

**This item is the archived peer-reviewed author-version of:**

Kinetic properties and heme pocket structure of two domains of the polymeric hemoglobin of *Artemia* in comparison with the native molecule

**Reference:**

Akbari Borhani Heshmat, Berghmans Herald, Trashin Stanislav, De Wael Karolien, Fago Angela, Moens Luc, Habibi-Rezaei Mehran, Dewilde Sylvia.- Kinetic properties and heme pocket structure of two domains of the polymeric hemoglobin of *Artemia* in comparison with the native molecule

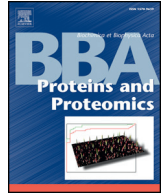
Biochimica et biophysica acta : proteins and proteomics - ISSN 1570-9639 - (2015), p. 1-10

DOI: <http://dx.doi.org/doi:10.1016/j.bbapap.2015.05.007>



Contents lists available at ScienceDirect

Biochimica et Biophysica Acta

journal homepage: [www.elsevier.com/locate/bbapap](http://www.elsevier.com/locate/bbapap)

## Q3 Kinetic properties and heme pocket structure of two domains of the 2 polymeric hemoglobin of *Artemia* in comparison with the 3 native molecule

Q4 Heshmat Akbari Borhani <sup>a,b</sup>, Herald Berghmans <sup>b</sup>, Stanislav Trashin <sup>c</sup>, Karolien De Wael <sup>c</sup>, Angela Fago <sup>d</sup>,  
5 Luc Moens <sup>b</sup>, Mehran Habibi-Rezaei <sup>a,e,\*</sup>, Sylvia Dewilde <sup>b,\*\*</sup>

6 <sup>a</sup> School of Biology, College of Science, University of Tehran, Tehran, Iran

7 <sup>b</sup> Department of Biomedical Sciences, University of Antwerp, Antwerp, Belgium

8 <sup>c</sup> Department of Chemistry, University of Antwerp, Antwerp, Belgium

9 <sup>d</sup> Department of Bioscience, Zoophysiology, Aarhus University, Aarhus, Denmark

10 <sup>e</sup> Nano-Biomedicine Center of Excellence, Nanoscience and Nanotechnology Research Center, University of Tehran, Tehran, Iran

### 1 1 A R T I C L E I N F O

#### 12 Article history:

13 Received 6 February 2015

14 Received in revised form 30 April 2015

15 Accepted 14 May 2015

16 Available online xxxx

#### 17 Keywords:

18 *Artemia* hemoglobin

19 Ligand binding kinetics

20 Heme pocket structure

21 Redox potential

### A B S T R A C T

In this project, we studied some physicochemical properties of two different globin domains of the polymeric he- 22  
moglobin of the brine shrimp *Artemia salina* and compared them with those of the native molecule. Two domains 23  
(AsHbC1D1 and AsHbC1D5) were cloned and expressed in BL21(DE3)pLysS strain of *Escherichia coli*. The recom- 24  
binant proteins as well as the native hemoglobin (A/Hb) were purified from bacteria and frozen *Artemia*, respec- 25  
tively by standard chromatographic methods and assessed by SDS-PAGE. The heme environment of these 26  
proteins was studied by optical spectroscopy and ligand-binding kinetics (e.g. CO association and O<sub>2</sub> binding af- 27  
finity) were measured for the two recombinant proteins and the native hemoglobin. This indicates that the CO 28  
association rate for AsHbC1D1 is higher than that of AsHbC1D5 and A/Hb, while the calculated P<sub>50</sub> value for 29  
AsHbC1D1 is lower than that of AsHbC1D5 and A/Hb. The geminate and bimolecular rebinding parameters indi- 30  
cate a significant difference between both domains. Moreover, EPR results showed that the heme pocket in A/Hb 31  
is in a more closed conformation than the heme pocket in myoglobin. Finally, the reduction potential of -0.13 V 32  
versus the standard hydrogen electrode was determined for A/Hb by direct electrochemical measurements. It is 33  
about 0.06 V higher than the potential of the single domain AsHbC1D5. This work shows that each domain in the 34  
hemoglobin of *Artemia* has different characteristics of ligand binding. 35

© 2015 Published by Elsevier B.V.

### 39 1. Introduction

42 Hemoglobins (Hbs) are a specific class of proteins consisting of a sin- 43  
gle or multiple globin chains. These globin chains display the specific 44  
globin fold consisting of 7 or 8  $\alpha$ -helical segments (indicated A to 45  
H) wrapped around a heme b moiety according to a 3-over-3  $\alpha$ - 46  
helical sandwich pattern [1,2]. The heme iron atom is penta- (F8His) 47  
or hexa-coordinate depending on the presence of an internal 6th ligand 48  
(usually E7His) [3,4]. Comparative studies demonstrate that Hbs occur 49  
in all kingdoms of life and that the canonical globin fold displays an

extreme flexibility [4–7]. Due to the reactivity of the heme iron atom, 50  
globins are involved in a diversity of reactions varying e.g. from O<sub>2</sub> me- 51  
tabolism (O<sub>2</sub> sensing, carrying, storing) to redox chemistry (nitroso and 52  
oxidative stress metabolism) [8]. In some invertebrate classes (*Annelida*, 53  
*Mollusca*, *Crustacea*), Hbs occur as high M<sub>r</sub> proteins dissolved in the ex- 54  
tracellular fluid or hemolymph. Such high M<sub>r</sub> is necessary e.g. to avoid 55  
pigment loss due to excretory filtration events. This high M<sub>r</sub> is obtained 56  
either by disulfide bond based aggregation (e.g. in *Annelida*) or by the 57  
covalently concatenation of globin domains into polymeric globin 58  
chains (e.g. in *Mollusca*, *Crustacea*) [7]. 59

An example of Hbs containing polymeric globin chains are those 60  
from the brine shrimp *Artemia*, a small branchiopod crustacean 61  
inhabiting worldwide diverse ponds with variations in O<sub>2</sub> partial pres- 62  
sure due to difference in salinity (e.g. up to 50% salinity) [9]. *Artemia* ex- 63  
presses genotypically four different globin chains (M<sub>r</sub> ~160,000) Q6  
namely C1, C2, T1 and T2 [10–12]. Structural analyses demonstrate 65  
that the T and C chains are ring-shaped polymers of nine genuine globin 66  
domains covalently joined by short inter-domain linkers (Fig. 1) [13]. 67  
All domains show different primary structures (identities: 23–39%) 68

\* Correspondence to: M. Habibi-Rezaei, School of Biology, College of Science, University of Tehran, Tehran, Iran. Tel.: +98 21 61 11 32 14; fax: +98 21 66 49 29 92.

\*\* Correspondence to: S. Dewilde, University of Antwerp, Universiteitsplein 1, 2610 Wilrijk, Belgium. Tel.: +32 3 265 23 23; fax: +32 3 265 22 48.

E-mail addresses: heshmat.akbariborhani@uantwerpen.be (H.A. Borhani), herald.berghmans@uantwerpen.be (H. Berghmans), stanislav.trashin@uantwerpen.be (S. Trashin), karolien.dewael@uantwerpen.be (K. De Wael), angela.fago@biology.au.dk (A. Fago), luc.moens@uantwerpen.be (L. Moens), mhabibi@ut.ac.ir (M. Habibi-Rezaei), sylvia.dewilde@uantwerpen.be (S. Dewilde).

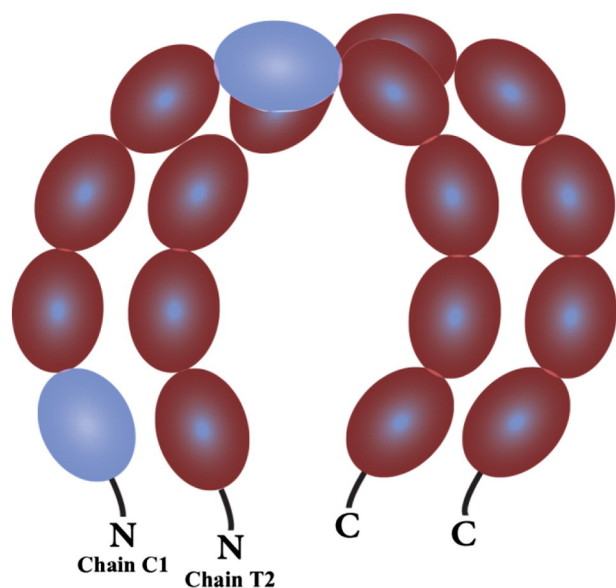


Fig. 1. Schematic view of AsHbII. AsHbC1D1 and AsHbC1D5 are presented in blue.

and are presumed to be copied originally from a single-domain gene [11,14]. Phenotypically, two of these ring-shaped globin chains dimerize, by coaxially stacking, to produce three heterodimeric Hb isoforms ( $M_r \sim 320,000$ ): HbI (C1C2), HbII (C1T2) and HbIII (T1T2) [12]. Analysis of the Hb quaternary structure demonstrates that in both globin chains the EF helices of all domains are in contact along the interpolymer surface, and that domain 1 of the T-polymer aligns with domain 1 of the C-polymer. Similar EF contacts are very common in cooperative Hbs [15].

