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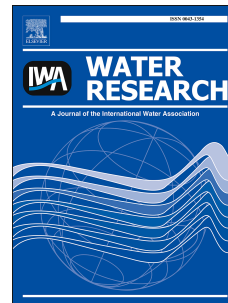
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High-resolution mapping and modeling of anammox recovery from recurrent oxygen exposure

D. Seuntjens, J.M. Carvajal-Arroyo, M. Ruopp, P. Bunse, C.P. De Mulder, S. Lochmatter, S. Agrawal, N. Boon, S. Lackner, S.E. Vlaeminck



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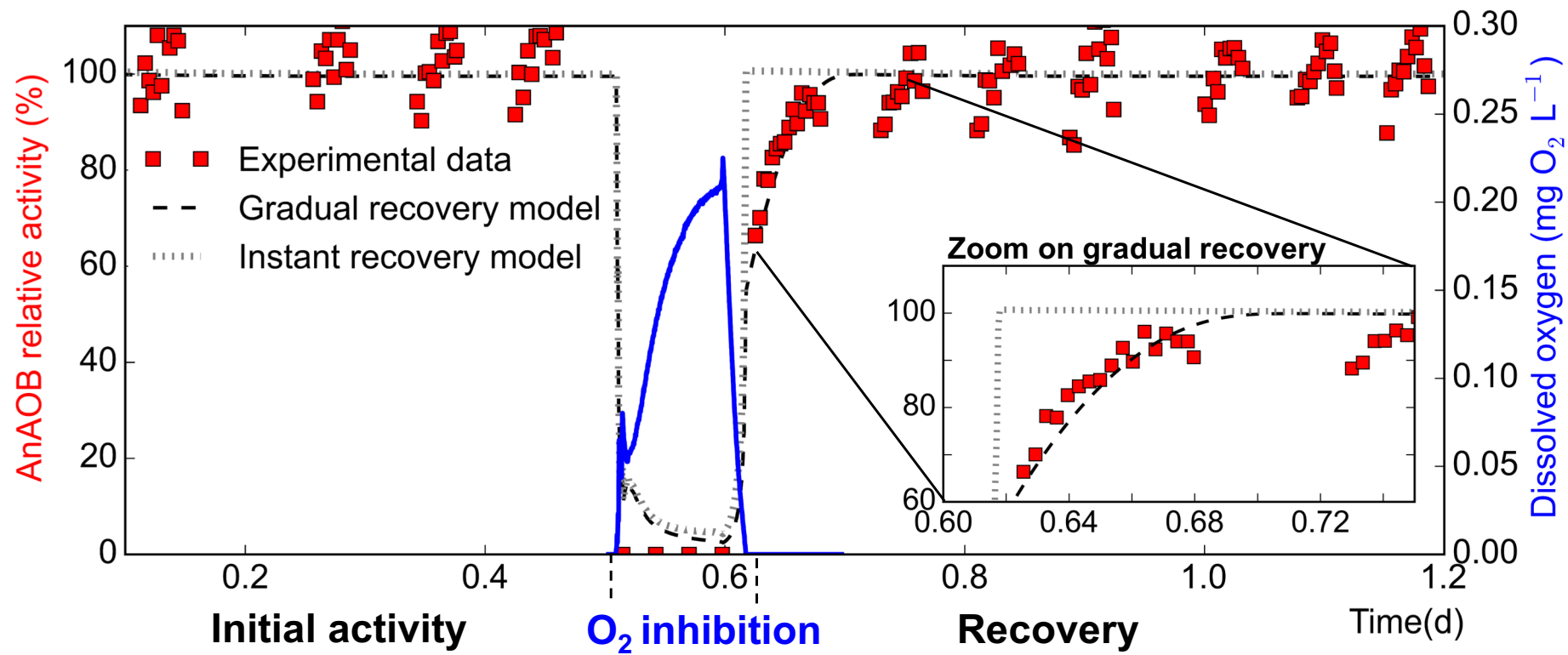
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1 **High-resolution mapping and modeling of anammox recovery**
2 **from recurrent oxygen exposure**

3 Seuntjens D.¹, Carvajal-Arroyo J.M.¹, Ruopp M.², Bunse P.², De Mulder C.P.³, Lochmatter S.¹,
4 Agrawal S.², Boon N.¹, Lackner S.^{2,§} and Vlaeminck S.E.^{1,4,§,*}

5

6 ¹ Center for Microbial Ecology and Technology (CMET), Ghent University, Gent, Belgium

7 ² Technische Universität Darmstadt, Institute IWAR, Chair of Wastewater Engineering, Darmstadt, Germany

8 ³ Biomath, Ghent University, Gent, Belgium

9 ⁴ Research group of Sustainable Energy, Air and Water Technology, University of Antwerp, Antwerpen, Belgium

10

11 § Equal contribution as senior authors

12 * Corresponding author: siegfried.vlaeminck@uantwerpen.be

13

14 **Keywords**

Monod, partial nitrification/anammox, inhibition, intermittent aeration, energy-positive, sewage treatment

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16 **Abstract**

17 Oxygen inhibits anammox, a bioconversion executed by anoxic ammonium oxidizing
18 bacteria (AnAOB). Nonetheless, oxygen is mostly found in the proximity of AnAOB in
19 nitrogen removal applications, being a substrate for nitrification. The experiments
20 performed to date were mostly limited to batch activity tests where AnAOB activity is
21 estimated during oxygen exposure. However, little attention has been paid to the
22 recovery and reversibility of activity following aerobic conditions, of direct relevance for
23 bioreactor operation. In this work, anoxic and autotrophic reactor cultivation at 20°C
24 yielded an enriched microbial community in AnAOB, consisting for 75% of a member of
25 the genus *Brocadia*. High-resolution kinetic data were obtained with online ammonium
26 measurements and further processed with a newly developed Python data pipeline. The
27 experimentally obtained AnAOB response showed complete inhibition until micro-
28 aerobic conditions were reached again ($<0.02 \text{ mg O}_2 \text{ L}^{-1}$). After oxygen inhibition,
29 AnAOB recovered gradually, with recovery times of 5-37h to reach a steady-state
30 activity, dependent on the perceived inhibition. The recovery immediately after inhibition
31 was lowest when exposed to higher oxygen concentrations (range: $0.5\text{-}8 \text{ mg O}_2 \text{ L}^{-1}$) with
32 long contact times (range: 9-24h). The experimental data did not fit well with a
33 conventional 'instant recovery' Monod-type inhibition model. Yet, the fit greatly improved
34 by incorporating a dynamic growth rate formula accurately describing gradual activity
35 recovery. With the upgraded model, long-term kinetic simulations for partial
36 nitrification/anammox (PN/A) with intermittent aeration showed a decrease in growth rate
37 compared to the instant recovery mode. These results indicate that recovery of AnAOB

38 after oxygen exposure was previously overlooked. It is recommended to account for this
39 effect in the intensification of partial nitritation/anammox.

