm im}$, ²³ and a local interaction interaction U_{f0} ,

$$H_{\text{int}} = \Delta \,\hat{n}_0 - V \sum_{\sigma} (c_{0\sigma}^{\dagger} f_{\sigma} + f_{\sigma}^{\dagger} c_{0\sigma}) - V_{\text{im}} (\hat{n}_f - 1)^2 + U_{f0} (\hat{n}_f - 1) (\hat{n}_0 - 1), \quad (1)$$

with $\hat{n}_0 = c_{0\uparrow}^\dagger c_{0\uparrow} + c_{0\downarrow}^\dagger c_{0\downarrow}$ and $\hat{n}_f = f_\uparrow^\dagger f_\uparrow + f_\downarrow^\dagger f_\downarrow$. Note that the effect of $V_{\rm im}$ is to renormalize both the orbital energy, $\tilde{\varepsilon}_f =$ $\varepsilon_f + V_{\text{im}}$, and the intra-atomic repulsion, $\tilde{U}_f = U_f - 2V_{\text{im}}$, favoring the charging of the F atom. The U_{f0} term, instead, takes into account the discharge induced on the Co atom upon charging of the F atom. In what follows, guided by the DFT calculations, we take t_{ij} with $i, j \neq 0$ equal to t = 2.8 eV and include the reduction of $t_{0j} = t_{j0}$ induced by the distortion in the sp^3 case. The energy shift of C_0 is given by $\Delta \simeq A^2 (\varepsilon_s - \varepsilon_p)^{11}$, where $\varepsilon_s \sim -8$ eV and $\varepsilon_p \equiv 0$ are the energies of the s and p carbon orbitals, respectively. Since the parameters t_{0j} , Δ , V, V_{im} , and U_{f0} depend on the distortion of the lattice, and to minimize the number of free parameters of the model, we consider two frozen configurations corresponding to the two limiting cases presented above: (i) the distorted lattice of Fig. 1 with $\Delta = -0.3t$, $t_{0j} = 0.7t$, and V = 1.4t;²⁴ (ii) the undistorted lattice with $\Delta = 0$, $t_{0j} = t$, and V = 0.4t the smaller value of the hybridization V reflects the larger F-C₀ distance shown in Table I. In the latter case, where the graphene sheet is essentially unalterated, we can consider that the situation is reminiscent of the classical problem of a point charge in front of a conducting plate and take, as a simple approximation, $V_{\rm im}^{sp^2} = \gamma e^2/4d_{\rm F-C_0}$, with $\gamma \sim 0.8.^{23,25}$ On the contrary, in the sp^3 case, the F is covalently bounded to C_0 and then one expects a stronger local screening of the F's excess charge. That is, we expect the term U_{f0} to be dominant. For the sake of simplicity, and to avoid a double counting of the Coulomb interaction, we take $V_{\text{im}}^{sp^3} = 0$ and choose

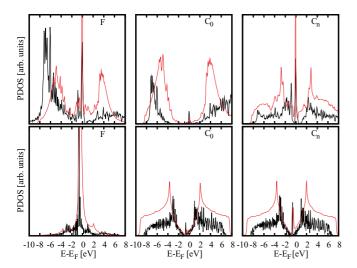


FIG. 3. (Color online) Partial density of states projected onto the p_z atomic orbitals for the F, C₀, and C_n atoms for undoped graphene (top panels) and for the electron doped (+2) case (bottom panels). The thick lines correspond to the DFT results while the thin lines are the results of the tight-binding model with the parameters discussed in the text. The unit cell contains 60 C atoms in both calculations.

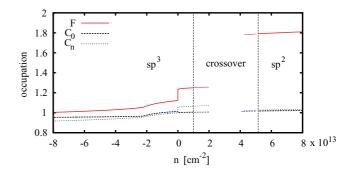


FIG. 4. (Color online) Occupation of the F, C_0 , and C_n atoms for a simplified tight-binding model in the sp^3 and sp^2 configurations, as a function of the carrier density. Note that the charge of the F atom is higher in the sp^2 configuration.

