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Nitrogen cycling in bioregenerative life support systems : challenges for waste refinery and food production processes

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1	Nitrogen cycling in Bioregenerative Life Support
2	Systems: challenges for waste refinery and food
3	production processes
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20 Abstract

21 In order to sustain human life in an isolated environment, an efficient conversion of wasted 22 nutrients to food might become mandatory. This is particularly the case for space missions where 23 resupply from earth or in-situ resource utilization is not possible or desirable. A combination of 24 different technologies is needed to allow full recycling of e.g. nitrogenous compounds in space. 25 In this review, an overview is given of the different essential processes and technologies that 26 enable closure of the nitrogen cycle in Bioregenerative Life Support Systems (BLSS). Firstly, a 27 set of biological and physicochemical refinery stages ensures efficient conversion of waste 28 products into the building blocks, followed by the production of food with a range of biological 29 methods. For each technology, bottlenecks are identified. Furthermore, challenges and outlooks 30 are presented at the integrated system level. Space adaptation and integration deserve key 31 attention to enable the recovery of nitrogen for the production of nutritional food in space, but 32 also in closed loop systems on earth.

33

34 Keywords

35 resource recovery, space, single cell protein, food production, organic waste, urine, CELSS

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40 **1. Introduction**

41 At present, human life in space flights and in the International Space Station (ISS) is guaranteed 42 by a regular resupply of food and water. However, in order to explore deep space with long-term 43 missions and space habitation with increasing crew size, resupply from and return of waste to earth becomes difficult because of the long transport time and high costs associated with mass 44 45 and volume restrictions for transportation [1-3]. The mass requirements of 5500-12800 kg per 46 crew member per year for Open Life Support Systems (OLSS) without recycling can be lowered 47 to 340-470 kg per crew member per year in a Physicochemical Life Support System (PLSS) (e.g. 48 ISS) by in situ generation of oxygen and water recycling, consuming ~300 W per crew member 49 [1]. The current launch cost advertised by SpaceX service is about \$12600 per kilo [1]. In order 50 to further decrease this payload mass and costs for resupply for long-term exploration or 51 permanent habitation in life support systems, in situ food production has been proposed. Such 52 systems are called Bioregenerative Life Support Systems (BLSS) or Closed/Controlled 53 Ecological Life Support Systems (CLSS) [4].

54 Certain physicochemical methods for recovery of water and air have been developed for use in 55 PLSS and are currently in use at the ISS, but technologies for food production based on nutrient 56 recovery are neither validated nor available for space deployment. On earth, we rely on a vast set 57 of biological production systems to produce food, mainly based on plants and animals, which are 58 in essence based on inorganic nutrients often supplied as fertilizers [5]. In BLSS research, the 59 development of an engineered bio-based system for food production has been investigated by major governmental space research agencies for the past half century [6-9]. The shared focus of these BLSSs has been the integration of different biological and physicochemical technologies for the breakdown and conversion of waste products into useful building blocks for plant food production in a closed material recycle. Nitrogen is a critical nutrient for this cycle, and will be the focus of this review. Processes for nitrogen retention, recovery, and resupply in a closed system will be considered here.

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67 **2. Refinery stages: converting waste into a fertilizer**

There is currently no reuse of nitrogen on the International Space Station (ISS). Fecal material is collected, stored and returned to earth without nutrient recovery, while fresh urine is treated by chromium trioxide and sulfuric acid dosing to avoid microbial growth, which would convert urea into volatile and potentially harmful ammonia. Water is recovered from pretreated urine by vapour compression distillation (VCD) through which nitrogen ends up in a brine which becomes a nitrogen dead end as well [10].

74 Human activity and crop production in a BLSS result in the production of different types of organic wastes, all containing nitrogen. A dietary protein intake of 0.8-1.5 g protein kg⁻¹ body 75 76 weight for a crew member with a body weight between 65 and 85 kg is expected to result in a urinary excretion of between 7 and 16 g N d⁻¹ (assuming ~16% N in proteins and ~80% N-77 excretion via urine [7, 11, 12]). Fecal nitrogen excretion is typically in the order of 1-2 g N d^{-1} 78 [13] (Figure 1). Based on the assumptions of Hu *et al.* [8], 5-6 g N d⁻¹ per crew member would be 79 collected as inedible biomass (crop residues, kitchen waste) in the proposed BLSS and ~ 1 g N d⁻¹ 80 81 would be collected as epithelial associated organic waste (hair, nails, saliva solids, dead skins cells, ...). In order to make this nitrogen available again for the production of food in a BLSS, these waste streams need to be treated to produce fertilizers adapted to the specific needs of the food production processes. Different technologies have been proposed over the past decades for BLSS (Figure 2) for conversion of organic wastes into carbon dioxide, water and nutrients. For recovery of this nitrogen in a bioavailable form, three main strategies can be distinguished: biological or physicochemical ammonification, and nitrification.

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2.1 Biological ammonification

90 Most of the nitrogen in the organic waste streams in BLSSs is bound in organic compounds. 91 Although using organic nitrogen (amino acids, urea) for plant production may have 92 biostimulatory effects [14-17], providing inorganic nitrogen (ammonia and nitrate) is often 93 preferred as it allows online monitoring and control of the nitrogen loading and uptake by crops 94 in hydroponic systems and by bacteria in bioreactors for microbial protein production. The first 95 step in converting this organic nitrogen to the desired form for food production is biological 96 ammonification of the organic waste (Figure 2): proteins and peptides are converted into amino 97 acids by proteases produced by living organisms, while amino acids and other amide containing 98 molecules can be hydrolyzed by amidases to form ammonia. Urea $(CO(NH_2)_2)$, which contains 99 more than 90% of the nitrogen in fresh urine [18], can be ammonified by the widespread enzyme 100 urease or by urea amidolyase [19].

In several concepts of the BLSS, microbial hydrolysis of organic waste occurs in a dedicated aerobic [8, 20-23] or anaerobic [24-26] bioreactor. Besides hydrolysis of organic compounds, biological ammonification and the release of other nutrients from the organic matrix is established with the help of microorganisms. In the current concept of the BLSS of the European Space Agency (ESA), the 'Micro-Ecological Life Support System Alternative' (MELiSSA), organic waste is fermented in a thermophilic anaerobic membrane bioreactor at pH 5.3 to inhibit methanogenesis and to maximize the formation of volatile fatty acids (VFA). Typically, between 18 and 71% of the organic nitrogen in the waste (plant residues and fecal material) could be converted into ammonium at a rate between 17 and 30 mg NH₄⁺-N L⁻¹ d⁻¹ in this waste treatment compartment [27-30].