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40 1. Introduction

41 Oxygen inhibits anoxic ammonium oxidation (anammox), a bioconversion executed by anoxic
42 ammonium oxidizing bacteria (AnAOB). Nonetheless, oxygen is mostly found in their cell-
43 proximity in biotechnological applications. Partial nitrification/anammox (PN/A) is a most cost-
44 effective technology to remove ammonium from wastewater autotrophically, i.e. without organic
45 carbon, producing mainly nitrogen gas and some nitrate. It is already successfully implemented
46 to treat ammonium-rich streams ($500\text{-}1000\text{ mg N L}^{-1}$) at higher temperatures ($30\text{-}35^\circ\text{C}$), as the
47 so-called sidestream PN/A process treating reject water from digested sewage sludge (Lackner
48 et al., 2014). The so-called mainstream application of PN/A is in the waterline of a sewage
49 treatment plant, following a carbon-removal step. This approach is less mature, yet of high
50 interest, as it can enable treatment facilities to become a net producer (rather than consumer) of
51 electricity (Verstraete & Vlaeminck, 2011). Specific challenges for mainstream PN/A relate to its
52 lower temperatures ($8\text{-}25^\circ\text{C}$ in central Europe), lower influent ammonium concentrations ($30\text{-}50$
53 mg N L^{-1}) and higher levels of organics (e.g. biodegradable COD/N around 2). Agrawal et al.
54 (2018) recently proposed a multi-controller approach to manage the complex microbial
55 competition network under these conditions, in essence based on controlling the activity
56 (ON/OFF) and retention (IN/OUT) of the main groups of microbes, promoting specifically
57 aerobic and anoxic ammonium-oxidizing bacteria (AerAOB and AnAOB). The (partial)
58 proliferation of nitrite-oxidizing bacteria (NOB) under mainstream conditions is a typical
59 example of imbalancing the AerAOB-AnAOB tandem, resulting in lower nitrogen removal.
60 Operational strategies suppressing NOB activity included residual ammonium ($>2\text{-}4\text{ mg N L}^{-1}$)
61 and short floccular residence times ($2\text{-}7\text{ d}$), combined with intermittent aeration at higher
62 dissolved oxygen (DO) setpoints e.g. $1.5\text{ mg O}_2\text{ L}^{-1}$ for $0.25\text{-}0.5\text{ h}$ (Han et al., 2016; Malovanyy
63 et al., 2015; Regmi et al., 2014). The potential AnAOB inhibition by these aeration strategies in
64 i.e. granular or biofilms on carriers, was however not yet addressed. It is unclear to which extent

65 oxygen hinders PN/A start-up or recovery after a collapse in practice, but an exploratory
66 modeling effort already revealed that full-oxygen penetration can occur in aggregates of 500 μm
67 diameter starting from 1 $\text{mg O}_2 \text{ L}^{-1}$ (See Figure A.1 - C.1).

68

69 Oxygen inhibition of AnAOB has been described most extensively for marine systems, as well
70 as for some laboratory freshwater cultures. A wide range of inhibitory values were reported for
71 marine systems, with 50% inhibitory concentrations (IC_{50}) ranging from 0.03 - 0.36 $\text{mg O}_2 \text{ L}^{-1}$
72 (See Figure D.1) (Babbin et al., 2014; Dalsgaard et al., 2014; Jensen et al., 2008; Kalvelage et
73 al., 2011). The same holds for laboratory freshwater cultures of *Brocadia anammoxidans* or
74 *Kuenenia stuttgartiensis*, with complete inhibition reported at micro-aerobic levels (<0.04-0.12
75 $\text{mg O}_2 \text{ L}^{-1}$) (Egli et al., 2001; Oshiki et al., 2015; Strous et al., 1997). In contrast, higher IC_{50}
76 were found for floccular cultures for *Brocadia sinica* (2 $\text{mg O}_2 \text{ L}^{-1}$) and *Brocadia caroliensis* (3.8
77 $\text{mg O}_2 \text{ L}^{-1}$) (Carvajal-Arroyo et al., 2013; Oshiki et al., 2011). This variability in reported DO
78 inhibitory levels can be due to inter-genera differences (Oshiki et al., 2015), potential adaptation
79 towards oxygen stress, protection by oxygen consuming bacteria in small aggregates
80 (Dalsgaard et al., 2014), and testing procedures (Dalsgaard et al., 2014) i.e. mixing, aggregate
81 size, etc. Furthermore, most studies done on fresh-water cultures tested an one-time exposure
82 on anoxically enriched AnAOB, potentially influencing kinetics (Carvajal-Arroyo et al., 2013.; Egli
83 et al., 2001; Lotti et al., 2012; Oshiki et al., 2011). Since AnAOB showed adaptation towards
84 oxygen stress in their transcriptome (Yan et al., 2012), tests with recurring oxygen exposure
85 should be performed with AnAOB to mimic real operation conditions and obtain closer-to
86 application kinetics.

87

88 Mathematical models have been widely used to study and optimize the PN/A process under
89 different conditions (Hao et al. 2005, Ni et al. 2014, Terada et al. 2007, Pérez et al. 2014). The
90 oxygen stress on AnAOB was a crucial factor for such modeling efforts, influencing reactor
91 performance e.g. by comparing different aeration patterns (Corbala-Robles et al. 2016). In this
92 case, oxygen inhibition was modeled by an instant recovery Monod inhibition model (Henze et
93 al. 2000). This is in line with the experimental studies so far, where almost all focused-on activity
94 loss during oxygen exposure, which can be described by such model. Yet, only two studies
95 looked at recovery of activity after anoxic conditions were restored (Lotti et al., 2012.; Strous et
96 al., 1997). Both Lotti et al. (2012) and Strous et al. (1997) reported no loss of activity after
97 exposure. As both studies used thick granules with a diameter of 1.1 mm, limited oxygen
98 diffusion through the biofilm could have prevented oxygen inhibition. On the other hand, the low
99 resolution of data, i.e. 6 data points over a period of 2h in the case of Strous et al. (1997), could
100 have disguised short-term recovery effects.

101
102 In this study, the effect of DO-pulse concentration and length on AnAOB activity was mapped,
103 by dosing oxygen into an autotrophically fed anammox reactor. Ammonium profiles were
104 measured online and converted to high-resolution AnAOB activities by a newly developed data
105 pipeline. Representative kinetic data were obtained by minimizing diffusional limitations by using
106 flocs ($\varnothing_{4.3} = 168 \mu\text{m}$), and by applying recurrent oxygen exposure. As the experimental data did
107 not fit the conventionally used 'instant recovery' Monod model, a new model is proposed based
108 on a more gradual recovery of AnAOB activity. This model was used to assess the long-term
109 impact and implications of oxygen exposure in a AnAOB process model.

110 **2. Material and methods**

111 **2.1. Anammox reactor**

112 **2.1.1. Inoculum**

113 For the reactor start-up, the goal was to obtain an inoculum rich in AnAOB with minimum
114 diffusion limitations (small particle size). Hereto the cyclone underflow fraction of a sidestream
115 PN/A treatment facility was sampled (WWTP Nieuwveer, Breda, NL). Then, the fraction highest
116 in AnAOB was further selected by sieving this sludge over a 200 μm sieve, assisted by adding
117 tap water. To minimize diffusional limitations, the recovered from the sieve was then blended (\pm
118 2 min) in a kitchen blender and sieved again with reactor medium lacking in N-substrate over a
119 200 μm sieve. The blended, floccular biomass was retained, while unwanted, larger (in)organic
120 particles were removed.