The *Artemia* Hbs are definitively involved in  $O_2$  storage/transport metabolism and serve e.g. to cope with the variable  $O_2$  tension in the environment. All three Hbs bind  $O_2$  cooperatively and with a different affinity [16]. Their biosynthesis is differentially controlled according to the species where they are expressed in, the ontogenetical stage and the temperature, pH and the  $O_2$  tension of the habitat [10,17–20].

The role of the individual domains in the polymeric globin chain(s) and the native Hb(s) is unclear. However, single or multi-domain fragments of *Artemia salina* (AsHbII), obtained by limited proteolysis, bind  $O_2$  non-cooperatively [21,22].

This paper aims to answer the following questions; firstly, whether the eighteen domains have the same role in the ligand binding or not; secondly, whether the general structure of the heme pocket in the native Hb of *Artemia* is the same as that of myoglobin, and finally, whether Hb of *Artemia*, with 18 heme centers, has a higher redox potential than the normal reference globins.

To provide an answer to these questions, we studied the physico-chemical characteristics (electron paramagnetic resonance (EPR), laser-flash photolysis and redox chemistry) of two recombinant globin domains [A. *salina* chain C1, domains 1 (AsHbC1D1) and 5 (AsHbC1D5) (Fig. 1)] of *Artemia urmiana* and *Artemia franciscana* from Urmia salt Lake, Iran and compared them with those of native Hbs of *A. franciscana* (AfHb).

## 2. Materials and methods

### 2.1. Purification of *A. franciscana* Hbs (AfHb)

Native *A. franciscana* Hbs were purified from frozen material (a gift from the Laboratory of Aquaculture and *Artemia* Reference Center, University of Ghent, Belgium) mainly as described previously [23]. Shortly, crude AfHb (50% ammonium sulfate precipitate) was further purified on a HiTrap DEAE column by step elution at 225 mM NaCl. Hb tracing was spectrophotometric at 412 nm.

### 2.2. RNA extraction and amplification

*A. urmiana* and *A. franciscana* were collected from Urmia Salt Lake, Urmia, Iran. Total RNA was prepared using the combination of TriZol method and PureLink RNA Mini Kit. cDNA was synthesized as described elsewhere [24]. The cDNA fragments encoding globin domains 1 (AsHbC1D1) and 5 (AsHbC1D5) of chain C1 were amplified by PCR [25].

### 2.3. Cloning, expression, and purification of recombinant proteins

The PCR products were cloned into the TOPO-TA vector (Invitrogen) followed by subcloning into pET23a vector. AsHbC1D1 and AsHbC1D5 were expressed in *Escherichia coli* strain BL21(DE3)pLysS. Cells were grown at 37 °C in Terrific Broth (TB) medium (1.2% bactotryptone, 2.4% yeast extract, 0.4% glycerol, 72 mM potassium phosphate buffer, pH 7.5) containing 200  $\mu$ g/ml ampicillin, 30  $\mu$ g/ml chloramphenicol, and 2.5 mM  $\delta$ -amino-levulinic acid. The culture was induced at  $A_{550} = 1.2$  by the addition of isopropyl-1-thio- $\beta$ -D-galactopyranoside (IPTG) to a final concentration of 0.4 mM, and expression was continued overnight (at 25 °C). The cells were harvested and resuspended in lysis buffer, 50 mM Tris-HCl (pH 7.5) containing 300 mM NaCl.

The cells were exposed to three freeze-thaw steps and sonication till completely lysed. The lysate was clarified by low speed (10 min at 10,000  $\times$ g) centrifugation. Then, imidazole was added (final buffer composition of 50 mM Tris-HCl pH 7.5, 300 mM NaCl, and 20 mM imidazole) and the extract loaded on a Ni-affinity His 60 super flow column (Clontech), equilibrated with the same buffer. After washing of the unbound material, the His-tagged recombinant protein was eluted by 50 mM Tris-HCl pH 7.5, 300 mM NaCl and 500 mM imidazole. The fractions containing the proteins of interest were pooled and dialyzed against 50 mM Tris-HCl pH 7.5 containing 150 mM NaCl and 0.5 mM EDTA. After concentration by ultra-filtration (Amicon PM 10), the samples were loaded on Superdex Gf75, 15  $\times$  1800 tricorn column (GE Healthcare) for gel filtration chromatography. All purification steps were assessed by SDS-PAGE.

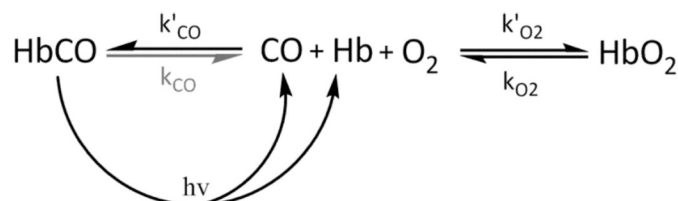
### 2.4. UV-visible spectroscopy

Optical measurements were done with a Varian Cary-5 UV-visible near-infrared spectrophotometer (Varian, Palo Alto, California). All UV-visible spectra were measured in the range from 250 to 700 nm.

### 2.5. Continuous wave EPR of the native protein

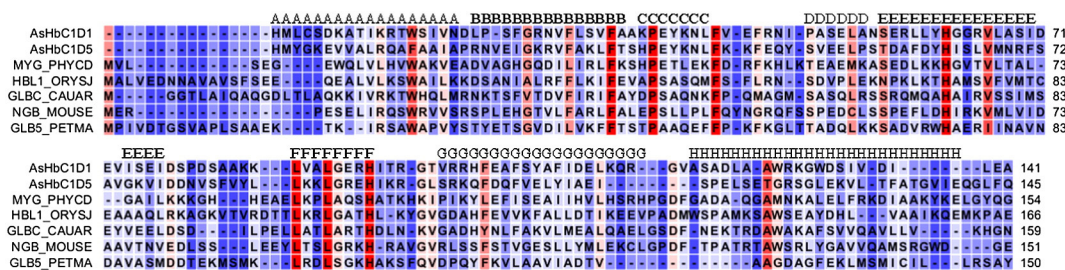
X-band continuous wave (CW) EPR measurements were performed on a Bruker ESP300E spectrometer with a microwave frequency of 9.45 GHz equipped with a gas-flow cryogenic system (Oxford Inc.), allowing for operation from room temperature down to 2.5 K. The magnetic field was measured with a Bruker ER035M NMR Gauss meter. During the experiments, a vacuum pump was attached to the EPR tube in order to remove  $O_2$  from the frozen sample. The spectra are measured with modulation amplitude of 0.8 mT, a modulation frequency of 100 kHz and a microwave power of 0.1 mW.

For the EPR measurements, 20% glycerol was added as a cryoprotectant. All spectra were simulated using EasySpin, a toolbox for MATLAB (Mathworks, Natick, Mass., USA).



Scheme 1.





**Fig. 2.** Alignment of the AsHbC1D1 and AsHbC1D5 with reference hemoglobin/myoglobin. Reference molecules are the globin domains of sperm whale Mb (MYG\_PHYCD, P02185), non-symbiotic Hb 1 of *Oryza sativa*, rice (HBL1\_ORYSJ, O04986), globin C of *Caudina arenicola* (GLBC\_CAUAR, P80018), mouse neuroglobin (NGB\_MOUSE, Q9ER97) and globin 5 of *Petromyzon marinus* (GLB5\_PETMA, P02208).

157 **2.6. Analytical gel filtration experiments**

158 The apparent  $M_r$  of the recombinant AsHbC1D1 and AsHbC1D5 in so-  
 159 lution was assessed by analytical gel filtration experiments using a  
 160 Superdex G27 (2 × 30 cm; buffer 50 mM Tris–HCl pH 7.5, 150 mM  
 161 NaCl and 0.5 mM EDTA; 0.5 ml·min<sup>-1</sup>) calibrated with human Hb  
 162 (68 kD), cytoglobin (42 kD) and myoglobin (17 kD). Protein elution  
 163 was monitored at 280 and 412 nm. Concentration dependent oligomer-  
 164 ization was tested by loading different Hb concentrations (3.3 μM,  
 165 11.1 μM, 42.5 μM and 85.1 μM).