The main challenges in aerobic ammonification systems are the difficulty of diffusion driven aeration under microgravity conditions and the high microbial sludge yield growth under aerobic growth. The lower sludge yield under anaerobic conditions is advantageous on one hand, but the separation and downstream processing of the gases that are produced (methane, hydrogen) is challenging on the other hand. Additionally, if nitrate rich organic crop residues are treated in anaerobic conditions, nitrate could be reduced to ammonia via dissimilatory nitrate reduction but denitrification, resulting in gaseous nitrogen losses, is likely to occur.

Bioreactors with immobilized urease enzymes have been proposed to convert the urea in human urine into ammonia and carbon dioxide for space applications [31]. Nicolau *et al.* [32] combined this concept with electrochemical oxidation of the ammonia for electrical power production.

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122 **2.2 Physicochemical ammonification**

Hydrolysis of nitrogen-containing organic compounds can also be achieved by means ofphysicochemical processes. Hydrolysis of proteins occurs under acidic and alkaline conditions,

but also the use of microwave radiation, high pressure and enzymatic treatment as well ascombinations of such treatments have been described [33, 34].

127 Hot water conversion is one commonly tried method to physicochemically hydrolyze nitrogenous 128 compounds. Hydrolysis in hot water is only effective at temperatures above 140-160°C (and at 129 corresponding pressures to keep water in a liquid state), as below these temperatures only 130 denaturation and insolubilisation of proteins occur [35-37]. Hydrolysis of proteins into amino 131 acids has the highest yield in the temperature range of 200 to 290°C [38-40]. When water is heated to just below its critical point ($T_c = 374$ °C and $p_c = 22.1$ MPa [41, 42]), its ionic product 132 will rise from 10⁻¹⁴ at ambient temperature, to 10⁻¹¹ at near-critical conditions [43]. As a 133 134 consequence, the higher concentration of protons and hydroxyl ions from the dissociation of 135 water will lead to a higher extent of acid and base catalyzed reactions with nitrogenous organic 136 compounds [44]. Further increasing the temperature in the sub-critical region (i.e. between 250 137 and 374°C) will enhance deamination reactions of the amino acids as intermediates, yielding free NH_4^+ and carboxylic acids [45, 46], although competing polymerisation reactions such as amide 138 139 formation and Maillard type reactions (i.e. reactions with sugars) [47, 48] could occur.

With respect to BLSS, Lissens *et al.* [27] reported that 95-100% of all nitrogen present in the biosolids from an anaerobic digester could be converted into water-soluble components through hydrothermal degradation (~350°C and ~240 bar). The anaerobic digester was fed with a mixed organic stream (food crops, fecal material, algae) resembling a concentrated organic waste stream produced by humans. About 60% of the nitrogen in the effluent of the hydrothermal unit could be identified as NH_4^+ -N and NO_3^- -N, while the remaining fraction was assumed to be other oxidized and solubilized nitrogen species. 147 The decomposition of the organic compounds contained in the waste streams may be enhanced in 148 hydrothermal conversion processes by the addition of oxidizing agents such as air, pure O_2 or 149 hydrogen peroxide. Below the critical point of water with air as an oxidizing agent, the process is 150 termed 'wet air oxidation' and can result in 90% conversion of organic nitrogen into mainly 151 ammonia and N₂ (at ~275°C and ~103 bar) [49, 50]. Johnson and Wydeven [51] reported an 152 increase in the formation of N_2 gas with increasing temperature (224°C to 300°C). Catalysts may 153 be added to increase selectivity towards N_2 -gas formation. Thu and Michele [49] used Ru/TiO₂ to 154 achieve a selectivity to N₂ and nitrate of 85% and 15%, respectively after 22 h, while in the 155 presence of Pt/TiO₂ the selectivity to N₂ was 89% after 8 h from model distillery wastewater 156 feed. For BLSS, the Kudenko process has been developed which is an electrochemical wet 157 oxidation process, using hydrogen peroxide as oxidizer, to treat organic waste (inedible crop 158 residue, urine, feces and grey water) at 90°C and ambient pressure [23, 52]. When treating a 159 mixture of crop residues and urine, about 53% of the nitrogen could be recovered in the effluent 160 as soluble nitrogen, mainly under the form of ammonium but also nitrate and nitrite were 161 generated [52].

162 Finally, the oxidation of wet organic residues can also be carried out in supercritical water, the 163 so-called supercritical water oxidation (SCWO) process. Therein, nitrogen-containing organic 164 compounds are converted to CO₂, ammonia, nitrate, nitrite, N₂ and N₂O [53, 54]. In SCWO, the 165 oxidation of the ammonia as an intermediate is a rate-limiting step, hence ammonia is considered 166 recalcitrant. Temperatures above 500°C are needed to initiate the oxidation of NH_4^+ [55] in the 167 presence of excess oxidizer [56, 57]. Up to temperatures of 650°C, ammonia is still being formed 168 in SCWO effluents despite its oxidation. Compared to (non)catalytic wet air oxidation, SCWO 169 reactions are typically completed in less than 1 min.

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171 **2.3 Nitrification**

172 Although in some cases urea and ammonia can be taken up directly by plants and 173 microorganisms grown for food production, it can be preferable to convert urea and ammonia, at 174 least partially, to nitrate in a BLSS [25]. In confined spaces, the occurrence of liquid streams with 175 high ammonia concentrations is considered a hazard as ammonia volatilization increases with an 176 increasing pH and temperature, and can accumulate to toxic levels in the atmospheric 177 compartment. Moreover, high ammonia concentrations resulting from failing or inadequate 178 dosage can become toxic as well to the plants and microorganisms, even when the pH is 179 controlled [26, 58] and high levels of ammonium can inhibit the uptake of key minerals in 180 hydroponic solutions [25]. Nitrate, on the other hand, is not volatile and is not considered to be 181 toxic in the concentrations that are expected and can accumulate in the leaves in case of a high 182 nitrate supply [25], although this oxidized form of nitrogen is energetically less favorable for 183 edible protein production.

The biological process in which ammonia is aerobically oxidized to nitrite or nitrate is called nitrification. Full nitrification to nitrate requires a significant amount of oxygen (theoretically 4.57 mg O_2 mg⁻¹ N) and its effective distribution to the nitrifying microorganisms (e.g. > 1 mg O_2 L^{-1}) is crucial for minimizing the formation of unwanted byproducts such as nitrite, free nitrous acid (HNO₂), nitrous oxide (N₂O) and dinitrogen gas (N₂).

189 The first step (nitritation) consists of the oxidation of ammonia over hydroxylamine (NH_2OH) to 190 nitrite (NO_2^-) and is catalyzed by sequential action of ammonium mono-oxygenase (AMO) and 191 hydroxylamine oxidoreductase (HAO). Nitritation is typically performed by 192 chemolithoautotrophic ammonia oxidizing bacteria (AOB) and archaea (AOA). Some 193 heterotrophic organisms are also able to oxidize ammonium to nitrite, but without energy 194 conservation [59]. The second step (nitratation) is the oxidation of nitrite to nitrate (NO₃⁻) by 195 nitrite oxidizing bacteria (NOB), catalyzed by nitrite oxidoreductase (NXR). Complete ammonia 196 oxidation (comammox) to nitrate in one organism has only recently been described [60] but it is 197 not clear yet whether it will be applicable for a BLSS where a nitrification system with a high 198 volumetric conversion rate is recommended.