121

122 **2.1.2. Reactor set-up and SBR operation**

123 The biomass was cultured in a 5.5 L anoxic sequencing batch reactor (SBR) and fed with a
124 synthetic autotrophic medium with ammonium and nitrite as nitrogen sources in a 1/1.3 ratio.
125 The medium was made anoxic through N_2 gas flushing, had a pH of 7.2, and contained per L,
126 0.5 g of $\text{NH}_4^+\text{-N}$ as NH_4Cl , 0.65 g of $\text{NO}_2^-\text{-N}$ as NaNO_2 , 0.5 g NaHCO_3 , 3.9 g HEPES, 0.1 g
127 $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$, 0.2 g $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$, 0.06 g $\text{Na}_2\text{HPO}_4 \cdot 2\text{H}_2\text{O}$ and 1 mL of trace elements solutions
128 A and B (Supplemental information E.1). Stirring occurred with an IKA RW20D (IKA, USA). DO
129 levels and temperature were measured online by an Oxymax COS61D sensor with a response
130 time of 1s (Endress and Hauser, CH), whereas pH, ammonium, and potassium were measured
131 by an IseMax CAS40D sensor with response time <2 min (Endress and Hauser, CH). The

132 signals were sent to a Liquiline CM448 (Endress and Hauser, CH), which transmitted the
133 signals to DaqFactory software (Azeotech, Oregon, USA). This software controlled and
134 monitored the reactor operation and measurements. A typical SBR cycle was as follows. 1.) A
135 new dose of influent (0.23 L over 4 min) was given every time nitrite was depleted, i.e. when
136 ammonium was below a certain residual level and when the rate was below $40 \text{ mg N L}^{-1} \text{ d}^{-1}$. A
137 residual ammonium level $>5 \text{ mg N L}^{-1}$ was chosen to ensure that nitrite was always the limiting
138 substrate. 2.) Similarly, 3 additional doses were given, raising the volume of the SBR from 4.25
139 to 5.17 L. 3.) After nitrite was depleted in the last spike, biomass settled for 30 minutes. 4.)
140 Hereafter, the effluent was pumped out over a period of 7.5 minutes, and the cycle
141 recommenced. The volumetric ammonium removal activity of the reactor determined the influent
142 flow rate fed to the system, resulting in hydraulic retention times varying between 1.7 and 2.5 d
143 over the course of the experiment.

144 **2.1.3. Reactor conditions**

145 The reactor was operated for 120 days, and the reactor conditions were chosen to induce stable
146 and optimal operation, with the added oxygen spikes as only disturbing factor. 1.) The reactor
147 was kept anoxic by N_2 -gas flushing (except during an oxygen spike), and the change of
148 headspace volume due to SBR cycling was compensated by a gas bag. 2.) pH was controlled at
149 pH 7.2 by a 0.02 M HCl solution to compensate for the increase in pH from the AnAOB reaction
150 and stripping when flushing with N_2 gas. Under stable operation, the control of pH had a
151 maximum $\Delta\text{pH} = 0.05$. No limitation for CO_2 was prevalent, because the reactor was buffered
152 well at pH 7.2 (no large pH fluctuations). 3.) The reactor was located in a temperature-controlled
153 room at 20°C (=no thermostat was used). The temperature was stable during all experiments
154 (avg. standard deviation = 0.1°C) except for the experiment on day 18 ($T = 21.5 \pm 0.8$), where
155 the measured activity was corrected with a temperature Arrhenius coefficient $\Theta_T = 1.10$ (Lotti et

156 al., 2014). 4.) Salinity (~43 mM NaCl), ammonium levels (5-40 mg N L⁻¹), nitrite levels (0-35 mg
157 N L⁻¹) and nitrate levels (~100 mg N L⁻¹) were all kept well below reported inhibitory levels for
158 the genus *Brocadia* (Oshiki et al., 2015). 5.) Floccular sludge ($\bar{\varnothing}_{4,3} = 168 \pm 24 \mu\text{m}$) and fast
159 stirring (200 rpm) was used to minimize diffusional limitations. Fast stirring also ensured no
160 biofilm growth on the reactor walls. Sludge was regularly taken out of the reactor for batch
161 activity tests, resulting in a stable biomass concentration of $1.95 \pm 0.08 \text{ g VSS L}^{-1}$ that showed an
162 activity of $1184 \pm 188 \text{ mg NH}_4^+ \text{ g VSS}^{-1} \text{ d}^{-1}$ over the course of the experiments.

163 **2.2. Oxygen response curves in the reactor**

164 **2.2.1. Experimental procedure**

165 Two types of oxygen exposure experiments were executed; one in presence and one in
166 absence of nitrite during oxygen exposure. Ammonium was always present during exposure.
167 For the experiments in the presence of nitrite, the oxygen was applied at the start of an SBR
168 cycle after feeding. Oxygen was introduced with a syringe that added a certain volume of air in
169 the bulk liquid. The oxygen level was adjusted manually to the desired level over a period of 2-
170 hours by adding extra air in the headspace of the reactor. After the oxygen exposure period, the
171 reactor was flushed with N₂-gas, to restore anoxic conditions and the recovery of AnAOB activity
172 was monitored. The experiments in absence of nitrite were executed in a similar way, but N
173 sources were supplied manually. Ammonium (~25 mg NH₄⁺-N L⁻¹) was supplemented before the
174 oxygen spike, and nitrite (~32.5 mg NO₂⁻-N L⁻¹) was added when the DO was below <0.02 mg
175 O₂ L⁻¹ during flushing. For both sets of experiments, different oxygen concentrations (C) and
176 exposure times (t) were applied (See Table 1).

177

178 **2.2.2. Data processing: from ammonium concentrations to removal rates**

179 Datasets of the separate experiments were analyzed using Python (PSF, Oregon, USA).
180 Several functionalities were implemented to obtain a reproducible workflow, also conceptually
181 explained in Supplemental Information F.1 and Figure F.2. In short, for each substrate doses in
182 the SBR cycle, the ammonium profile was monitored. These online ammonium measurements
183 were first filtered. Double ammonium values and measurement errors were removed. In a next
184 step, a calculation-window, over which the slopes were calculated was selected manually. This
185 was done to remove the peak of substrate dose at the start and to remove the lower slopes due
186 to nitrite limitation at the end. Within this calculation-window, a smaller Δt value of 10 minutes
187 was chosen to calculate a higher frequency of slopes. From all the calculated slopes, a
188 response curve was constructed. All code written for the data analysis, along with an example
189 Jupyter Notebook and example datasets, are available online:
190 [https://zenodo.org/account/settings/github/repository/UGentBiomath/2017_HighResolutionRate](https://zenodo.org/account/settings/github/repository/UGentBiomath/2017_HighResolutionRateCalculation)
191 [Calculation](https://zenodo.org/account/settings/github/repository/UGentBiomath/2017_HighResolutionRateCalculation).

193 **2.2.3. Definitions**

194 For interpretation of the response curve, different parameters were defined:

- 195 • The activity 'lag phase' was defined as the time before activity starts after oxygen
196 exposure.
- 197 • The 'initial recovery' after oxygen inhibition was defined as the first rate calculated after
198 oxygen was depleted.
- 199 • The 'recovery time' was calculated as the time between the initial recovery and the first
200 timepoint when full recovery or steady-state was observed, i.e. when the relative %

201 change in average activity in the actual substrate spike compared to the previous
 202 substrate spike was <1%.

203 • 'Irreversible inhibition' was calculated as the difference between steady-state activity and
 204 activity prior to oxygen exposure.