166 **2.7. O<sub>2</sub> equilibria**

167 The O<sub>2</sub> equilibrium curve for AfHb was determined using a modified  
 168 diffusion chamber technique described previously [26,27]. Shortly,  
 169 water-saturated gaseous mixtures of O<sub>2</sub> and ultrapure (>99.998%) N<sub>2</sub>  
 170 created by Wösthoff (Bochum, Germany) gas mixing pumps were  
 171 used to equilibrate a thin smear (190 μm heme) of the AfHb solution  
 172 with different values of O<sub>2</sub> tension ( $P_{O_2}$ ). Changes in absorbance upon  
 173 oxygenation were recorded continuously at 436 nm by a  
 174 photomultiplier (model RCA 931-A, Hamamatsu, Hamamatsu city,  
 175 Japan) and an Eppendorf model 1100 M photometer (Hamburg,  
 176 Germany) coupled to a potentiometric linear recorder. For reference, 0  
 177 and 100% O<sub>2</sub> saturation levels were obtained by equilibrating with  
 178 pure N<sub>2</sub> and O<sub>2</sub>, respectively, at the beginning and end of the experi-  
 179 ment.  $P_{50}$  ( $P_{O_2}$  at half-saturation) and  $n_{50}$  (cooperativity) values were  
 180 calculated from the zero intercept and slope, respectively, of Hill plots:  
 181 log [Y / (1 – Y)] versus log  $P_{O_2}$ , where Y is the fractional saturation of  
 182 Hb. The curve consists of four to five saturation steps. Experiments  
 183 were carried out in duplicate in 100 mM potassium phosphate,  
 184 pH 7.0 at 20 °C.

185 **2.8. Ligand-binding kinetics**

186 **2.8.1. CO binding kinetics**

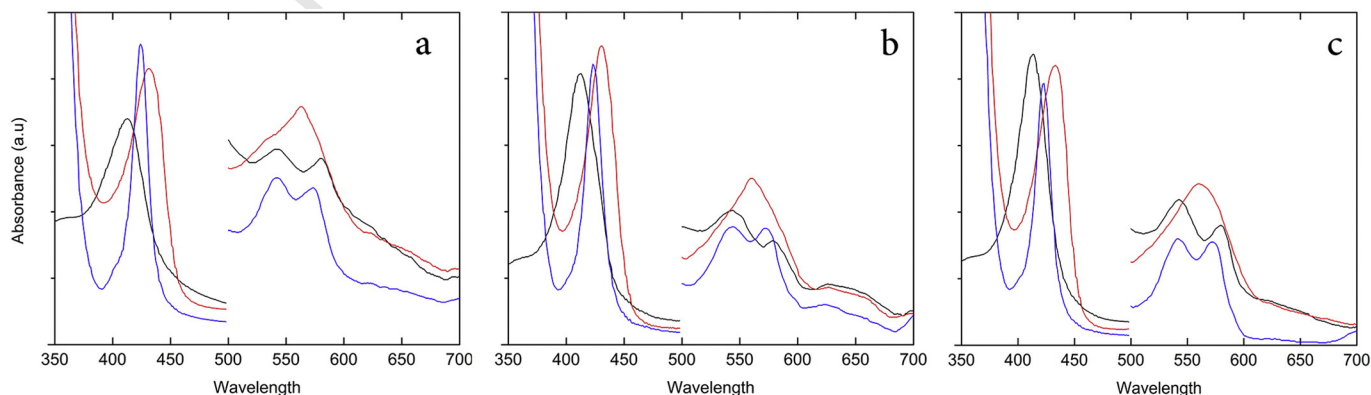
187 To measure the rates of CO association, the Fe<sup>3+</sup> Hb stock solution was  
 188 diluted with a 100 mM potassium phosphate buffer, 1 mM EDTA, pH 7.0,  
 189 at 20 °C to a final concentration of ~5 μM, equilibrated in different concen-  
 190 tration (200–800 μM) of CO and anaerobically reduced with 1 mM sodi-  
 191 um dithionite. The experimental setup was described in detail  
 192 elsewhere [28,29]. Shortly, photolysis of HbCO samples was carried out  
 193 on a laser photolysis system (Edinburgh Instrument LP920) at 20 °C  
 194 using the second harmonic (532 nm) of a frequency-doubled Q-  
 195 switched Nd:YAG laser (Spectra Physics Quanta-Ray). The absorbance  
 196 changes were recorded with a Tektronix TDS220 digitizing oscilloscope  
 197 fitted to a simple exponential expression. CO association rate value ( $k'$   
 198  $co$ ) was calculated as the slope of plots of  $k_{obs}$  versus [CO]. For all  
 199 ligand-binding kinetic studies, measurements were done at least in tripli-  
 200 cate and averaged.

201 **2.8.2. Geminate recombination**

202 The geminate rebinding time courses were fitted to one exponential  
 203 decay. The observed rates of geminate rebinding ( $k_{gem}$ ) were indepen-  
 204 dent of ligand concentration. The fraction of geminate recombination  
 205 ( $F_{gem}$ ) was calculated using Eq. (1):

$$F_{gem} = \Delta A_{gem} / \Delta A_{gem} + y_0 \quad (1)$$

207 where  $\Delta A_{gem}$  represents the absorbance change associated with the internal  
 208 geminate rebinding and  $y_0$  (the offset) is the difference between the  
 209 absorbance after complete geminate recombination and the absorbance  
 210 of the original ground state, which is observed prior to the photolysis or  
 211 at the long times after the complete rebinding of bimolecular from the  
 solvent. As the concluded rates for CO entry and exit appeared to be the



**Fig. 3.** Absorption spectra of AsHbC1D1 (a), AsHbC1D5 (b) and AfHb (c). As purified-oxygenated form (black line), dithionite-reduced Fe<sup>2+</sup> deoxygenated form (red line) and the carbonylated form (blue line).

212 same for those of O<sub>2</sub> and NO due to the almost same size and polarity of  
213 three diatomic gases [30], the geminate rebinding parameters were measured for CO.  
214

### 215 2.8.3. O<sub>2</sub> binding kinetics

216 Time courses for O<sub>2</sub> association and dissociation were measured  
217 after complete laser photolysis of Hb samples containing various mix-  
218 tures of O<sub>2</sub>/CO. The mixed atmosphere allows measurement of the O<sub>2</sub>  
219 dissociation rate. In these experiments, Hb samples were prepared in  
220 five different buffers with 100 mM potassium phosphate buffer,  
221 pH 7.0, at 20 °C containing various concentration of CO and O<sub>2</sub> (e.g.  
222 892 μM of CO, 750 μM of CO and 312.5 μM of O<sub>2</sub>, 500 μM of CO and  
223 625 μM of O<sub>2</sub>, 250 μM of CO and 937.5 μM of O<sub>2</sub>, and 1250 μM of O<sub>2</sub>).  
224 The sample is essentially bound to CO at equilibrium because the parti-  
225 tion coefficient between both ligands is displaced toward the CO form as  
226 occurs for most globins.

227 As shown in Scheme 1, the photolysis of HbCO by a laser beam gener-  
228 ates a deoxyHb species which react rapidly with either ligand and  
229 two distinct phases were observed. The rate of bimolecular rebinding  
230 phase (*k*<sub>fast</sub>) that was dominated by O<sub>2</sub> rebinding due to its larger asso-  
231 ciation constant (*k'*<sub>O<sub>2</sub></sub>), was monitored by a large decrease in the absor-  
232 bance of unliganded Hb at 436 nm [31]. The value of *k*<sub>fast</sub> is given by  
233 Eq. (2),

$$k_{fast} = k'_{O_2}[O_2] + k_{O_2} + k'_{CO}[CO]. \quad (2)$$

235 Generally, the CO dissociation occurs on a timescale slower compared  
236 to the O<sub>2</sub> and then can be treated as an irreversible process.

237 The slow replacement phase represents the displacement of transi-  
238 ently bound O<sub>2</sub> by CO, which, although kinetically less reactive, has a  
239 higher affinity for heme. The time course of this replacement reaction  
240 is best monitored by a large increase in the difference of absorbance at  
241 422–424 nm. The rate of this process, *k*<sub>slow</sub>, is given by Eq. (3),

$$k_{slow} = \frac{k_{O_2}}{1 + \frac{k'_{O_2}[O_2]}{k'_{CO}[CO]}}. \quad (3)$$

243 An iterative, nonlinear, least squares fitting routine (Solver in  
244 Microsoft Excel) was used to optimize the values of *k'*<sub>O<sub>2</sub></sub>, *k*<sub>O<sub>2</sub></sub>, and *k'*<sub>CO</sub>  
245 to give the best fit to the dependence of *k*<sub>fast</sub> and *k*<sub>slow</sub> on [O<sub>2</sub>]/[CO], in-  
246 cluding conditions of [O<sub>2</sub>] = 0 and [CO] ≈ 0 [31].