Proton production during nitritation requires a buffering capacity of 71 meq (mg N)⁻¹ to maintain the pH. If the buffer capacity of the medium is not sufficient, only partial nitrification can be expected [61, 62], unless pH control with base addition is foreseen [63]. Biological nitrification activity is typically not observed if the pH drops below 5.5-6.0, although specific AOB adapted to low pH have been described to oxidize ammonium down to pH 2.6 [64].

204 For BLSS, nitrification systems with an open, mixed microbial community are typically proposed 205 [62, 63, 65-68]. On one hand, these self-organizing microbial communities can evolve and adapt 206 to changing conditions and microbial invasions, which increases the robustness of the system. On 207 the other hand, such complex microbial communities and interactions are difficult to capture in a 208 mechanistic mathematical model, which might be required for space application where a high 209 level of predictability is desired. Also, the use of a microbial community with unknown species, 210 which might be pathogenic, is highly undesired in confined spaces as it presents health hazard to 211 the crew members of a BLSS.

For this reason, in the MELiSSA loop, nitrification of inorganic streams is envisaged to be carried out by an 'axenic' synthetic co-culture of the AOB *Nitrosomonas europaea* ATCC 19718 and of the NOB *Nitrobacter winogradskyi* ATCC 25391 [69]. Given the slow growth of these bacteria (maximum specific growth rate is about 1.6×10^{-5} s⁻¹ for *Nitrosomonas* and 1×10^{-5} s⁻¹ for *Nitrobacter*), the biomass is immobilized on carriers to minimize washout. The complexity of the reactor construction and operation increases when such synthetic co-cultures are used, but the specific conversion rates obtained (1.7-1.9 g N m⁻² d⁻¹ or 0.55-0.59 g N L⁻¹ d⁻¹[69]) are in the range of terrestrial biofilm-based wastewater nitrification systems using an open, mixed microbial community.

221 As a significant fraction of the nitrogen in the waste of a life support system will be present in the human urine, recovery of this nitrogen is a prerequisite. Besides organic nitrogen (\sim 5-8 g N L⁻¹) 222 and inorganic nitrogen (~0.4 g N L⁻¹), fresh urine contains organic compounds (~9 g COD L⁻¹) 223 and elevated levels of salts (~21 mS cm⁻¹). As the uncharged urea gets hydrolyzed, ammonium 224 225 and bicarbonate ions are being formed which further give rise to very high electrical conductivity levels (> 70 mS cm⁻¹), especially if base (e.g. NaOH) is added to stabilize the pH. It has recently 226 227 been shown that complete nitrification can occur in a nitrification reactor fed with undiluted urine 228 at this level of electrical conductivity [63]. In case a synthetic co-culture is used for urine 229 nitrification, specific heterotrophic strains will have to be introduced as well to oxidize the 230 organic compounds.

One of the challenges of applying nitrification in BLSS in space, is the survival and storage of the strains during launch and space flight. Two recent experiments demonstrated that nitrifying pure strains, synthetic co-cultures as well as mixed nitrifying microbial communities could successfully be reactivated after spaceflight in low earth orbit (Ilgrande *et al.* in preparation). Reactivation of denitrification and anaerobic ammonium oxidation activity could also be demonstrated after a space flight in orbit. A nitrification biofilter on lava grains was incorporated 237 in the C.E.B.A.S. mini module in which an animal compartment with fishes and snails was 238 combined with a plant compartment with rootless water plants [70]. During the STS-89 flight 239 experiment, which lasted 9 days, the chemical water parameters remained within the limits of 240 what was considered to be a good water quality [71]. During other short term experiments (up to 241 17 days) flown on IML-2 (STS-65 – AAEU facility) and on the Neurolab missions STS-90 and 242 STS-95 (VFEU for marine fish), nitrification could be demonstrated in the bacterial filter of a 243 space aquarium to convert the ammonia excreted by fish into nitrate [72]. To enable long-term 244 experiments (up to 90 days) in the ISS, the 'Aquatic Habitat' (AQH) is being developed to 245 combine ammonia oxidation with heterotrophic denitrification to maintain both ammonium and 246 nitrate concentrations low for an optimal water quality in the aquarium.

247 Another challenge for the application of nitrification in microgravity conditions, which 248 significantly affects fluid dynamics [73-75], is the reduced convection and strong cohesion forces 249 in space which make efficient gas-liquid interactions no longer possible with major consequences 250 for aeration [73, 75, 76]. Due to the lower diffusion rate of oxygen in water, the oxygen transfer 251 rate might become problematic. Therefore, an aerated rotating membrane bioreactor system 252 (ARMS), has been developed at Kennedy Space Center. It contains rotating hollow fiber membranes for the enhanced diffusional transfer of oxygen, which would avoid gas-liquid 253 254 interactions under microgravity conditions [76, 77].

Besides biological nitrification, chemical nitrification is also possible by combining pure ammonia with oxygen at 700-800°C in the presence of a catalyst (Platinum with 10% Rhodium; Ostwald process) to produce nitric acid (HNO₃) [78]. Also, ammoniumnitrite which can be biologically produced at low pH values, spontaneously decomposes into N_2 and H_2O , while nitrite can also be chemically oxidized, e.g. nitrous acid (HNO₂) self-decomposition into nitrate [64]. 260

3. Production stages: from fertilizer to protein

262 Human daily dietary requirements have been estimated several times leading to varying dietary 263 recommendations. Most of the differences in food requirements are attributed to the parameter 264 settings regarding initial body weight, age, gender, physical activity etc. The major part of the 265 energy supply in the food is typically derived from carbohydrate sources. The requirement for 266 moderately active females (19-30 years old) is 1500-2500 kcal, while males of the same age 267 need 2500–3300 kcal. Energy is also extracted from protein and lipids but this should be limited 268 (for protein about 10%) as these are less favorable energy sources and are mainly necessary for 269 replacing metabolic building blocks that cannot be autonomously synthesized. Indeed, nine 270 amino acids found in human tissues must be supplied by dietary sources that need to be digested 271 in our intestinal tract. It is therefore not recommended to target a certain daily protein intake as 272 such, but instead focus on the quality of the protein. The US Food and Drug Administration and 273 the Food and Agricultural Organization of the United Nations (FAO/WHO) have adopted the 274 Protein Digestibility Corrected Amino Acids Score (PDCAAS) to assess this [79]. Based on this 275 method, there is strong variability between different protein sources. As astronauts will have to 276 adopt a rather monotonous diet based on non-animal sources, special care must be taken to ensure 277 certain limiting essential amino acids and vitamins are provided. However, Young and Pellett 278 [80] demonstrated that a mixture of plant proteins can serve as a well-balanced and complete 279 source of amino acids to effectively meet human dietary requirements. As such, it can be 280 hypothesized that a clever blend of plant, fungal and microbial proteins could meet all nutritional 281 requirements for crew members in space.