205

206 **2.3. Mathematical modeling**

207 A simple mathematical model was constructed in Aquasim 2.1g (Reichert, 1998) to conceptually
 208 represent the gradual recovery phenomena that were observed in the experiments. The model
 209 only included AnAOB biomass at a constant concentration (X_{AnAOB}) during the simulation runs.
 210 Because the simulations (and experiments) ran only for 1-2 days, no biomass growth was
 211 expected, and was thus not accounted for. Influent ammonium and nitrite concentrations were
 212 50 mg N L⁻¹ and 60 mg N L⁻¹, respectively. The model was composed of a conventional Monod-
 213 type growth process and a decay process (see Tables G.1 and H.1). To represent a delayed
 214 recovery of the AnAOB activity, the growth rate μ_{AnAOB} was modeled as a state variable and was
 215 allowed to change depending on the specific conditions. The differential equation (eq. 1) that
 216 defined the change of μ_{AnAOB} over time was:

$$217 \quad \frac{d\mu_{AnAOB}}{dt} = -k_d \cdot \mu_{AnAOB} \cdot \frac{S_O}{S_O + K_O} + \alpha (\mu_{m,AnAOB} - \mu_{AnAOB}) \cdot \left(\frac{K_O}{K_O + S_O} \right) \quad (\text{eq. 1})$$

218 where k_d is the deactivation constant of AnAOB under anoxic conditions (used to set the
 219 minimum activity) and α is an adaptation parameter of AnAOB to aerobic conditions. The effect
 220 that is introduced by this equation can be linked to the deactivation of enzymes under aerobic
 221 conditions which are critical for AnAOB. k_d is representing the deactivation of these enzymes
 222 similarly to the model by Kornaros et al. (2010), and α the ability to adapt to environmental

223 changes (Kornaros et al. 1998). S_o is the oxygen concentration (imported experimental data)
 224 and K_o the oxygen inhibition concentration for AnAOB from literature.

225 To determine the unknown variables k_d and α , the parameter estimation function in Aquasim
 226 was used employing the secant method (Reichert, 1998), fitting the experimentally determined
 227 relative AnAOB activity to the relative activity calculated by the model. The adapted, 'gradual
 228 recovery' Monod-type model, with dynamic μ_{ANAOb} , was compared to the conventional, instant
 229 recovery Monod-type model with a constant μ_{ANAOb} .

230 Model fits were compared based on χ^2 , the sum of the squares of the weighted deviations
 231 between measurements (Reichert, 1998) and simulated model results (eq. 2). Due to the lack of
 232 activity data during the oxygen exposure, only the datapoints after the DO peak (starting with
 233 the first activity data point) were used to calculate χ^2 .

$$234 \quad \chi^2(p) = \sum_{i=1}^n \left(\frac{y_{data,i} - y_i(p)}{\sigma_{data,i}} \right)^2 \quad (\text{eq. 2})$$

235 with $y_{data,i}$ being the i -th measurement, $\sigma_{data,i}$ the standard deviation, $y_i(p)$ the calculated value of
 236 the model variable corresponding to the i -th measurement and $p = (p_1, \dots, p_m)$ the model
 237 parameters, and n the number of data points.

238 To extrapolate the results to a process level, the gradual-recovery model was used to test
 239 intermittent aeration patterns of 30 min on/off and 180 min on/off at two different DO set-points
 240 of 0.2 and 1.2 mg O₂ L⁻¹ over a period of 5 days (intermittent aeration until day 4). The
 241 parameters α and k_d were both set to a value of 2 for this exercise.

242

243 **2.4. Oxygen response in batch tests**

244 A batch activity test was executed to determine AnAOB activity after extreme oxygen stress
245 conditions: exposure to 8 mg O₂ L⁻¹ for 24h in presence of 40 mg NH₄⁺-N L⁻¹. Flocs were taken
246 from the reactor at day 114 and placed in a pH 7.2 corrected medium of 3.87 g HEPES L⁻¹, 0.2
247 g CaCl₂·2H₂O L⁻¹, 0.1 g MgSO₄·7H₂O L⁻¹, 0.005 g Na₂HPO₄·2H₂O-P L⁻¹, 0.5 g NaHCO₃ L⁻¹, 40
248 mg NH₄Cl-N L⁻¹ and 1 mL L⁻¹ trace elements solutions A and B (Table E.1). The Schott bottles
249 were put on a stirrer and exposed to the air for 24h (=aerobic treatment), reaching 8 mg O₂ L⁻¹
250 within the first hour. The bottles were then flushed with N₂ gas prior to substrate dosing. Control
251 bottles (anoxic control) were not exposed to oxygen, and flushed at the moment when the
252 aerobic exposure time started. 75 mg NH₄Cl-N and 75 mg NaNO₂-N L⁻¹ were spiked at the
253 beginning of the test. Concentrations of NO₂⁻, NO₃⁻ and NH₄⁺ were monitored over time. Volatile
254 and total suspended solids (VSS and TSS) were measured to assess floccular sludge
255 concentration to calculate biomass specific rates as mg N g VSS⁻¹ d⁻¹. Each test was performed
256 in triplicate.

257

258 **2.5. Physicochemical analyses (offline)**

259 Nitrite and nitrate were measured with a 761 Compact IC (Metrohm, CH) and NH₄⁺ with the
260 Nesslerization method (Greenberg et al., 1992). To follow up biomass levels, volatile and total
261 suspended solids (VSS and TSS) were measured with Standard Methods 2540D and E
262 (Greenberg et al., 1992). The particle size distribution of the sludge was measured with an
263 EyeTech (Ambivalue, NL). The high-resolution B&W CCD video camera determined the volume
264 weighed average particle diameter in a ACM-101 magnetic stirring cell with the following
265 settings: sample dilution 1:10, camera lens DW (range 10-600 µm), and setup = sludge DW
266 A101.

267

268 2.6. Molecular biomass analyses

269 Total genomic DNA was extracted from the biomass samples using the Fast DNA Spin kit for
270 soil (MP Biomedicals, US) according to a modified manufacturer's protocol. The quality of the
271 DNA was checked using gel electrophoresis, and the concentration was measured using a
272 QUBIT (Thermo Fisher, US). Using the 16S Ion Metagenomics Kit [™] and Ion Xpress Plus
273 Fragment Library Kit [™] (Thermo Fisher, US), V3-V4 16S rRNA hypervariable region amplicon
274 library was generated, and each sample was tagged using Ion Xpress Barcode Adapters 1–16
275 [™] (Thermo Fisher, US), according to the manufacturer's protocol. Each sample was adjusted to
276 a concentration of 10 picomolar. All samples were pooled in equal volume and processed with
277 One-Touch 2 and One-Touch ES systems (Thermo Fisher, US) according to the manufacturer's
278 instructions. Sequencing was performed on a Ion Torrent (ION Torrent Ion S5) using the 200-bp
279 kit and the 520 chip. Base calling and run demultiplexing were conducted by Torrent Suite
280 version 4.4.2 (Thermo Fisher, US) with default parameters. The Ion reporter software was used
281 for, (1) separating sequences based on their respective targeted regions; (2) then picking OTU
282 (operational taxonomical unit) with its default settings. Overall, the de novo clustering of OTUs
283 was done with 97% identity, corresponding to species level. The sequences were then classified
284 based on the taxonomy in the Greengenes database (97% confidence threshold, version 13.5
285 May 2013) (McDonald et al., 2012). The sequencing data was analyzed in R, using ggplot2
286 (v0.9.3.1), to determine the community composition of the biomass in the reactor.

287

288 **3. Results**

289 **3.1. Microbial community enriched in *Ca. Brocadia***

290 Over the experimental period, the reactor community was enriched in one AnAOB OTU of the
291 genus *Ca. Brocadia* from 64% (day 38) to 75% (day 106), based on 16S rDNA amplicon
292 sequencing (See Figure I.1 for more detailed community analysis). As this AnAOB is commonly
293 encountered in engineered systems, it is anticipated that its response behavior to oxygen is
294 applicable to PN/A applications (Oshiki et al., 2015).