### 247 2.9. Reduction potential

248 Electrochemical measurements were conducted in a conventional  
249 three-electrode cell using a μAutolab III interface controlled by Nova  
250 1.10 software (Metrohm-Autolab BV, Netherlands). A saturated calomel

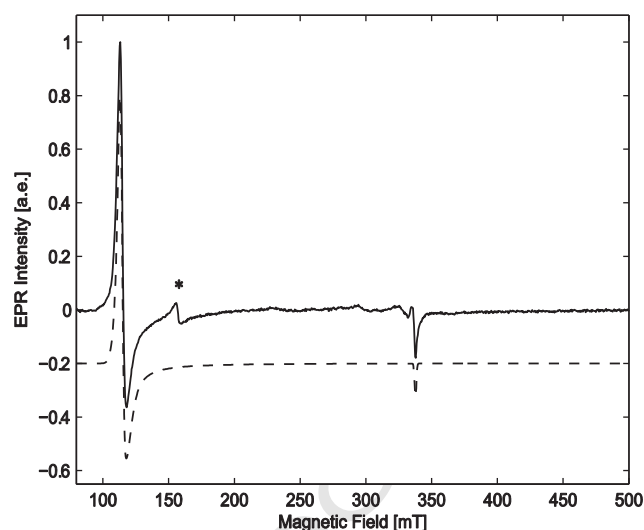


Fig. 4. X-band CW-EPR spectra of ferric native AβHb (solid) and simulation (dashed). The asterisk shows the contribution to the signal of non-heme iron. All spectra were recorded at 10 K.

251 electrode (SCE, 0.248 V versus standard hydrogen electrode (SHE) at  
252 20 °C) and a glassy carbon rod were used as the reference and auxiliary  
253 electrodes, respectively. Prior to modification, the gold electrodes 253  
(1.6 mm in diameter, BASI, USA) were polished with 1 and 0.05 μm alu-  
254 mina slurry, washed, and sonicated in ultrapure water for about 5 min.  
255 Next, the electrodes were electrochemically cleaned by cyclic voltamm-  
256 etry (CV) in 0.5 M NaOH at the potential range from –0.35 to –1.35 V  
257 and in 0.5 M H<sub>2</sub>SO<sub>4</sub> at the potential range from 0.2 to 1.5 V vs. SCE until  
258 repeatable cyclic voltammograms. Then, the electrodes were incubated  
259 overnight in 10 mM 6-mercapto-1-hexanol (97%, Sigma-Aldrich) solu-  
260 tion in water. Finally, the electrodes were thoroughly washed with  
261 water and dried in air stream. Electrochemistry of the proteins was  
262 studied using membrane electrodes as described [32,33]. Briefly, 2–  
263 3 μl of a protein solution was entrapped between the electrode and dial-  
264 ysis membrane (MWCO 12 kD). Differential pulse voltammetry  
265 (DPV) was recorded before and after placing the samples on the  
266 electrodes. 267

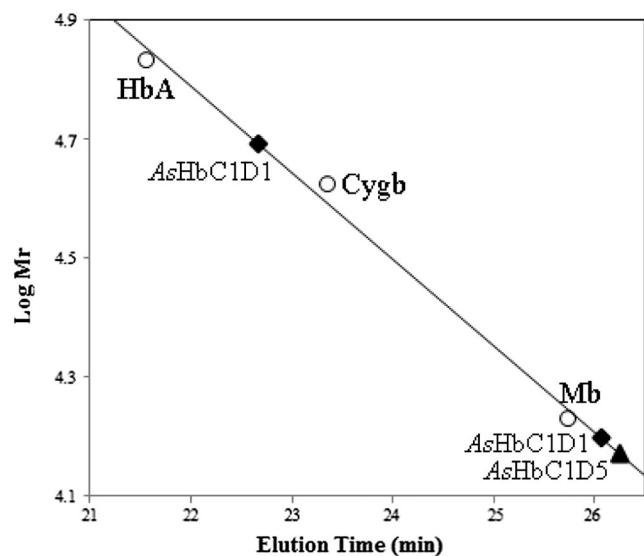


Fig. 5. Analytical gel filtration experiments. Dependence of the molecular mass on elution time; the globin standards (open circles) were human hemoglobin (HbA), human cytoglobin (Cygb), horse heart myoglobin (Mb) along with AsHbC1D1 (◆) and AsHbC1D5 (▲).

Table 1

The absorption maxima of AsHbC1D1, AsHbC1D5 and AβHb in different forms.

	Soret band (nm)	β-Band (nm)	α-Band (nm)
<i>As purified oxy-form</i>			
AsHbC1D1	413	540	579
AsHbC1D5	412	544	579
AβHb	413	540	582
<i>Reduced deoxy-form</i>			
AsHbC1D1	431	Shoulder 528	562
AsHbC1D5	431		559
AβHb	433		563
<i>Carbon monoxy-form</i>			
AsHbC1D1	424	544	573
AsHbC1D5	423	545	571
AβHb	423	540	572

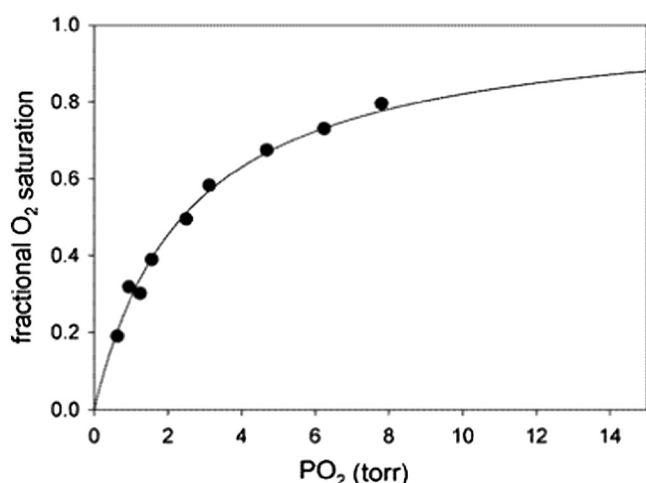


Fig. 6. Oxygen binding equilibrium curve for A/Hb.

### 268 3. Results

#### 269 3.1. Cloning, expression and purification

270 The obtained cDNA sequences for domains 1 and 5 from both species  
 271 were compared with the reported sequence of *Artemia* by Manning et al.  
 272 [10]. For cDNA sequences of domain 1 from *A. urmiana* and  
 273 *A. franciscana* two differences (positions 102 and 393) and for that of  
 274 domain 5 one difference (position 391) were noted. The corresponding  
 275 protein sequences for both domains were exactly the same as reported  
 276 in literature. Therefore, AsHbC1D1 and AsHbC1D5 were used as abbrevi-  
 277 ations to refer to the recombinant proteins of both species. AsHbC1D1  
 278 and AsHbC1D5 are composed of 141 and 145 residues, respectively.  
 279 The primary structure comparison indicated 26.1% identity and 41.8%  
 280 similarity between the domains. Moreover, AsHbC1D1 sequence is  
 281 more similar to the non-symbiotic Hb1 of *Oryza sativa*, rice (38.8%)  
 282 and to the globin C of *Caudina arenicola* (41.0%) whereas AsHbC1D5 is  
 283 similar to the mouse neuroglobin (40.6%) and the globin 5 of  
 284 *Petromyzon marinus*, sea lamprey (39.1%) (Fig. 2). Expression and puri-  
 285 fication of the recombinant AsHbC1D1 and AsHbC1D5 yielded two re-  
 286 combinant proteins with a molecular mass ~16–17 kD, as verified by  
 287 SDS-PAGE and gel filtration. The recombinant proteins were, however,  
 288 not that stable and hard to store. A/Hb was successfully purified from  
 289 the brine shrimp and showed a molecular mass of 160 kD by SDS-PAGE.