282 As a result, the production of animal protein is not strictly necessary in a BLSS and a vegan diet 283 is most likely to be adopted in an eventual self-sustaining BLSS in space. Moreover, the low 284 conversion efficiency from feed to meat would entail an excessive need for feed production 285 whereas the requirements for volume, mass, nutrients, energy and water demand in a BLSS 286 should be minimized. Furthermore, the difficulties with waste management (e.g. defecation, 287 slaughter, ...) in reduced gravity for typical tetrapod meat sources (chicken, goat, pig, ...) might 288 be impossible to solve [70]. There might also be ethical objections against husbandry of sentient 289 animals in confined spaces, but this discussion is outside the scope of this review.

Breeding fish as a source of animal protein for human consumption in combination with the cultivation of edible higher aquatic plants which float or buoy (e.g. hornweeds and duckweeds) and waste degrading microorganisms has been proposed by Blum *et al.* [70, 81] since conceptually oxygen transfer and partial nutrient recovery can occur without phase state changes (gas-liquid).

The conversion of inedible organic waste residues by edible worms (e.g. mealworms, silkworms, earthworms) has not only been proposed to generate 'soil-like substrates' (SLS) or 'biohumus' for plant production [82], but also as a source of animal protein in BLSS [8, 14, 23, 24, 83]. In this review we will focus only on the protein production through plants, microorganisms, and fungi.

300

301 3.1 Food production based on plants

302 Researchers and space agencies have historically focused on the cultivation of higher plants to 303 produce food for crew members in future BLSS [8, 23, 84, 85]. On a physiological level, 304 ammonium is required for synthesis of amino acids and proteins as it is directly incorporated into 305 the glutamate – so whatever the source, the plant must be able to ultimately convert it to 306 ammonia. Nitrogen is typically taken up from the environment as nitrate, ammonium or urea by 307 plants, but alternative forms of nitrogen can be used that include mostly organic forms [86]. 308 Preference towards either form may exist for specific plants, but usually ammonium and urea are 309 used during early crop development and nitrate during later stages. In soil, microorganisms 310 rapidly mineralize the organic matter and thereby release nitrate and ammonium. When applied 311 in conventional hydroponic systems, organic fertilizers are known to cause growth inhibition and 312 have phytotoxic effects. Plant growth chambers for life support systems use conventional 313 hydroponic cultivation systems and are, therefore, not suitable for the direct use of organic 314 fertilizers. Another complication associated with organic fertilizers is that the dosing cannot be 315 controlled to accommodate the nitrogen requirement of crop plants throughout the growth cycle. 316 Usually high nitrogen levels are required during the early growth phase while lower levels near 317 the harvesting time promotes greater yields [87].

Direct addition of human urine, which harbors the majority of nitrogen in a BLSS (Figure 1), to hydroponic systems has been proposed and evaluated [84, 88-90]. The major problems with direct urine addition include phytotoxic effects of high ammonium, problems relating to organic compounds, and stress caused by the high salt concentration in urine. Several strategies have been proposed to cope with the high salinity. Dilution can be used to maintain the salt level below a specific threshold, but on the longer term salt accumulation might still occur in hydroponic systems if the plants don't take up these salts at the same rate as other nutrients in the 325 urine. Other strategies that have been proposed include using halophytes [91], or maintaining a 326 low salt diet for the crew members to reduce the salinity in the urine in combination with plants 327 that can accumulate salt in their leaves [92]. Physicochemical techniques that have been proposed 328 for salt removal include hydrothermal treatment of the urine [93] and electrodialysis of irrigation 329 water [91, 94].

330 Given these issues, to reach optimal crop growth, hydroponic systems should be supplied with 331 fully synthetic mineral mixtures which can be monitored and adjusted in accordance to the 332 system requirements. Ammonium and nitrate are usually supplied as a mixture since ammonium 333 uptake acidifies while nitrate basifies the nutrient solution. The amount of ammonium added to 334 the nutrient solution should not exceed the absorption and storage capacities of the plant because 335 excess ammonium causes cell damage. Most plants can directly take up urea but metabolize it 336 with varying efficiencies, depending on the species. Under field conditions, urea is converted into 337 ammonium by urease, a common enzyme in soil microorganisms, while under hydroponic 338 conditions this does not occur and urea has been shown to be a very poor nitrogen fertilizer [95].

339 The total amount of nitrogen taken up by a plant depends on various environmental factors, two 340 of the most important being concentration and dose of nitrogen that is supplied. Indeed, shoot 341 biomass strongly increases with a generous dose of nitrogen. Under field conditions, the nitrogen 342 cycle is strongly influenced by ammonium evaporation and leaching and to some extent to biotic 343 nitrogen fixation and atmospheric deposition. There are ample studies on nitrogen use efficiency 344 (ratio of grain yield over available N) in wheat under field conditions as it is an important 345 economic factor and determines end-use quality. In a recycling hydroponic system, volatilization 346 of ammonium can be kept to a minimum either by closing the nutrient solution circuitry or by 347 supplying nitrogen as nitrate. Beyond nitrogen, different plant species show variation in mineral

348 uptake efficiency, which may cause salinity issues in hydroponic systems. To compensate for 349 this, a life support food production system should include a variety of crops, collectively 350 cultivated with the same hydroponic solution such that optimal mineral use efficiency is attained.

351 Next to the cultivation requirements imposed by the choice of crops, food production should 352 fulfill the needs of the crew. Continuous supply of fresh produce is achieved by a staggered 353 cultivation scheme whereby harvesting is possible on a daily basis. Although this approach offers 354 advantages in terms of low storage requirements and preservation of food quality, estimations on 355 crop production requirements usually assume one crop per plant growth unit, and batch culturing 356 [96]. Biomass and food production levels are strongly dependent on the type of crop and, 357 consequently, the cultivation area is largely determined by the food quality requirements. A daily 358 consumption of 3000 kcal for a male astronaut executing moderate physical activities can be 359 provided by a balanced diet of macronutrients of 70% carbohydrates, 10% protein and 20% fat. 360 The majority of the carbohydrates are likely going to come from wheat and potatoes. Wheat 361 grains are high in starch content (60-70% carbohydrate), and, therefore, ideal to provide energy. 362 Potato tubers are also high in starch (85% of the dry matter), but, in addition, are rich in minerals 363 and vitamins and therefore often combined with other carbohydrate sources such as wheat. A 364 suitable crop composition, however, has to provide nutritional complementarity and variety, 365 which can require combining more than 10 crop species [97]. Balancing cultivation requirements 366 and yield potential of different crops with human nutritional requirements is a complex problem 367 for which different outcomes have been calculated, in terms of minimizing cultivation area, 368 optimizing biomass production rates, crop selection, etc. [98]. Typically, the required surface area 369 for higher plant production for one crew member has been assumed to be in the order of 40-50 m² 370 [8]. However, Do et al. [99] have postulated that, depending on the desired variety of crops, 46-