295

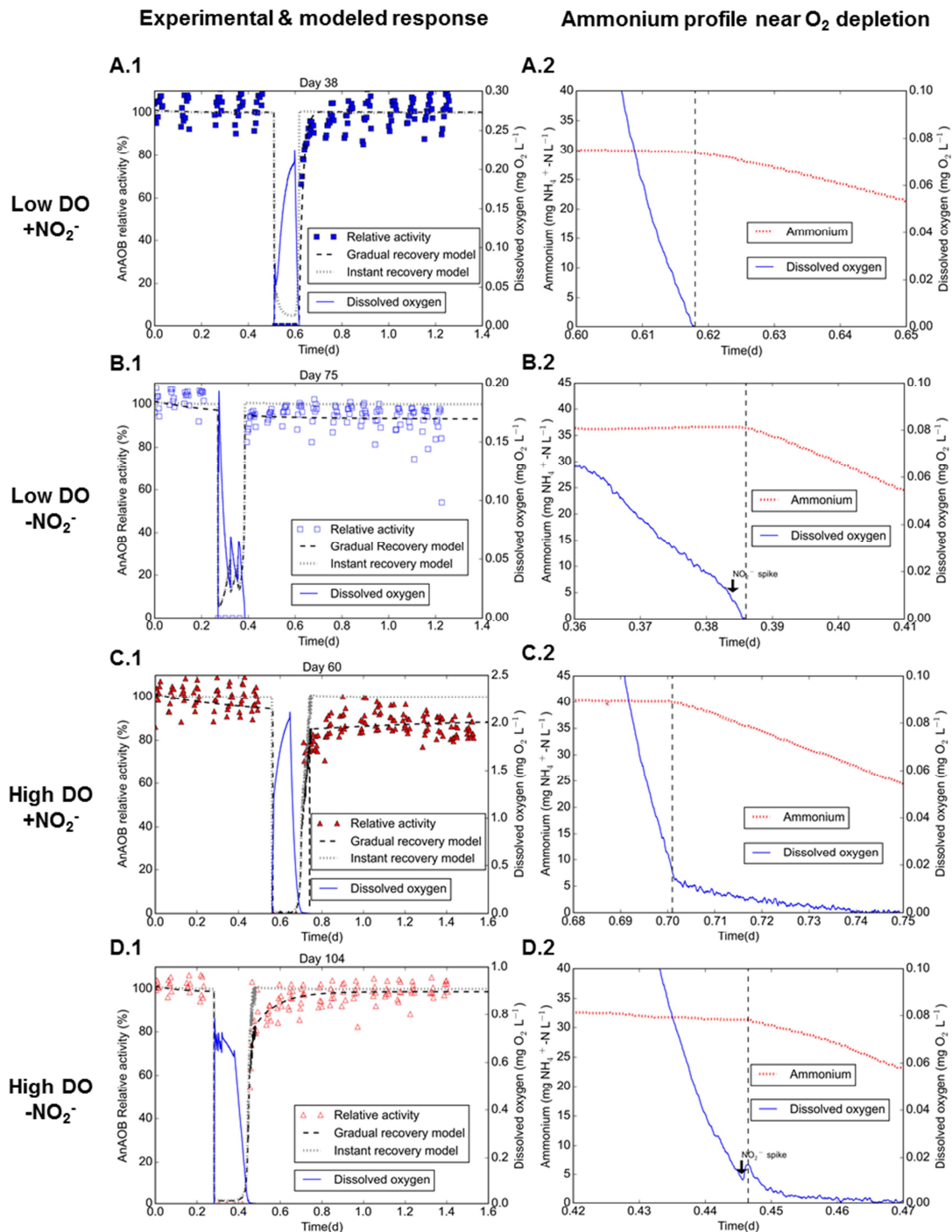
296 **3.2. Experimental oxygen response curves**

297 AnAOB responded differently towards tested combinations of oxygen concentrations, exposure
298 times and presence or absence of nitrite. An overview of the experiments is given in Table 1,
299 while Figure 1 depicts typical activity curves including modeling efforts. The AnAOB response
300 curve consisted of the following parts which will be described below: AnAOB activity prior and
301 during exposure, an eventual activity lag phase, initial recovery after exposure, and gradual
302 recovery with or without irreversible inhibition.

303

304 **3.2.1. AnAOB activity in presence of O₂**

305 During oxygen exposure, no ammonium was consumed, and AnAOB activity was only regained
306 when oxygen concentrations were below micro-aerobic conditions <0.02 mg O₂ L⁻¹ (Figure 1,
307 Panel C.2 and D.2), indicating a direct and high intolerance to oxygen.



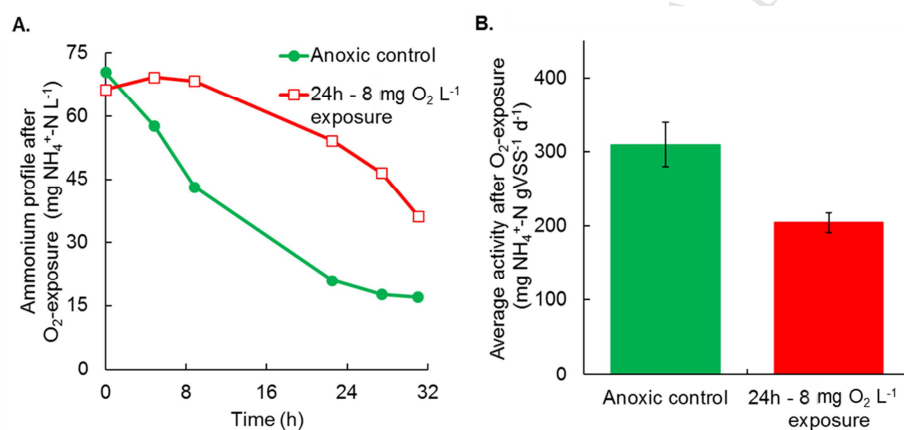
308

309 **Figure 1.** Experimental and modelled response of AnAOB to oxygen exposure in the reactor. Conditions
 310 during exposure: low DO (blue squares), high DO (red triangle), subsequently with nitrite (filled) or without
 311 nitrite (empty). Ammonium was always present. The left-side graphs show the experimental data
 312 compared to an instant recovery Monod model and the newly developed gradual recovery Monod model.
 313 The right-side graphs indicate AnAOB activity at micro-aerobic levels. The vertical dashed line represents
 314 the moment when ammonium starts to deplete.

3.2.2. AnAOB activity lag phase after oxygen inhibition

After oxygen disappearance, AnAOB reversibly and immediately reinitiated activity to 55-80% of the pre-exposure activity, depending on the conditions during exposure. This drop of initial recovery after oxygen inhibition was not due to substrate (ammonium and nitrite) starvation during exposure, because longer anoxic periods without substrate did not result in activity loss. In an *ex-situ* batch test, the effect of longer oxygen exposure (24h) at a higher oxygen concentration (8 mg O₂ L⁻¹) was tested (Figure 2). Activity could be detected only after a 5h lag phase and it increased gradually from 10% to 91% of the initial activity after 26h under anoxic conditions.