#### 290 3.2. UV-visible spectroscopy

291 Fig. 3 shows the UV-visible absorption spectra of expressed/purified  
 292 species of AsHbC1D1, AsHbC1D5 and A/Hb in the oxy-, deoxy- and

carbon monoxy-forms. The optical absorption spectra of the oxy-form 293  
 of AsHbC1D1 and AsHbC1D5 show the Soret maximum at 412 nm and 294  
 $\alpha$  and  $\beta$  maxima at 579 nm and 540 nm, respectively (Fig. 3, Table 1). 295  
 This is a typical absorbance spectrum for oxygenated globins, such as 296  
 sperm whale myoglobin. Moreover, the absorption spectrum of the 297  
 freshly purified A/Hb exhibits the Soret maximum at 413 nm and  $\alpha$  298  
 and  $\beta$  maxima at 579 nm and 540 nm, respectively. 299

Upon CO binding to ferrous AsHbC1D1, AsHbC1D5 and A/Hb, the 300  
 Soret maxima are shifted to 424, 423 and 423 nm, respectively. The  $\alpha$  301  
 and  $\beta$  maxima are observed at 571–573 nm and 540–545 nm (Table 1). 302

#### 303 3.3. Continuous wave EPR of the native Hb

The EPR spectrum of ferric native A/Hb is shown in Fig. 4 and indi- 304  
 cates that the system is in a high-spin (HS) state. In this state, the distal 305  
 side of the central heme iron is known to bind weak ligands like H<sub>2</sub>O or 306  
 have no distal ligand. Simulation of this spectrum gives the following  $g$ - 307  
 values:  $g_x = 5.92$ ,  $g_y = 5.91$  and  $g_z = 1.996$ . After addition of imidazole 308  
 (1:1 ratio imidazole:heme), which is a strong base, we expect imidazole 309  
 to bind to the distal side of the heme iron because of the competition 310  
 with the weak ligand, as is reported for metmyoglobin [34]. The EPR 311  
 spectrum after addition of imidazole however shows no appearance of 312  
 a low-spin state typical for an imidazole-ligated globin. This indicates 313  
 that the imidazole is not able to enter the heme-pocket, and thus native 314  
 A/Hb has a more closed heme-pocket structure than myoglobin. 315

#### 316 3.4. Oligomerization state analysis by gel filtration

Analytical gel filtration experiments were performed to assess the 317  
 quaternary structure of AsHbC1D1 and AsHbC1D5 in solution (expected 318  
 molecular mass = 16.5 and 17.2 kD, respectively, as calculated from se- 319  
 quence analysis). Elution time for AsHbC1D1 (Fig. 5) corresponds to two 320  
 peaks with molecular mass around 50 kD and 16 kD, values consistent 321  
 with a tetramer and a monomer assembly, respectively. The elution 322  
 time for AsHbC1D5 correspond to a molecular mass of 15 kD, which indi- 323  
 cates a monomeric structure (Fig. 5). The elution times for both pro- 324  
 teins were the same for all tested concentrations, which indicates that 325  
 the oligomerization of both proteins is independent of protein 326  
 concentration. 327

#### 328 3.5. O<sub>2</sub> equilibria

A/Hb O<sub>2</sub> binding curve was hyperbolic (Fig. 6) with a  $P_{50}$  value of 329  
 2.46 Torr, while the reported  $P_{50}$  values for *Artemia* Hb I, Hb II and Hb 330  
 III were 5.34, 3.70 and 1.80 mm Hg, respectively [17]. Moreover, the cal- 331  
 culated  $P_{50}$  value based on the association constant of O<sub>2</sub>,  $K^{O_2}$ , was 332  
 1.51 Torr (Table 3). The difference can be explained by the different ex- 333  
 perimental conditions (e.g. pH and temperature). The native Hb showed 334  
 a cooperativity value ( $n_{50}$ ) close to 1.04, which is different from the 335

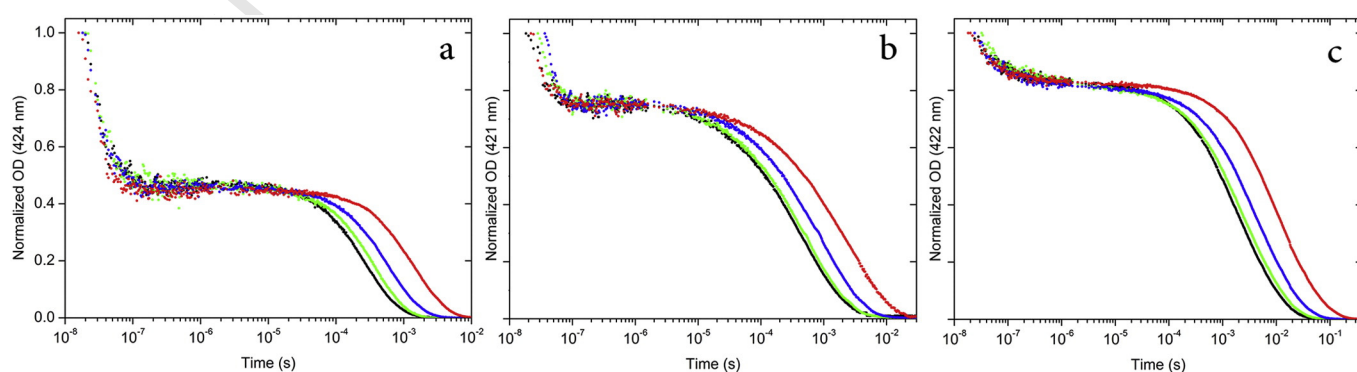


Fig. 7. CO rebinding after nanosecond laser photolysis. CO rebinding kinetics to AsHbC1D1 (a), AsHbC1D5 (b) and A/Hb (c) solution at 20 °C (black, 1000  $\mu$ M CO; green, 800  $\mu$ M CO; blue, 400  $\mu$ M CO; red, 200  $\mu$ M CO).



**Table 2**  
Kinetic values for O<sub>2</sub> and CO reactions of AsHbC1D1, AsHbC1D5 and AfHb at pH 7, 20 °C with some selected hemoglobin/myoglobin.

	$k'_{O_2}$	$k_{O_2}$	$K_{O_2}$	$P_{50}$ , Torr	$k'_{CO}$	$k_{gem}$	$F_{gem}$	$k_{bond}$	$k_{escape}$	$k'_{entry}$	Ref.
	$\mu M^{-1} s^{-1}$	$s^{-1}$	$\mu M^{-1}$		$\mu M^{-1} s^{-1}$	$\mu s^{-1}$		$\mu s^{-1}$	$\mu s^{-1}$	$\mu M^{-1} s^{-1}$	
AsHbC1D1	10.5 ± 1.6	8.1 ± 0.8	1.30 ± 0.18	0.42 <sup>a</sup>	3.32 ± 0.18	55 ± 3	0.57 ± 0.01	31.35 ± 3	23.65 ± 4.2	5.82 ± 0.18	This work
AsHbC1D5	13.2 ± 1.2	15.3 ± 2.1	0.86 ± 0.05	0.64 <sup>a</sup>	15.9 ± 1.6	61 ± 0.7	0.23 ± 0.01	14.03 ± 0.7	46.97 ± 1	69.13 ± 1.6	This work
AfHb	8.1 ± 0.7	35.4 ± 4.9	0.23 ± 0.05	2.39 <sup>a</sup> (2.46 <sup>b</sup> )	1.7 ± 0.35	17 ± 0.6	0.19 ± 0.01	3.23 ± 0.6	13.77 ± 0.84	8.95 ± 0.35	This work
Carp Mb1	7.1	14.9	0.48	1	3.2, 0.9	11				1.05 ± 0.02	[26]
Carp Mb2	16.6	75.2	0.22	1.7	4.4, 2.0	5					[26]
Hb α	32	13	2.6		5.2	24	0.14	3.4	21	36	[30]
Hb β	82	28	3.1		9.2	7.6	0.22	21.7	6	42	[30]
SWMb	16	14	1.14	0.9 <sup>c</sup>	0.7	12.12 <sup>d</sup>	0.47 <sup>d</sup>	5.7 <sup>d</sup>	6.3 <sup>d</sup>	34 <sup>d</sup>	[35–37]
CerHb <sup>e</sup>	230	190	1.2		32	80	0.05	4	76	640	[38]
Chironomus Hb	300	218	1.37		0.27						[39]

<sup>a</sup> These values are calculated based on  $K_{O_2}$  values (i.e.  $P_{50} = 1/(1.82 * K_{O_2})$ ).

<sup>b</sup>  $P_{50}$  value was measured by O<sub>2</sub> equilibrium.

<sup>c</sup> Calculated in 22 °C.

<sup>d</sup> The kinetic values are measured based on O<sub>2</sub> as ligand, whereas those of the recombinant proteins are measured based on CO.

<sup>e</sup> *Cerebratulus lacteus* mini-hemoglobin.

value of the previous report by D'Hondt study [17]. However, they reported the  $n_{50}$  value for each type of *Artemia*'s Hb.

### 3.6. Ligand binding kinetics

#### 3.6.1. CO association rate constant

The CO association rates of AsHbC1D1, AsHbC1D5 and AfHb were measured using Laser Flash Photolysis. Fig. 7 reports the fraction of deoxy heme as a function of time after the photolysis for selected conditions. Two distinct processes in the CO rebinding curves are clear: i) A fast rebinding phase or geminate rebinding, which was completed in less than 2000 ns, and shows no dependence on CO concentration; and ii) a slower phase, whose apparent rate constant depends on CO concentration, and is called bimolecular rebinding. The latter indicates the rebinding of ligands to heme iron from the solvent phase.