117 m² would be a more realistic estimate for a 3040 kcal person⁻¹ d⁻¹ plant based diet. As a 371 372 comparison, Cassidy et al. [100], estimated that approximately 1000 m² would be necessary to provide one person with a plant based diet on earth (2700 kcal⁻¹ person d⁻¹, 41 different crops), 373 374 not accounting for animal feed, biofuels, nor other non-food products. It can be assumed that the 375 scale of a higher plant compartment will have a great impact on the feasibility of realizing a 376 future closed loop BLSS for the first time. Therefore, first plant production systems for human 377 consumption in the near future will most likely cover only a small fraction of the diet, 378 supplemented with terrestrial resupply.

379 With these different opinions on the required cultivation area in mind, a general estimation of the 380 nitrogen consumption by the cultivated crops is equally difficult to assess with current 381 information. Nonetheless a rough estimation can be made. Considering an average productivity between 10 and 200 g fresh weight m⁻² day⁻¹ [8, 101], and assuming that about 0,5% of fresh 382 383 green parts of the plant consists of N, the daily consumption of nitrogen per m^2 is estimated to be 384 5 to 40 grams. In case a LED based artificial lighting system is used (~2200 kWh electricity kg⁻¹ 385 dry weight, calculated from Do et al. [99]), several layers of crops can be stacked in a higher 386 plant growth chamber, so an average specific protein production rate would be, depending on the crop, in the order of 0-4 kg protein m^{-3} year⁻¹ (calculated from Do *et al.* [99]). 387

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3.2 Food production based on microorganisms and fungi

The term 'single cell protein' (SCP) has been coined to describe whole cells of microalgae, bacteria and unicellular fungi (such as yeast) intended for consumption for nutritional purposes [102, 103]. Next to SCP, multicellular fungi are an interesting source of protein due to their versatility in substrate usage. This feature makes fungi applicable in recycling the inedible lignin
rich biomass of higher plants in BLSS [14, 82, 104].

395 3.2.1 Bacterial and microalgal single cell protein production

396 SCPs are highly attractive for protein production on earth as well as in space because they can be 397 produced with highly efficient use of resources. Indeed, microorganisms can be highly efficient 398 in their nutrient conversions, and thus near full recycling of the nitrogen and phosphorus from a 399 recovered stream is feasible, depending on their ratio. One important feature of SCPs is their high volume or surface specific productivity rates, as high as 1.5 kg dry weight m⁻³ d⁻¹ for microalgae 400 401 cultivation in indoor photobioreactors [105] and 1.45 kg dry weight m⁻³ d⁻¹ for purple non-sulfur 402 bacteria (PNSB) cultivated in outdoor photobioreactors [106], resulting in a compact type of 403 engineered food production system. In case artificial light is necessary (e.g. LED), the electrical 404 energy demand for lighting can be in the order of 180 kWh kg⁻¹ dry weight (assuming 135 μ mol $m^{-2} s^{-1}$; 1.1x10⁻² kg dry weight $m^{-3} h^{-1}$; 25 m⁻¹ [107] and 1.7 μ mol J⁻¹ [108]). Finally, given their 405 406 wide range of metabolic capacities, they are able to use various sources of carbon, energy and 407 electrons, offering multiple options to plug them into resource recovery loops. Microbial biomass 408 has the highest protein content among organisms (50-70% DW) and offers additional functional 409 benefits in terms of vitamins, pigments and potentially prebiotic compounds [102, 103].

SCP is facing three important challenges that must be overcome to further improve its cultivation, nutritional value and dietary consumption. Firstly, producing a semi-solid food ingredient from microbial biomass requires some high energy processing steps. In photobioreactors for instance, biomass levels are around 1 g dry weight L⁻¹ [109], hence 99.9% of the reactor content consists of water. Typical techniques to concentrate the biomass are energy intensive, including filtration, 415 centrifugation (~2 kWh kg⁻¹ dry biomass [110]) and drying [111]. Secondly, microorganisms 416 contain the highest levels of nucleic acids (DNA+RNA) among all organisms, about 15-16% of 417 the dry weight [103], mostly in the form of RNA. For humans, consumption above 2 g per person 418 per day can lead to gout and kidney stone [112]. Due to these health-related issues result in the 419 fact that only 40-50% of the protein intake can be substituted by microbial protein [113, 114]. 420 Although this is already a considerable portion, various physical and chemical post-treatment 421 techniques can lower the SCP RNA levels increasing their usage potential in a healthy diet. A last 422 challenge, and often overlooked challenge is that of human acceptance of microbial food 423 products. Humans already consume items produced by microorganisms that often also contain 424 residual microorganisms (e.g. beer, cheese, yoghurt...), yet microorganisms as a main food 425 ingredient remains a novel idea, and one that consumers may find repugnant. Unappealing flavor 426 and texture of food containing microorganisms are likely causes of this distaste. In long-term 427 spaceflight, it is known that the psychological well-being of the astronauts is partially depending 428 on food flavor and variety [115]. In addition, possible rejection and concerns about product safety 429 can be linked with psychological aspects of consuming a novel food that is produced with 430 innovative technologies.

The microorganisms *Arthrospira platensis*, also known as "*Spirulina*", and *Rhodospirillum rubrum* have been studied in the MELiSSA concept as an alternative source of proteins [116]. *A. platensis* biomass contains as much as 46-71% protein on a dry weight basis [117-119] and the biomass is rich in vitamins, minerals, β -carotene, essential fatty acids, such as γ -linolenic acid [120], and antioxidants [121]. The true protein digestibility of this microorganism is 75.5% [122], which is fairly high. In terms of essential amino acids, cultivation can lead to a nutritionally appealing composition for crew members, yet the opposite is also possible. Figure 3 displays the 438 variability of levels of individual essential amino acids, with a factor seven difference between 439 the minimum and maximum values, depending on cultivation conditions. Further optimization in 440 this area should be able to guarantee the protein quality. A practical advantage of A. platensis is 441 its relatively large size, as multiple cells (3-16 µm [121]) form spiral-shaped filaments (100-200 442 μ m [123]) which can more easily be separated, for instance through filtration. Furthermore, this 443 cyanobacterium is one of the rare bacteria containing a relatively low nucleic acid concentration 444 (4-6% of dry weight [114]). For nutritional purposes, A. platensis is being cultivated at full-scale 445 on synthetic chemicals, while other applications have also explored the use of waste streams to 446 support growth [109, 124].