324



325

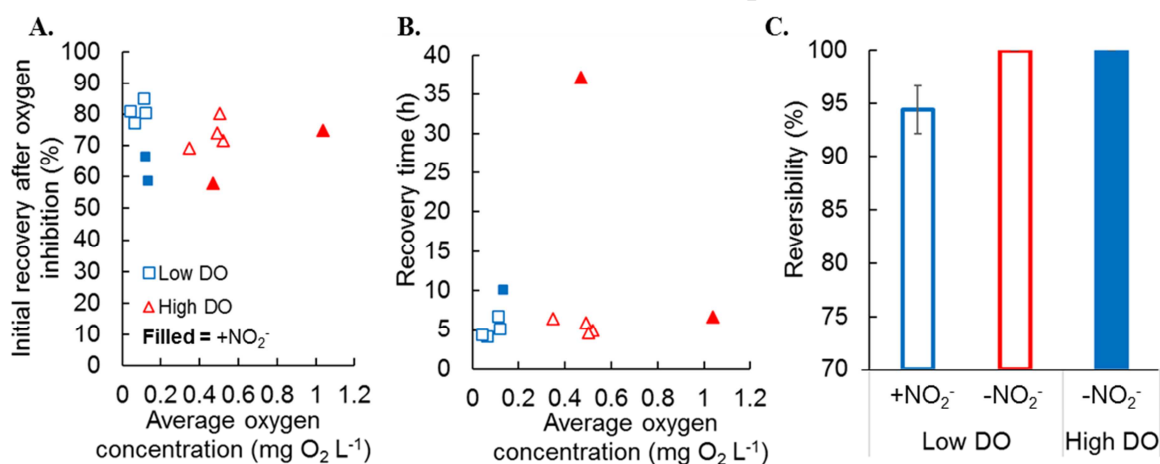
Figure 2. AnAOB recovery after exposure to 8 mg O₂ L⁻¹ for 24h with 40 mg NH₄-N L⁻¹, in the absence of nitrite. Ammonium (35 mg N L⁻¹) and nitrite (75 mg N L⁻¹) were dosed after oxygen exposure. Panel A depicts a typical measured ammonium profile after exposure, compared to an anoxically kept control, while Panel B shows the resulting sludge-specific activity after exposure (average ± standard deviation).

329

330 3.2.3. Initial recovery after oxygen inhibition

331 For the *in-situ* reactor tests, initial recovery after oxygen inhibition decreased from $81\pm 3\%$ to
 332 $74\pm 5\%$ when AnAOB were exposed to higher compared to lower oxygen concentrations in
 333 absence of nitrite, on avg. 0.47 ± 0.08 vs. 0.08 ± 0.04 mg O₂ L⁻¹ (Figure 3). This increased drop in
 334 initial recovery could also be due to longer exposure times for the high DO-setpoint (Figure J.1),
 335 which were on avg. 3.7 ± 0.3 h vs. 2.8 ± 0.2 h for the low DO-setpoint. A lower initial recovery after
 336 oxygen inhibition was also recorded when AnAOB were exposed to 0.5 ± 0.2 mg O₂ L⁻¹ for 8.7h in
 337 the presence of nitrite, with a gradual recovery from 58% to full recovery after 37h (Day 18).

338



339

340 **Figure 3.** Depicting the calculated kinetic parameters from the experimental reactor data. A.) Measured
 341 initial recovery as a function of the oxygen exposure concentration, B.) time to reach steady state activity
 342 after exposure (= recovery time) and C.) reversibility of activity after exposure. Plotted data can be found
 343 in Table K.1.

344

345 3.2.4. Gradual recovery

346 After exposure, AnAOB gradually recovered towards a steady-state activity. This recovery time
 347 until steady state was similar for the low and high DO-setpoint, and in the range of 4.2-6.8h, with

348 on avg. 6.3 ± 0.9 h. The recovery time also did not show any correlation with the concentration
349 (C) (Figure J.1), exposure time (t) (Figure J.1), integrated Cxt (Figure J.1) or initial recovery after
350 oxygen inhibition. AnAOB did not always fully recover (see Figure 2, Panel C and Figure 1,
351 Panel B and C), and consistent irreversible loss of activity of about 5% was seen when the
352 biomass was exposed to low oxygen setpoints of on avg. $0.06\text{-}0.12$ mg O₂ L⁻¹ in absence of
353 nitrite. A short 4h exposure with a high DO setpoint of on avg. 1 mg O₂ L⁻¹ and in the presence
354 of nitrite also led to more severe irreversible loss of activity of about 9% (Figure 1, Panel C).
355 Conversely, the short-term (~3h) and long term (~8h) exposures at DO setpoints in a range of
356 on avg. $0.12\text{-}0.5$ mg O₂ L⁻¹ with nitrite, as well as the higher DO setpoints of on avg. $0.35\text{-}0.52$
357 mg O₂ L⁻¹ in absence of nitrite, did not reveal any irreversible inhibition. This indicates that the
358 complex interplay of oxidative and nitrosative (O₂ + N-species) conditions, combined with
359 substrate presence might regulate how AnAOB cope with oxidative and nitrosative stress.

360

361 **3.3. Modeling gradual recovery after oxygen inhibition**

362 A gradual-recovery model was constructed to describe the gradual recovery after oxygen
363 inhibition. A comparison with the classical instant recovery model (Figure 1 and Table 1)
364 revealed a considerably improved fit of the gradual-recovery model to the data than the instant
365 recovery model, evident from the 2 – 4-fold lower χ^2 . The newly developed gradual-recovery
366 model included a dynamic growth rate to simulate the delay in the recovery of the AnAOB
367 activity (Figure 1).

368

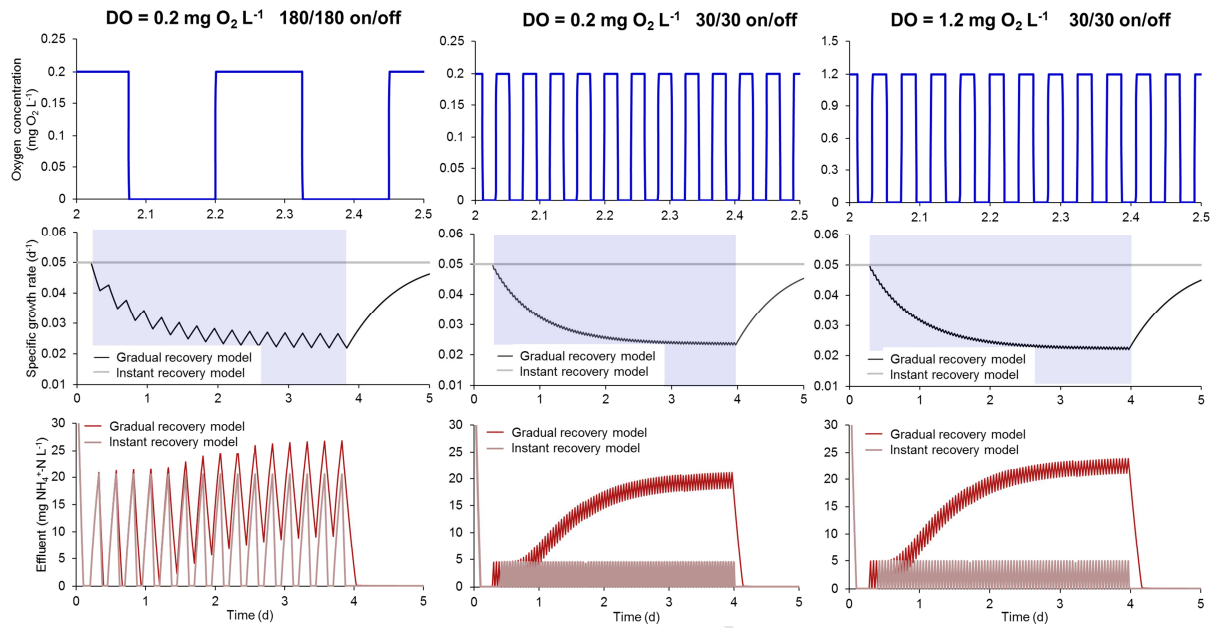
369 The automated fitting produced sometimes large variations and/or standard errors for the
370 parameters α and k_d . One reason is the high fluctuation of the measured activities during one

371 SBR cycle of up to $\pm 10\%$. Another reason is the lack of data for the AnAOB activity during
372 oxygen exposure. To more accurately determine both parameters, more data during initial
373 oxygen exposure and shortly after oxygen depletion is necessary. Especially k_d is highly
374 influenced by the slope of loss and regain of the activity, and the data does not allow for a highly
375 accurate parameter estimation (also visible from the standard deviations of k_d). Since k_d
376 represents the deactivation of the AnAOB activity due to the oxygen, higher k_d values are
377 required for bringing down the activity to zero during oxygen exposure. The parameter α mainly
378 formed the shape of the activity curve during recovery and with higher α values the recovery
379 became faster (Table 1).