As shown in Fig. 7, the ligand recombination in AsHbC1D1 and AsHbC1D5 is remarkably different from each other. AsHbC1D1, showed monophasic kinetics and the extracted association rate constant is 3.32  $\mu M^{-1}$  (Table 2). Furthermore, AsHbC1D1 has slightly higher association rate constant than AsHbC1D5, suggesting a greater accessibility of this domain for CO. CO bimolecular rebinding of both AsHbC1D5 and AfHb, however, shows biphasic kinetics that could be fitted by a double exponential relaxation. Therefore, as indicated in Table 3, the bimolecular CO association rate constants of AsHbC1D5 are 15.9 and 1.3  $\mu M^{-1}$ , and that of AfHb are 1.7 and 0.17  $\mu M^{-1}$ . This kinetic heterogeneity for AsHbC1D5 and AfHb suggests the existence of two different conformations at equilibrium. Moreover, it is noteworthy to mention that, as there are 18 globin domains in AfHb, we cannot make a distinction between the rates of the separate domains when measuring on the native Hb.

#### 3.6.2. Geminate rebinding parameters

Internal CO geminate recombination was examined on nanosecond time scales to first determine the fraction of geminate recombination ( $F_{gem}$ ) and secondly to estimate rates of internal ligand bond formation and escape. The  $F_{gem}$  values of the two recombinant proteins are 0.57 and 0.23, respectively, which indicate a different heme pocket structure or a different escape route in these proteins. The  $F_{gem}$  value of AfHb is more close to the value of AsHbC1D5 (i.e. 0.19) (Fig. 8, Table 2).

The above mentioned kinetic features are shared with sperm whale Mb (SWMb) [35], although there are some small variations in the relative extent of the different phases. In particular, the fraction of geminate recombination for AsHbC1D1 (0.57) is higher than that of SWMb (0.47), whereas  $F_{gem}$  of AsHbC1D5 (0.23) is lower than that of SWMb (Table 2).

However, it should be noted that the  $F_{gem}$  and other kinetic values of SWMb in [35] were measured based on O<sub>2</sub> as ligand.

As shown in Fig. 8, the geminate CO recombination in all three samples appears to be a simple first order process. The  $k_{gem}$  values of AsHbC1D1, AsHbC1D5 are close (i.e. 55 and 61  $\mu s^{-1}$ , respectively) which is nearly 4 times greater than that of AfHb (i.e. 17  $\mu s^{-1}$ ) (Table 2).

The bimolecular and geminate recombination parameters could be used in calculating the internal binding and escaping rates of ligand to or from the inside of the protein.

A two-step binding scheme was assumed for analysis that involved internal bond formation between the ligand and iron atom ( $k_{bond}$ ) and for ligand escape ( $k_{escape}$ ) from or bimolecular return ( $k'_{entry}$ ) to the geminate state (Hb·CO) in Scheme 2 [30]. The rate parameters which fit to these processes, define the observed rates ( $k_{gem}$ , Eq. (5)) and the fractions ( $F_{gem}$ , Eq. (4)) of geminate recombination and the overall bimolecular CO association rate constant ( $k'_{CO}$ , Eq. (6)) [30,40]:

$$F_{gem} = \frac{k_{bond}}{k_{bond} + k_{escape}} \quad (4)$$

$$k_{gem} = k_{bond} + k_{escape} \quad (5)$$

$$k'_{CO} = \frac{k'_{entry}k_{bond}}{k_{escape} + k_{bond}} = k'_{entry}F_{gem} \quad (6)$$

The rate constant for the ligand entry,  $k'_{entry}$ , into the individual proteins is calculated empirically as the observed bimolecular rate constant divided by the fraction of geminate recombination ( $k'_{entry} = k'_{CO} / F_{gem}$ ). Therefore,  $k'_{entry}$  is calculated independently [30,40].

The values of internal binding, ligand entry and escaping parameters for studied proteins are listed in Table 3. As it is clear from Table 3, the  $k_{bond}$  values for AsHbC1D1, AsHbC1D5 and AfHb are 31.35, 14.03 and 3.23  $\mu s^{-1}$ , respectively, while the  $k_{escape}$  values are 23.65, 46.27 and 13.77  $\mu s^{-1}$ , respectively. It must be noted that the  $k'_{entry}$  for AsHbC1D1

**Table 3**  
Reduction potential of AfHb, AsHbC1D1 and AsHbC1D5 based on the differential pulse voltammograms.

	Ep, reduction (V vs. SCE)	$E_{1/2}$ (V vs. SCE) <sup>a</sup>	$E_{1/2}$ (V vs. SHE) <sup>b</sup>
AfHb	-0.361 ± 0.005	-0.374 ± 0.005	-0.126 ± 0.005
AsHbC1D1	-0.399 ± 0.005	-0.412 ± 0.005	-0.164 ± 0.005
AsHbC1D5	-0.426 ± 0.005	-0.439 ± 0.005	-0.191 ± 0.005

<sup>a</sup>  $E_{1/2} = E_{peak} + E_{modul} / 2$ .

<sup>b</sup> Potential of SCE is +0.248 V vs. SHE at +20 °C.

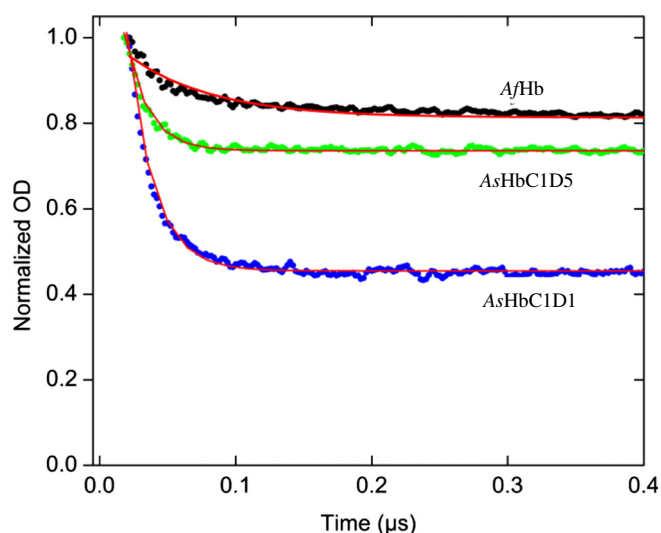


Fig. 8. Time course of CO geminate rebinding to AfHb, AsHbC1D1 and AsHbC1D5.

is  $5.82 \mu\text{M}^{-1} \text{s}^{-1}$ , whereas this parameter for AsHbC1D5 and AfHb has two values (see Table 2).

#### 3.6.3. $\text{O}_2$ association and dissociation rate constant

Table 3 summarizes the kinetic properties of recombinant and native *Artemia* proteins in comparison with those of some reference Hb/Mb.  $\text{O}_2$  association rate constants for AsHbC1D1, AsHbC1D5 and AfHb are 10.5, 13.2 and  $8.1 \mu\text{M}^{-1} \text{s}^{-1}$ , respectively. Although the association rate constants are very similar for all different proteins, the  $\text{O}_2$  dissociation rate constants are not. Indeed the AsHbC1D5 domain has a higher  $k_{\text{O}_2}$  ( $15.3 \text{s}^{-1}$ ) whereas it is even higher for AfHb ( $35.4 \text{s}^{-1}$ ). The  $\text{O}_2$  dissociation rate constants of AsHbC1D5 and AfHb are respectively a factor of 2 and 4 higher, which makes the affinity for  $\text{O}_2$  for these two proteins lower.

Calculated  $P_{50}$  values of AsHbC1D1 and AsHbC1D5 are 0.42 and 0.64 Torr, respectively. The calculated  $P_{50}$  for the native protein is 2.39 Torr, which is very close to the measured  $P_{50}$  by  $\text{O}_2$  equilibria (2.46 Torr). Therefore, the  $P_{50}$  value for native Hb is significantly higher and  $\text{O}_2$  affinity is lower in comparison with the recombinantly expressed domains.  $\text{O}_2$  binding constant of AsHbC1D1 ( $1.30 \mu\text{M}^{-1}$ ) is similar to *Chironomus thummi thummi* Hb [39] and the values of AsHbC1D5 ( $0.86 \mu\text{M}^{-1}$ ) is more like to Mb2 of carp [26] (Table 2)

#### 3.7. Reduction potential

Fig. 9 shows differential pulse voltammograms of the native and recombinant proteins. The redox potential is found from the peak position and differs noticeably among the proteins. As shown in Table 4, AfHb has

the highest reduction potential  $E_{1/2}$  ( $-0.13 \text{V}$  versus SHE) compared to AsHbC1D1 ( $-0.16 \text{V}$ ) and AsHbC1D5 ( $-0.19 \text{V}$ ).