447 A. platensis fulfills a vital role in the photosynthetic compartment (IVa) of the MELiSSA loop 448 [116]. In this loop, A. platensis consumes nitrate in nitrified effluents, consumes CO₂ produced 449 by by the crew and in the other compartments, produces O_2 and provides a dietary protein source 450 for the crew members. Previous studies towards the growth of A. platensis on urine reported 451 success using highly diluted human urine [125, 126] and nitrified urine [127]. The high dilutions 452 used (10-30% dilution) however, indicates that those systems were operated at low salinities. 453 Coppens et al. [63] observed that relevant salt levels had no significant effect on its growth and 454 nitrate resulted in the most optimal growth compared to ammonium and urea as the sole nitrogen source, provided that the nitrate concentrations did not exceed 1 g N L⁻¹. When growing a food 455 456 product on urine, particular attention needs to be directed towards safety with regards to 457 pathogens and micropollutants such as pharmaceutical residues [128].

458 Next to *A. platensis*, the cyanobacterium *Aphanizomenon flos-aquae* is grown commercially as a 459 food supplement [129]. This species, however, was never considered for use in a BLSS. Another 460 cyanobacterium, which lives in symbiosis with the edible water-dwelling fern Azolla, is 461 *Anabaena azollae*. Liu *et al.* [90] studied this organism as part of a BLSS that also included 462 biological purification of a urine solution using Azolla and a UV photocatalytic oxidation by 463 TiO₂. According to the authors, this treatment could produce effluent to meet standards for 464 drinking water quality.

465 Furthermore, the green microalgae Chlorella vulgaris is cultivated commercially and sold as a 466 food supplement. Although Chlorella contains a high protein content of 51-58% on a dry weight 467 basis, it also possess a robust cellulosic cell wall which makes human digestion difficult [119]. 468 Chlorella has been proposed for O₂ generation in a BLSS rather than for food production. To 469 prevent accumulation of dead-end products in a BLSS, the biomass could be burned when the 470 CO₂ concentration is low, or fed to mealworms (Tenebrio molitor L.) [130]. Another use for 471 *Chlorella* in BLSSs was proposed by Li *et al.* [130] in which *C. vulgaris*, grown on human urine, 472 was used to control the balance of CO_2 and O_2 .

473 Rhodospirillum rubrum, which has been proposed for compartment II of the MELiSSA loop, 474 belongs to the group of the purple non-sulfur bacteria (PNSB), and grows photoheterotrophically 475 on VFAs. As indicated by Godia et al. [116], PNSB represent an interesting yet unexplored 476 source of microbial protein for consumption. The potential for producing these bacteria on 477 fermentation effluent derives from their metabolic and physiologic features. Firstly, they have a 478 near perfect organic carbon uptake efficiency (when grown on volatile fatty acids) in comparison 479 with other organoheterotrophs, yielding a very high protein output per carbon input [131]. 480 Secondly, PNSB also have a high growth rate with respect to photoautotrophs like microalgae, 481 leading to a more compact production system [132, 133]. Thirdly, their unique potential to grow 482 under specific infrared wavelengths adds a selectivity tool for non-axenic cultivation applications 483 [134, 135]. In terms of nutritional quality, the PNSB biomass composition has other appealing

484 features. Firstly, its methionine content, usually a limiting essential amino acid, is much higher in 485 comparison with other types of microbial protein. Secondly, the biomass contains a considerable 486 amount of essential vitamins (B12, B2, B6, C, E, D and folic acid) [136]. Finally, compounds 487 with health stimulating benefits are present, such as carotenoid pigments [113, 137, 138] and 488 poly-\beta-hydroxybutyrate (PHB) [139]. R. rubrum has been demonstrated to reduce for instance 489 LDL-cholesterol in mice [140]. Despite of several animal feeding trials, for instance by Banerjee 490 et al. [141], Shapawi et al. [142] and Kobayashi and Kobayashi [136], PNSB have not yet been 491 tested for human consumption, to the authors' knowledge. It is unclear whether full-scale 492 production facilities are terrestrially in place at this moment.

In the MELiSSA loop, the main challenge with the use of *R. rubrum* as a food source for crew members is its proximity to Compartment I. As fecal matter, and the potentially associated pathogens, enter Compartment I directly, it might be a challenge to maintain a sterile fermentation effluent through membrane filtration at all times.

497 Besides the phototrophic protein production by bacteria or microalgae, the dark production of 498 single cell proteins for life support systems by chemolithoautotrophic bacteria has been proposed 499 by Hendrickx and Mergeay [143] and Verstraete et al. [144] for BLSS. Methane or hydrogen 500 formed in the BLSS (e.g. in anaerobic fermentor) or from outside the BLSS could be used as 501 energy source for non-phototrophic microbial food production. Hydrogen could be produced 502 from olivine found on Mars [143]. Matassa et al. [145] recently reported a hydrogen oxidizing biomass production rate of 0.28 g dry weight g⁻¹ COD-H₂ (81% H₂ conversion efficiency, 76% 503 504 protein content), so even if hydrogen would be generated through water electrolysis (e.g. 53 kWh 505 kg^{-1} H₂ [146]), the electrical energy demand for water electrolysis would be in the order 29 kWh

506 kg⁻¹ dry weight, which is lower than the electricity demand for phototrophic growth with artificial
507 lighting systems.

508 3.2.2 Fungal protein production

In general, fungi contain a lower protein content (15-45% DW) in comparison with the earlier discussed microalgae and bacteria. However, both unicellular and multicellular fungi are valued for their ability to grow on a variety of lignocellulosic substrates. The exact composition of hemicellulose, cellulose and lignin, will influence which fungi can be cultivated [103].

513 Gitelson et al. [147] could convert approximately 10% of the courser part of the inedible 514 biomass, which was ~45% of the total inedible biomass, into edible fungal biomass (Pleurotis 515 florida; protein content of 16%). When grown for 60-70 days on wheat straw, 6-7% of the 516 substrate could be converted into *Pleurotis florida* Fovose having a protein content of 23% [23]. 517 Use of fungi in the biological waste treatment has typically been combined with other biological 518 treatment steps, e.g. organic waste digestion with worms and bacteria to produce 'soil-like 519 substrates' as an organic matrix for subsequent plant production [82]. He et al. [14] used 520 aerobically fermented residues of wheat and rice to cultivate oyster mushrooms (Pleurotus 521 ostreatus) and then earthworms. Although the degradation efficiency of cellulose and lignin 522 slightly increased from 98.6% to 99.5% and from 93.1% to 98.6% respectively by the addition of 523 oyster mushroom to the earthworm treatment, the produced soil-like substrate resulted in a lower 524 lettuce productivity afterwards and the authors did not consider all the oyster mushroom biomass 525 as food as the lignocelluloses in oyster mushroom cannot be digested well by humans.