380

381 Process modeling was further used to estimate the impact of the newly observed oxygen
382 response on a PN/A. This implied evaluating the impact of longer phases of intermittent aeration
383 on the effluent composition of a continuously operated PN/A reactor, with simulations run over 5
384 days employing different aeration patterns (see M&M). Figure 4 summarizes the results of these
385 simulations, which were evaluated based on the nitrogen species in the effluent exemplarily
386 plotting the ammonium concentration. It was evident that the ammonium concentrations in the
387 effluent as predicted by the gradual-recovery model were higher than the ones predicted by the
388 instant recovery model. From the different conditions tested, a less frequent oxygen alteration
389 (180/180 on/off) rather than a higher oxygen level (1.2 vs 0.2 mg O₂ L⁻¹) yielded high peak
390 ammonium concentrations earlier on. These effects were caused by the decrease in the growth
391 rate in the gradual-recovery model (Figure 4, left row) due to the imposed oxygen peaks. **This**
392 **decrease in growth rate is influenced by α and k_d (which were set to 2 in this example),**
393 **leading to more severe effects for lower α values and ratios of α / k_d of < 1 .**

394



395

396 **Figure 4.** Results of the simulation runs over 5 days employing different intermittent aeration patterns.
 397 First row: section of the dissolved oxygen (DO) pattern; second row: growth rate (μ) for the instant
 398 recovery Monod model (constant, instant recovery) and the gradual recovery Monod model (dynamic,
 399 gradual recovery) and; third row: ammonium effluent concentrations of both models.

400 **4. Discussion**

401 **4.1. AnAOB inhibition by oxygen**

402 **4.1.1. AnAOB gradual recovery after oxygen inhibition**

403 To our knowledge this is the first report showing at high temporal resolution how AnAOB activity
404 is affected by oxygen exposure. AnAOB were completely inhibited by oxygen, and after short
405 exposure resumed activity directly, with a gradual recovery until reversible activity was
406 overcome. Other anoxic/anaerobic bioconversions, for instance by sulfate reducing bacteria or
407 denitrifying bacteria, which similarly to AnAOB rely on substrates derived from oxidative and
408 reducing conditions at oxic and anoxic/anaerobic interfaces, responded kinetically similarly after
409 oxygen exposure (Dolla et al., 2006; Fareleira et al., 2003; Kornaros & Lyberatos, 1998).
410 AnAOB and other microbes living under recurrent oxygen exposure thus adapted towards these
411 conditions, with production of anti-oxidative enzymes and recovery of activity after exposure,
412 which cannot be explained by growth (Dalsgaard et al., 2014; Yan et al., 2012).

413

414 The measured kinetic response is most likely overlooked so far because most studies
415 investigated activity only during exposure. In the scarce cases studying also recovery, the
416 outcome might have been biased due to lower data resolution after exposure, *i.e.* 4-6 datapoints
417 over a course of 2-6 h (Dalsgaard et al., 2014.; Strous et al., 1997) or aggregate size (Lotti et al.,
418 2012; Strous et al., 1997). For example, Strous et al. (1997) reported no influence on biomass
419 specific activity rates when a granular biomass (*Brocadia*) was repeatedly exposed for 20 days
420 to 2h aerobic and anoxic conditions. Similarly, Lotti et al. (2012) reported no loss in activity after
421 an extreme oxygen exposure of 24h and 5 mg O₂ L⁻¹ for granular biomass ($\phi=1.1$ mm). In both

422 cases, diffusional limitations could have underestimated initial biomass specific activities, and
423 consequentially inhibition could be masked by this underestimation.

424

425 **4.1.2. Factors impacting initial recovery after oxygen inhibition**

426 With the current dataset, it was not feasible to deduct whether the oxygen concentration (Figure
427 2) or exposure time (Figure J.1) had a larger influence on the initial recovery, as higher oxygen
428 concentrations also had longer exposure times. It was hypothesized that 'concentration times
429 exposure time' (Integrated Cxt, Figure J.1) steered the AnAOB response, but Spearman rank
430 correlation analysis did not yield any correlation. From the gradual-recovery model, the
431 exposure time had a higher influence than the concentration, while other factors like adaptation
432 and nitrite presence during exposure might differentiate the response and thus fitted kinetic
433 parameters. Both aspects require further experimental research to elucidate the exact factors
434 determining initial recovery after oxygen inhibition.

435

436 **4.1.3. Novel experimental procedure and impact on measured kinetics**

437 In this study, AnAOB responded very sensitive towards oxygen, with an observed IC_{100} 0.02 mg
438 $O_2 L^{-1}$ or below. This showed that diffusional limitations in rather large flocs of ~200 μm or
439 oxygen consumption by putative heterotrophs in the AnAOB satellite community did not shield
440 AnAOB from oxygen. AnAOB sensitivity towards oxygen was comparable with some freshwater
441 AnAOB, *Brocadia sp.* and *Kuenenia stuttgartiensis*, which were for 100% inhibited at micro-
442 aerobic conditions of <0.04 and <0.12 mg $O_2 L^{-1}$ respectively (Egli et al., 2001; Strous et al.,
443 1997). In contrast, two *Brocadia*-enriched floccular sludge types were very tolerant towards
444 oxygen exposure, with IC_{50} values of 2 and 3.8 mg $O_2 L^{-1}$ respectively, which cannot be

445 explained when compared to our or other studies (See Figure D.1) (Carvajal-Arroyo et al., 2003;
446 Oshiki et al., 2011). Also, for marine AnAOB, higher IC_{100} values between 0.09 and 0.64 mg O₂
447 L⁻¹ were reported (Babbin et al., 2014; Dalsgaard et al., 2014; Jensen et al., 2008; Kalvelage et
448 al., 2011). For the latter, the difference between measured sensitivities was hypothesized to
449 come from either different testing conditions, *i.e.* absence of mixing or presence in aggregates,
450 shielding AnAOB (partially) from oxygen, or from more adapted AnAOB due to presence in
451 oxygen minimum zones (Dalsgaard et al., 2014). As in our testing conditions AnAOB were
452 repeatedly exposed to oxygen with low diffusional limitations, the obtained data probably
453 resembled intrinsic and more adapted kinetics. The newly developed methodology, which
454 obtained high-resolution data processed by a data-pipeline, is therefore highly suitable for
455 studying kinetic responses of stressors and inhibitors on AnAOB.