#### 4. Discussion

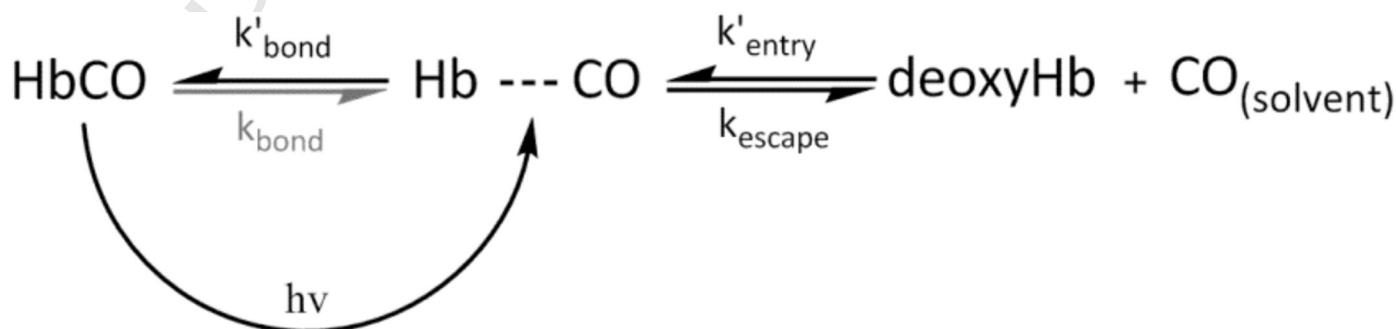
This present study was conducted to examine the role of the different domains in ligand binding of *Artemia* Hb by comparing the characteristics of the isolated domains to the native molecule. By cloning and expressing the two individual domains (AsHbC1D1: a N-terminal located domain; and AsHbC1D5: an internally located domain) separately, we have created one-domain, single-heme globin-like proteins, which mediate ligand binding.

When comparing the C1D1 domain of *A. franciscana* and *A. urmiana* there is no difference in the sequence at the protein level (Fig. 2) and only few differences at the DNA level. The same conclusion can be taken for the C1D5 domain. This suggest that strain specific sequence differences are negligible.

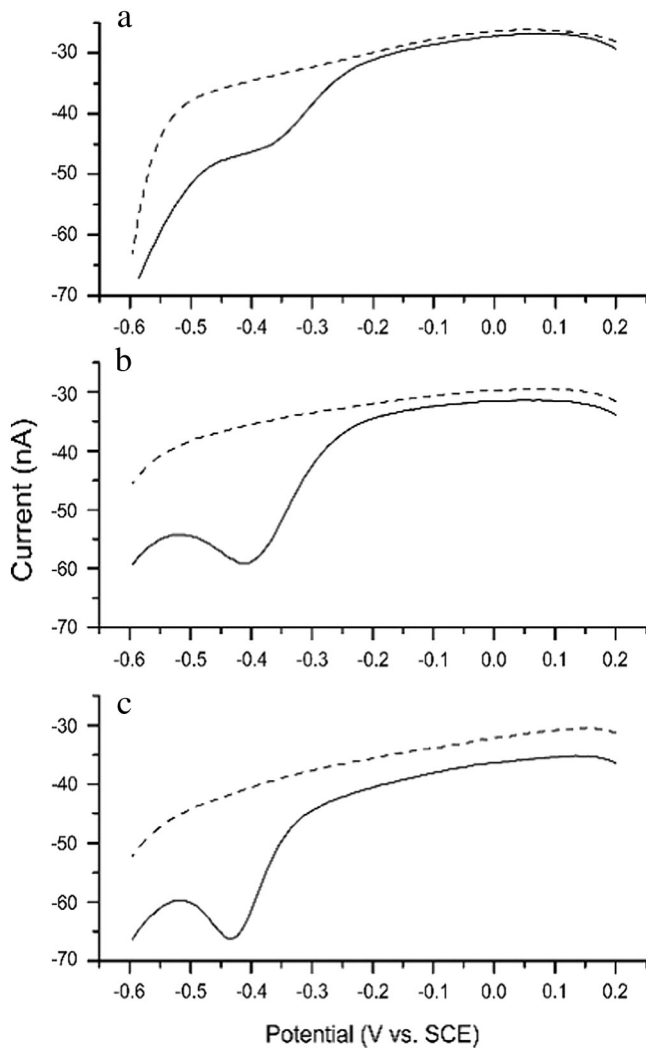
Although there is a low identity between both recombinant domains and SWMb (~25%) virtually all key hydrophobic and heme-binding residues are conserved suggesting that both domains display the genuine globin fold (Fig. 2). A first indication that confirms this, is the UV-Vis spectroscopy, which shows that both recombinant domains as well as the native AfHb, are pentacoordinated like SWMb (Fig. 3). EPR spectroscopy, however, demonstrates that ferric native AfHb is unable to bind imidazole into the heme pocket suggesting a more closed structure than seen in SWMb (Fig. 4).

The quaternary structure of both recombinant globin domains in solution was studied by gel filtration experiments. They show that AsHbC1D1 occurs predominantly as tetramer and monomer whereas AsHbC1D5 occurs only as monomer. This suggests that the function of both domains in the assembly of the native AsHb molecule is different (Fig. 5). Furthermore as each globin domain separately is not stable in physiological buffers, the polymeric structure is essential to stabilize the Hb structure in *Artemia*. Hence, this confirms that the concatenation of globin domains into polymeric globin chains, as mentioned by Weber et al., is indispensable for stability [7].

The ligand binding kinetics and equilibrium studies of the native AfHb and the recombinant domains C1D1 and C1D5 also indicate slight differences in properties. The  $P_{50}$  value of 2.46 Torr obtained by equilibrium measurements is very close to the  $P_{50}$  calculated from the kinetic rate constants (2.39 Torr). They are, however, different from the published values (17) probably due to the fact that the AfHb we used is a mixture of the three Hb types and the difference in the experimental conditions (e.g. pH and temperature) used (Fig. 6; Table 3). Calculated  $P_{50}$  values of AsHbC1D1 and AsHbC1D5 were 0.42 and 0.64 Torr, respectively. The difference in  $\text{O}_2$  affinity is due to a 1.88-fold higher  $\text{O}_2$  dissociation rate constant ( $k_{\text{O}_2}$ ) of AsHbC1D5 compared to AsHbC1D1, whereas the  $\text{O}_2$  association rate constant ( $k'_{\text{O}_2}$ ) of AsHbC1D5 is only 1.25-fold higher than AsHbC1D1 (Tables 2, 4). Furthermore, the  $\text{O}_2$  association equilibrium constant of the native protein is lower than both







**Fig. 9.** Differential pulse voltammograms of AfHb (a), AsHbC1D1 (b) and AsHbC1D5 (c). Broken lines show background of the electrodes recorded before placing the proteins.  $E_{\text{modul}} = -25$  mV,  $E_{\text{step}} = -5$  mV;  $t_{\text{modul}} = 50$  ms,  $t_{\text{step}} = 0.5$  s. Measuring buffer was 10 mM HEPES (pH 7) containing 0.1 M NaCl.

domains ( $K_{O_2} = 0.23 \mu\text{M}^{-1}$ ) resulting mostly again from a higher dissociation rate constant ( $k_{O_2} = 35.42 \text{ s}^{-1}$ ). Therefore, the  $P_{50}$  value for native AfHb is significantly higher and thus  $O_2$  affinity lower in comparison with the recombinant expressed domains. Nonetheless, the calculated and measured oxygen affinity is moderate and in the range of that of Mb.

**Table 4**  
Kinetic values<sup>a</sup> of AsHbC1D1 and AsHbC1D5 in comparison with wild type SWMb and its some point mutants based on [35].

	$k'_{O_2}$	$k_{O_2}$	$K_{O_2}$	$k_{\text{gem}}$	$F_{\text{gem}}$	$k_{\text{bond}}$	$k_{\text{escape}}$	$k'_{\text{entry}}$	Common with	Position of the mutation
	$\mu\text{M}^{-1} \text{ s}^{-1}$	$\text{s}^{-1}$	$\mu\text{M}^{-1}$	$\mu\text{s}^{-1}$		$\mu\text{s}^{-1}$	$\mu\text{s}^{-1}$	$\mu\text{M}^{-1} \text{ s}^{-1}$		
SWMb WT	16	14	1.14	12.12	0.47	5.7	6.3	34		
AsHbC1D1	$10.5 \pm 1.6$	$8.1 \pm 0.8$	$1.30 \pm 0.18$	$55 \pm 0.07$	$0.57 \pm 0.01$	$31.35 \pm 3$	$23.65 \pm 4.2$	$5.82 \pm 0.18$	AsHbC1D1	Second Shell of distal Residue
V66G	25	14	1.79	27.8	0.58	14	9.8	43	AsHbC1D1	Second Shell of distal Residue
T69R	12	11	1.08	13.3	0.47	6.3	7.0	26	AsHbC1D1	Second Shell of distal Residue
AsHbC1D5	$13.2 \pm 1.2$	$15.3 \pm 2.1$	$0.86 \pm 0.05$	$61 \pm 0.7$	$0.23 \pm 0.01$	$14.03 \pm 0.7$	$46.97 \pm 1$	$69.13 \pm 1.6$ $5.65 \pm 0.05$		
G65I	17	32	0.52	13.5	0.49	6.6	6.9	34	AsHbC1D5	Second Shell of distal Residue
I99V	18	8.8	2.1	16.1	0.74	8.8	7.3	18	AsHbC1D5	Proximal side of heme
F138A	24	20	1.2	7.1	0.50	3.5	3.6	48	AsHbC1D5	Proximal side of heme
Q8V	15	30	0.49	9.1	0.41	3.7	5.4	36	AsHbC1D5	Distant from heme
W14F	16	24	0.66	9.5	0.43	3.9	5.15	36	AsHbC1D5	Distant from heme

<sup>a</sup> The kinetic values of the SWMb and its mutants are measured based on  $O_2$  as the ligand, whereas those of the recombinant protein are measured based on CO.