526 One drawback in the production of fungi is the possible contamination with fungal species that 527 produce mycotoxins. Small amounts of these mycotoxins can potentially cause neurotoxicity, allergies and rashes on the skin. The monitoring of invasive species and their potential toxins is aprerequisite in producing fungal biomass [148].

To recycle the inedible biomass of higher plants in a BLSS, Strayer *et al.* [149] investigated the production of the yeast *Candida ingens* on volatile VFAs, which were produced in a first step using anaerobic digestion of potato crop residue. A low pH of 5 was maintained to favor growth of the yeast while slowing bacterial growth. On average 0.4 g yeast biomass was produced per g VFA consumed and no phytotoxic VFA were detected in the effluent, indicating their complete removal.

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539 4. Nitrogen loss, fixation and in situ utilization

540 One of the main challenges of the nitrogen recycle in BLSS, is to avoid nitrogen losses. Main 541 points of loss in these systems can be identified. First, if organic waste is disposed or removed 542 from the BLSS after dewatering without further treatment nor recovery, the nitrogen fixed in this 543 organic material is not available anymore [10, 52, 84, 91, 150-155] (Figure 1, c1). Several other 544 nutrients can be recovered after incineration of the dried organic waste, but the nitrogen is 545 typically converted into N₂ and NO_x [22, 84, 151, 154-157] (Figure 1, c2) and lost in the process. 546 Depending on the temperature and pressure, hydrothermal treatment of organic waste also results 547 to some extent in the formation of N₂ and NO_x [50-52, 93, 150, 158] (Figure 1, c3). When nitrate, 548 e.g. from inedible crop residues, is introduced in an anoxic bioreactor in the presence of organic compounds, denitrification is very likely to occur [25, 27, 62, 67, 77, 159] (Figure 1, c4). For specific mission scenarios where complete nutrient recovery is not mandatory, the formation of N_2 might be useful to help pressurizing the cabin [160], but not for a BLSS where a (near) complete recycling of all the elements is envisaged. During nitrification and denitrification, substantial N₂O formation might occur in case of specific process conditions (e.g. oxygen concentration, nitrite concentration, COD/N ratio) [161], which can also become a potential health hazard in a closed system.

556 Other examples of nitrogen losses that can occur in a BLSS include ammonia volatilization (e.g. 557 from urine at high pH), nitrous acid volatilization at low pH values, and struvite formation in 558 vessels or piping e.g. as a result of urea hydrolysis. Nitrogen can also be lost from the concentrate 559 of membrane filtration steps or the biosolids waste from various bioreactors if these streams are 560 not adequately recycled.

In case it is not possible or too complex to avoid the formation some of these nitrogen losses and if terrestrial resupply is not possible or desirable, the introduction of a N_2 fixation step can be a solution. Nitrogen fixation in BLSS can be established in a microbial or chemical way.

Biological N_2 fixation can occur in the root nodules of specific plants (e.g. legumine family) via a symbiosis between nitrogen fixing bacteria and plants. It might be necessary to select bacterial strains compatible with specific plants may need to be selected and inoculated to achieve a high nitrogen-fixing efficiency [162]. However, biological nitrogen fixation requires a high energy input (15-16 ATP per molecular N_2) and the nitrogen demand of plants cannot be fully satisfied only by biological nitrogen fixation so nitrogen-fixing plants may be less productive compared to those supplied with mineral nitrogen [162, 163]. Besides bacterial symbiosis with plants, free 571 living microorganisms such as *Azotobacter, Clostridium* and specific PNSB and cyanobacteria,
572 are capable of biological nitrogen fixation. Cyanobacteria that combine nitrogen fixation with
573 oxygenic photosynthesis in a dedicated bioreactor have been proposed for BLSS [164].

Nitrogen can also be fixed as ammonia in a chemical way by combining dinitrogen and hydrogen gas in the presence of a catalyst at high temperature and pressure [93]. Whereas industrial ammonia production typically occurs at 200-300 bar and 400-500°C (Haber-Bosch process) with the energetic cost of 27 GJ t⁻¹ NH₃, Sakamoto *et al.* [78] demonstrated a nitrogen fixing unit for a Life Support System that could be operated at 7-8 bar and 300°C at a rate that was in the order of the nitrogen metabolism rate of one crew member.

580 In specific environments where nitrogen may already be present (e.g. Mars), the in situ utilization 581 of nitrogen could be considered to compensate for nitrogen losses in a BLSS. Oxidized nitrogen-582 bearing compounds were recently discovered in Martian sedimentary deposits, which indicates 583 the theoretical possibility to extract nitrate from Martian surface [165]. However, separating the 584 nitrogen from other, possibly toxic, compounds (e.g. perchlorate) might be challenging. To deal 585 with pressure losses of the habitat atmosphere caused by leakage and extravehicular activity from 586 habitat on Mars, Do et al. [99] suggested an atmospheric processor module to capture high purity 587 N₂ from the CO₂-rich Martian atmosphere with the utilization of a zeolite membrane filter and 588 cryocooler.

589

5. Broader challenges for nitrogen cycling in BLSS

590 Besides crew members, a BLSS contains by definition living organisms, at least for food 591 production and, depending on the BLSS outline, possibly also for waste processing. In space, 592 living organisms are subjected to extreme and hostile conditions including broad ranges of

593 gravity and ionizing radiation, both factors known to have a profound effect on the physiology, 594 morphology and functionality [73, 166-168]. Without sufficient protection (shielding), radiation 595 can induce random DNA damage and mutations, possibly causing alterations in gene expression, 596 which can be detrimental for the growth and survival of microorganisms and plants [73, 74, 169, 597 170]. Exposure to variations in gravity, ranging from 'hypergravity' during vehicle launch (3.2 g) 598 and reentry (1.4 g), to reduced gravity at the lunar (0.17 g) and Martian (0.38 g) surface and even microgravity during orbital flight (~10⁻⁶ g), brings about mainly indirect effects linked to the 599 600 altered extracellular fluid properties affecting the nutrient acquisition and waste removal [73, 74, 601 166, 167, 170-172]. In case extraterrestrial habitation becomes possible, genetic evolution of 602 terrestrial species into novel extraterrestrial species is likely to occur over time and the different 603 levels of gravity and the specific composition of the radiation penetrating the shielding of a BLSS 604 in space might affect this process in a different way than on earth. As a result, in depth research 605 on the long-term effects of radiation and microgravity on living organisms in a BLSS in general, 606 and on nitrogen metabolism specifically, will be mandatory for permanent extraterrestrial 607 habitation.