456

457 **4.2. Modeling AnAOB inhibition by oxygen**

458 In mathematical modeling, oxygen inhibition of AnAOB has typically been modelled by an
459 instant-recovery Monod model with an inhibition constant $K_{O,AnAOB}$ of 0.01 mg L⁻¹ (based on
460 Strous et al., 1998). With this kinetic constant, 10% AnAOB activity is expected at 0.1 mg O₂ L⁻¹,
461 which overestimated the expected activity of AnAOB during oxygen exposure in our
462 experiments (See Figure 1). Further fine-tuning of this kinetic constant is thus necessary. After
463 oxygen exposure, AnAOB directly resumed activity only in a range of 30-80% of the initial
464 activity. This direct recovery of activity was conventionally reported and perceived as reversible
465 activity of AnAOB without any activity loss (Dalsgaard et al., 2014; Egli et al., 2001; Lotti et al.,
466 2012; Strous et al., 1997). Instant-recovery models therefore do not account for a delay in
467 recovery to less than 100%, while the newly developed gradual-recovery model describes this
468 more accurately. Gradual recovery has been previously modelled for several other nitrogen bio-

469 conversions. Alex et al. (2009) and Casarus et al. (2005) introduced an enzyme as additional
470 state variable, and processes for its activation and deactivation to simulate bacterial lag phases
471 for nitrification (Alex et al., 2009) and denitrification (Casarus et al., 2005). Identically to our
472 approach, the NOB lag phase after anoxic phases (Kornaros et al. 2010) was successfully
473 modeled with the growth rate as dynamic variable thus influencing the maximum specific growth
474 rate. The latter enabled to exploit these lag phenomena to optimize NOB suppression in their
475 reactors. These examples show that incorporating additional equations may lead to more
476 accurate models and further insights on how to optimally operate mainstream PN/A. Additional
477 research is needed, however, due to the limited amount of data, especially during oxygen
478 exposure, to accurately estimate the newly introduced parameters, α and k_d , independently.
479 Furthermore, other effects like exposure time, oxidative stress conditions and adaptation could
480 also impact the modeled parameters, and should be taken into account in further modeling
481 exercises.

482

483 **4.3. Long-term recurrent exposure: exploratory simulations**

484 From a modelling perspective, the exposure time is the critical parameter when looking at the
485 implications on effluent quality (Figure. 4). The simplified model showed that a delay in the
486 recovery of full activity will potentially lead to a decrease of the overall turnover capacity of the
487 system, which is reflected in the effluent concentration (exemplified by ammonium). An instant-
488 recovery model, where the activity is back to 100% right after oxygen depletion, does not
489 capture such a delay and will therefore not account for any impact this had on the reactor
490 performance. The model suggested that longer anoxic periods in combination with short intense
491 aeration intervals might be the strategy of choice. However, the model in its present form does
492 not account for other effects, that might influence recovery, such as adaptation (Yan et al.,

493 2012), niche differentiation (Oshiki et al., 2015), aggregate morphology, or nitrite availability,
494 which also seemed relevant from the experimental observations.

495

496 In practice, also aerobic processes and mass transfer limitations partially shield AnAOB in
497 biofilms from oxygen exposure. For this reason, increased AnAOB resilience was reported in
498 biofilm systems compared to flocs, e.g. lower nitrite build-up was detected when nitrification
499 (and thus oxygen shielding) failed due to sudden or long-term disturbances in temperature
500 drops (Gilbert et al., 2016.; Wells et al., 2017). However, completely anoxic conditions might be
501 hard to achieve, and will also depend on the height of the oxygen peaks (See Figure in B.1).
502 AnAOB recovery from oxygen inhibition will thus always be present in PN/A reactors, and
503 studies on long-term repeated exposure are warranted to achieve the process intensification,
504 i.e. maximize nitrogen removal rates and efficiencies.

505

506 **5. Conclusions**

507 AnAOB reside mostly in proximity of oxygen in wastewater treatment processes such as PN/A,
508 possibly experiencing repeated inhibition. Nonetheless, it was not well understood how AnAOB
509 activity is affected by oxygen stress. A newly developed methodology, which combined high
510 temporal resolution measurements with a fit-for-use data-pipeline enabled to study this, and
511 corroborate the findings with a newly developed mathematical model.

- 512 1. AnAOB showed no activity during exposure ($<0.02 \text{ mg O}_2 \text{ L}^{-1}$). Due to oxygen
513 inhibition, AnAOB lost activity compared to pre-exposure rates. This initial loss in
514 activity was followed by a gradual (partial) recovery of activity, dependent on the
515 oxidative (or nitrosative) stress conditions perceived during oxygen exposure.
- 516 2. The initial recovery after oxygen inhibition was slower when exposed towards higher
517 dissolved oxygen concentrations and contact times.
- 518 3. Exposure to low oxygen concentrations of $0.05\text{-}0.2 \text{ mg O}_2 \text{ L}^{-1}$ did not result in full
519 recovery ($\sim 95\%$).
- 520 4. A new gradual-recovery Monod model described the gradual recovery more
521 accurately than the current instant-recovery model.
- 522 5. Long-term simulated exposure with intermittent aeration and the gradual-recovery
523 model led to reduced AnAOB growth rates compared to an instant recovery model.
524 Considering oxygen inhibition on AnAOB can thus be an important factor to optimize
525 aeration patterns in single-stage PN/A reactors, allowing to maximize nitrogen
526 removal rates and efficiencies.

527

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533

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670
671

672

1 **Table 1.** Overview of reactor experiments executed with low dissolved oxygen (DO) (blue) and high DO (red). Table shows DO exposure
 2 concentrations (C), times (t) and integrated exposure (Cxt), along with the parameters obtained from fitting the gradual-recovery model. For both
 3 models, i.e. conventional Monod with instant recovery and adapted Monod with gradual recovery, the estimated χ^2 values are given. The lower,
 4 the better a fitted model represents the experimental data. Response curves of the days highlighted in bold are shown in detail in Figure 1. n° =
 5 replicate number.

	Day	n°	DO concentration C		Exposure time t (h)	Integrated exposure Cxt (mg O ₂ L ⁻¹ h)	Gradual-recovery model kinetic parameters		Estimated fit (χ^2)	
			Average (mg O ₂ L ⁻¹)	Peak Height (mg O ₂ L ⁻¹)			α	k_d	Instant recovery model	Gradual recovery model
Low DO	■ + NO ₂ ⁻	38 (1/2)	0.12±0.07	0.23	2.64	0.014	63.7±2.74	2937±4217	11378	5832
		40 (2/2)	0.13±0.06	0.23	2.56	0.014	71.91±5.49	15686±1323	16712	7907
	□ - NO ₂ ⁻	67 (1/4)	0.06±0.06	0.27	2.88	0.007	2.56±1.02	1.34±0.56	10295	1292
		75 (2/4)	0.06±0.04	0.19	2.64	0.007	1.95±0.72	0.15±0.33	10025	5345
		81 (3/4)	0.12±0.09	0.38	2.64	0.012	4.81±0.92	1.19±0.37	6721	2809
82 (4/4)	0.12±0.08	0.35	3.12	0.015	4.56±0.71	0.98±0.19	5528	1853		
High DO	▲ + NO ₂ ⁻	18 (1/1)	0.47±0.21	0.75	8.71	0.177	9.20±0.43	10131±6771	48482	7281
		60 (1/1)	1.04±0.8	2.11	4.08	0.071	1.30±0.21	0.58±0.13	31000	10080
	△ - NO ₂ ⁻	85 (1/4)	0.52±0.26	0.75	3.29	0.056	60.5±52.4	6.03±458	9501	7438
		98 (2/4)	0.35±0.24	0.81	3.89	0.088	55.24±7.5	21.95±10.8	10315	4040
		104 (3/4)	0.49±0.24	0.78	4.08	0.088	8.79±1.23	1.42±0.21	723	2208
106 (4/4)	0.50±0.30	0.84	3.84	0.014	6.94±3.38	0.16±0.32	17054	16274		

Highlights

1. During oxygen exposure, anammox bacteria (AnAOB) showed no activity
2. Interestingly, the AnAOB activity recovered only gradually after oxygen exposure
3. The proposed model fitted the data better than the conventionally used model
4. The recovery impacts achievable removal rates for partial nitritation/anammox