 $U_{f0}^{sp^3} \simeq V_{\rm im}^{sp^2}$. Similarly, we take $U_{f0}^{sp^2} = 0$. In the following, we solved the Hamiltonian $H = H_{\rm graph} + H_f + H_{\rm int}$ in the Hartree-Fock approximation. From comparison to the DFT results for the PDOS of the F, C₀, and C_n atoms we estimate the effective parameters $U_f \sim 2.5t$ and $\varepsilon_f \sim -1.8t$ while we set $U_c = t$.

Figure 3 shows a comparison of the DFT and TB results for the PDOS projected onto the p_z orbitals. While the latter is not expected to give an accurate description of the former, there is a good qualitative agreement between both methods for the general features of the PDOS, which gives support to the chosen effective TB parameters. Figures 4(a) and 4(b) show the occupation of the F, C_0 , and one of the C_n atoms as a function of the graphene's electron density for the sp^3 and sp^2 configurations, respectively. In the sp^3 case the F, C₀, and C_n atoms are moderately (dis-)charged as the graphene is doped with electrons (holes); note that C₀ is slightly discharged when F is charged. This differs from the sp^2 case where the F atom is heavily charged while the C_0 and C_n atoms remain essentially uncharged. This is consistent with the DFT results, where a significant increase of the charge of the F atom is observed in the electron doped case. In addition to that, the DFT shows a small discharge of the C₀ atom, which in our model can be obtained by including a small $U_{f0}^{sp^2} \simeq 0.1t$.

It is worthy of mention that the simplified model also helps to understand the absence of magnetism in the DFT calculations. This is so because the main features can be described by an even simpler model where interactions are only included on the F, C_0 , and C_n atoms. In that case, the system behaves as an effective Anderson impurity and then the absence of magnetism can be understood in terms of its effective parameters.²¹

The mechanism behind this covalent-ionic crossover is the competition between the hydridization of the F and C_0 atoms and both the distortion of the lattice and the Coulomb interaction. In the neutral case, the sp^3 configuration is more stable since the energy gain due to F-C₀ hybridization overcomes the energy lost due to the lattice distortion. From DFT calculations, we estimate the latter to be $\sim 0.4t$. Since it mainly comes from the σ bands this contribution is not included in our TB model; we assume it to be doping independent. Upon doping, the increase in the occupation of both the F and the C_0 (enhanced in the latter case by the energy shift Δ) increases

the energy due to the local Coulomb interaction U_{f0} while, at the same time, it leads to a reduction of the gain due to V ("antibonding" states become occupied). On the contrary, the sp^2 configuration, which does not pay the distortion energy, gains some energy due to $V_{\rm im}$ when the F is charged. To verify this scenario, we calculate the energy difference between the two configurations, 23 $\Delta E = E_{sp^2} - E_{sp^3} = \int \langle \frac{\partial \hat{H}}{\partial \lambda} \rangle d\lambda$, where λ is a parameter that linearly interpolates all the parameters that change between the two configurations (Δ , t_{oj} , V, V_{im} , and U_{f0}) and $\langle \cdots \rangle$ is the average value in the Hartree-Fock state. Within our TB model, we found that $\Delta E \sim 0.4t$ is roughly constant for hole doping but it drops as the graphene sheet is doped with electrons. While this Hartree-Fock result is sensitive to the chosen parameters and even though this is an oversimplified picture of the real process where the local configuration is progressively distorted by the doping (see Table I), it clearly gives further support to our interpretation the DFT results.

In summary we have shown that the nature of the F-C₀ bonding changes with doping from covalent for hole-doping

to ionic for electron-doping. This is followed by a change of the structure of the defect $(sp^3 \rightarrow sp^2)$ with an important distortion of the lattice for the hole-doped system to an almost undistorted lattice for electron-doped graphene. This structural change can serve as a new knob to control graphene's electronic properties. As mentioned before, the properties of several elements and molecules on neutral graphene have been extensively studied.^{7,10} Our results point out the relevance of doping and call for a more exhaustive analysis of other elements, beyond F, as they could present similar effects.

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