Besides selecting and optimizing the different subprocesses in a BLSS for nitrogen and other elements, the integration and real time control over the different processes is the final hurdle still to overcome to allow for the realization of a BLSS for reuse of nitrogenous compounds for food production in space. As an example, the integration of the MELiSSA compartments is being thoroughly tested in the MELiSSA Pilot Plant facility, located at the Universitat Autònoma de Barcelona [173, 174]. Moreover, each system developed for space should eventually comply with the basic rules for space design, including limited volume and mass, low consumable and energy 615 input, a high level of automation to limit the workload of the crew and a high level of robustness,616 reliability and safety.

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618 **6. Outlooks**

619 When does a BLSS, with at least partial food production, become favorable for space missions? 620 Olson et al. [175] estimated that the cumulative cost and mass savings for a BLSS with 50-97% 621 food production compared with a PLSS (in situ water recycling and O₂ production, terrestrial 622 food supply) start to pay of between 1 and 12.9 years depending on the mission scenario. Do and 623 Owens [99] assessed the technical feasibility of the 'Mars One' mission plan (missions with 4 624 crew members every 26 months for a one way trip) and evaluated the payload requirements for 625 supplying terrestrial produced food versus in situ food production via BLSS. They concluded that 626 for the first seven missions, the costs and mass to be delivered to Mars were higher for the BLSS 627 with in situ food supply, but that the mass and costs for resupply (food and spare parts) increased 628 faster for the scenario with terrestrial resupply for an increasing Mars population. They also 629 concluded that the assumptions made by 'Mars One' do not lead to a feasible mission plan and 630 that further technological development (e.g. life support reliability) is mandatory to achieve the 631 goal of sustainable habitation on Mars [99]. The BLSS scenario with in situ food production in 632 this assessment did, however, not take the recovery of nutrients from crop residues and human 633 organic waste stream into account for food production. Czupalla et al. [152] evaluated different 634 BLSS concepts (MELiSSA, BIOS, ALM and an Equivalent System Mass (ESM) optimized life 635 support system) with partial nutrient recovery for a 780 day mission to Mars based on an ESM 636 analysis. In their ESM optimized life support system, the use of the Kudenko process (see section 637 2.2) would allow the partial reuse of nitrogen for the plant production. Further research is 638 however needed to perform more enhanced assessments since the mass criterion of ESM is rather 639 minimal and the outcome of the assessment might differ significantly depending on the duration 640 (long-term or permanent BLSS) and the destination of the space mission. Furthermore, increasing 641 the fraction of microbial food production in the crew diet might enhance the feasibility of a BLSS 642 with in situ food production. Moreover, hybrid scenarios combining in-situ food production with 643 terrestrial food resupply will most likely be considered for near future long term space missions 644 and space habitation and these scenarios can be stepping stones towards self sustaining colonizations in outer space. 645

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647 These scientific and technological endeavors for enhanced nitrogen recovery in BLSS are not 648 only favoring future extraterrestrial civilizations, they also drive technological innovation on 649 earth in areas such as bioprocessing, circular technology, cleantech and life sciences, among many others [73, 176, 177]. On the one hand, studies on the behavior of microorganisms and 650 651 higher plants in space can provide new insights into more fundamental microbiology and plant 652 biology [73, 178, 179]. On the other hand, spin-off terrestrial technologies can open up 653 opportunities for terrestrial life support in confined spaces or extreme conditions as well as in 654 recycling schemes for sustainability concepts (eco-cities, eco-buildings, ...) [74, 176]. Potential 655 fields of application where these nitrogen reuse technologies can help close or balance the 656 nitrogen loop on earth include treatment and valorization of source separated urine, decentralized 657 water treatment, aquaponics, vertical farming and SCP feed or food production.

659 **7. Conclusions**

660 In the context of BLSS, the focus for recovery is often mainly on water, oxygen and carbon while 661 other (micro)nutrients receive little attention. In this review, the state-of-the-art and remaining 662 challenges of the different technologies and subsystems with respect to nitrogen is presented and 663 discussed. Efficient nutrient management will become mandatory for a sustainable and quasi-664 independent extraterrestrial colonization of human beings, as resupply for increasing populations 665 will not be feasible anymore. Therefore, more research and technological BLSS development 666 focus should be oriented to minimize and remediate the losses of nitrogen, along with other 667 essential elements. 668 669 670 671 Acknowledgements 672 673 This article has been made possible through the authors' involvement in the MELiSSA project, 674 ESA's life program support system research 675 (http://www.esa.int/Our Activities/Space Engineering Technology/Melissa). We kindly 676 acknowledge the financial support of the Belgian Science Policy (BELSPO).

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Figure 1. Schematic diagram of the theoretically calculated nitrogen balance in a BLSS for one





Figure 2. Schematic overview of the key processes for nitrogen recycle in Bioregenerative Life
Support Systems

- **689** (a1 [22, 84, 153-157]; a2 [84, 151, 152, 155, 157, 180]; a3 [10, 152, 155]; b1 [22, 84, 153, 154,
- 690 156, 157]; **b2** [84, 151, 155, 157, 180]; **c1** [10, 52, 84, 91, 150-155]; **c2** [22, 84, 151, 154-157];
- c3 [50-52, 93, 150, 158]; c4 [25, 27, 62, 67, 77, 159]; d1 [9, 23, 50-52, 93, 152, 155-158, 181];
 d2 [52, 91, 93, 150, 152, 155, 182]; d3 [51, 52, 91, 93, 94, 150, 155, 182]; d4 [4, 8]; d5 [181]; e1
 [4, 8, 14, 20, 21, 23-27, 29, 66, 183-188]; e2 [4, 8, 29, 66]; e3 [62, 63, 65-67]; f1 [8, 23]; f2 [4,
- $694 \quad 8, 62, 63, 65-67, 77, 186] ; \textbf{f3} [4, 25]; \textbf{g1} [25, 62, 67, 77] ; \textbf{g2} [25, 27] ; \textbf{h1} [23] ; \textbf{h2} [4, 14, 23, 14, 23] ; \textbf{h2} [4, 14, 23] ; \textbf{h3} [4, 14, 14] ; \textbf{h3} [4, 14] ; \textbf{h$
- 695 25, 26, 63, 66]; h3 [4, 63, 66]; h4 [63, 126]; h5 [164]; i1 [8, 23, 24, 83]; i2 [14, 24]; i3 [14]; j1
 696 [4, 8, 25, 26, 66, 184-187, 189]; j2 [50, 91, 150, 151]; j3 [84, 88, 89, 190]; j4 [66, 188, 189]; j5
- 697 [21]; **j6** [23, 26]; **j7** [78, 93]; **j8** [78, 93])

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Figure 3. High variation in protein digestibility corrected amino acid scores (PDCAAS) for *A*. *platensis*, calculated on available amino acid data from cultivating at different pH and temperature [191], using a digestibility of 75.5% [122]. The line depicts the target threshold to fulfill for every amino acid for a perfect match with the human nutritional requirements.

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