When comparing the CO-ligand binding properties of the recombinant domains we see differences between themselves as well as with the native AsHb molecule. Indeed AsHbC1D1 shows monophasic CO binding kinetics, while AsHbC1D5 shows a double exponential relaxation. This points to a heterogeneity of the AsHbC1D5 domain and may point to the existence of two conformations. It is noteworthy that the CO association rate of AsHbC1D1 is 1.3-fold higher than AsHbC1D5 (Table 2). Furthermore, from the geminate rebinding, we can conclude that there is a different fraction of geminate recombination for the proteins measured (Fig. 8).

Scott et al. studied the effect of point mutations in different positions of SWMb and compared the kinetic values of wild type and mutated proteins [35]. Some mutations in SWMb are in the same position as in AsHbC1D1 or AsHbC1D5 and are listed in Table 4. As an example, in position 65, the residue is Gly in SWMb and Ile in AsHbC1D1. In Table 4, the kinetic values of mutation Gly to Ile in SWMb are compared with that of AsHbC1D1. The comparison of the kinetic values of the recombinant domains with SWMb shows that  $K_{O_2}$  and  $F_{\text{gem}}$  of AsHbC1D1 are more close to the wild type myoglobin than those of AsHbC1D5. Mutation in the second shell of distal residue, such as Val66Gly (smaller residue) increases the  $F_{\text{gem}}$  from 0.47 to 0.58 which is finally similar to  $F_{\text{gem}}$  of AsHbC1D1 (0.57). This also affects the  $K_{O_2}$  value and increases it from 1.14 to  $1.79 \mu\text{M}^{-1}$  while the other mutation in this shell (Thr69Arg) does not vary the  $F_{\text{gem}}$ . Moreover, the point mutation Gly65Ile in the wild type SWMb, changes the  $K_{O_2}$  of SWMb (from 1.14 to  $0.52 \mu\text{M}^{-1}$ ) closer to the  $K_{O_2}$  of AsHbC1D5 ( $0.86 \mu\text{M}^{-1}$ ) while the  $F_{\text{gem}}$  does not change significantly (from 0.47 to 0.49). Hence, it shows that the position 65 may have a significant role in  $K_{O_2}$ . In other words, changing a small residue to a bigger one decreases the  $K_{O_2}$  value by  $0.62 \mu\text{M}^{-1}$ . The other mutation in the distant place of heme such as Gln8Val modifies the  $K_{O_2}$  of SWMb from 1.14 to  $0.49 \mu\text{M}^{-1}$  and closer to  $K_{O_2}$  of AsHbC1D5 ( $0.86 \mu\text{M}^{-1}$ ) (Table 4). It should be considered that as the tertiary structure of the domains is not available, the implementation of the functional and structural relationship may not be totally accurate.

Furthermore, as it was mentioned above, the value of  $k'_{\text{entry}}$  is defined experimentally as the rate constant for the ligand entry into the Hb. It is calculated as the observed bimolecular rate constant ( $k'_{\text{CO}}$ ) divided by the observed total fraction of the internal rebinding ( $F_{\text{gem}}$ ) (Eq. (6)). As it was also shown in Table 4,  $k'_{\text{entry}}$  for AsHbC1D1 ( $5.84 \mu\text{M}^{-1} \text{ s}^{-1}$ ) is 6-times less than the value of SWMb, whereas the value for AsHbC1D5 ( $69.13 \mu\text{M}^{-1} \text{ s}^{-1}$ ) is twice as big as the value of SWMb ( $34 \mu\text{M}^{-1} \text{ s}^{-1}$ ). This means that the two domains show variant ability for the ligand to enter the protein compared to SWMb. The rate of ligand entry in the AsHbC1D5 is higher than that in the SWMb and shows that the ligand can more easily enter the heme pocket of the domains. The lower  $k'_{\text{entry}}$  of the native AfHb confirms the closer heme-pocket structure seen using EPR. It must be noted that the values for SWMb are measured based on  $O_2$  as the ligand, whereas we have used CO as ligand.

The rates of ligand exit from the interior of the protein or  $k_{\text{escape}}$  for AsHbC1D1 and AsHbC1D5 are 3.7- and 7.5-folds, respectively, higher than that of SWMb, which indicates that the ligand escapes faster from the inside of the domains than SWMb. To find the possible structural description for this behavior, we considered that ligands in myoglobin, are trapped in a “webbing” pocket of the distal site, and are surrounded by residues 28, 29, 32, 68, and 107 [35]. The size of this pocket determines the ligand entry equilibrium constant of the protein ( $K_{\text{entry}}$ ), which is mathematically defined as  $k'_{\text{entry}} / k_{\text{escape}}$ . In Table 5, residues of these certain positions in AsHbC1D1, AsHbC1D5 and reference SWMb, as well as the corresponding values are summarized. As shown in the table, the smaller the residues, the lower the  $K_{\text{entry}}$ , hence, less ligand entry into the protein. If we compare the residues, we encounter that the residue in position 68 is the same in all proteins, and the size of residues in positions 28 and 29 in the recombinant proteins (Val and Phe) are almost equal with the size of corresponding residues in SWMb (Ile and Leu). Therefore, the difference in  $K_{\text{entry}}$  values is related to the size of the residue in positions 32 and 107. More specifically, in AsHbC1D1 these positions are occupied by smaller residues (Val and Ser, respectively), where the ligand entry constant is the lowest ( $0.24 \text{ M}^{-1}$ ). In contrast, in SWMb due to the large size of residues in these two positions (Leu and Ile, respectively), the  $K_{\text{entry}}$  is the higher ( $5.39 \text{ M}^{-1}$ ). Moreover, in AsHbC1D5, the size of the residues and the  $K_{\text{entry}}$  value ( $1.47 \text{ M}^{-1}$ ) are moderate. This is in accordance with Scott's proposal [35]. In addition to the size, the nature and orientation of the residues play an important role in the ligand entry and binding to the proteins. To further elucidate this, more structural and mutational analysis is necessary.

As given in Fig. 9 and Table 3, the direct electrochemical measurements reveal variation in the reduction potential ( $E_{1/2}$ ) of the studied proteins. Interestingly, the reduction potential for AfHb with 18 globin domains ( $E_{1/2} = -0.13 \text{ V}$  versus SHE) is the highest, whereas the value of AsHbC1D1 ( $-0.16 \text{ V}$ ), existing majorly in tetramer quaternary structure (Fig. 5), lies in between those of AfHb and the single domain AsHbC1D5 ( $-0.19 \text{ V}$ ). Thus, the higher number of globin domains stabilizes the iron in the heme in the  $\text{Fe}^{2+}$  state, i.e. 18 globin domains stabilize  $\text{Fe}^{2+}$  on about 1.5 kcal/mol ( $\Delta G_{\text{ET}} = -nF\Delta E$ ) compared to the single domain. The latter can be important for minimization of methemoglobin ( $\text{Fe}^{3+}$ ) formation, which is unable to bind molecular  $\text{O}_2$ .

## 5. Conclusion

To provide explicit answers to the above-mentioned questions, we cloned, expressed and purified two different recombinant globin domains in *E. coli* and purified the native Hb from the frozen animal. We utilized electron paramagnetic resonance, laser-flash photolysis and redox chemistry to investigate the physicochemical characteristics of two recombinant globin domains (AsHbC1D1 and AsHbC1D5) and the native one (AfHb). Particularly, we focused on the comparison of the ligand binding kinetics, heme pocket structure and redox potential. UV-visible spectroscopy and analytical gel filtration of the recombinant proteins indicated single-heme globin-like proteins, which mediate ligand binding. However, there were difficulties in stability of AsHbC1D5 in vitro. Moreover, the gel filtration indicated that there is equilibrium

**Table 5**

Ligand entry constant ( $K_{\text{entry}}$ ) dependency on residue in AsHbC1D1 and AsHbC1D5 in comparison with SWMb based on [35].

	Positions					$k'_{\text{entry}}$ $\mu\text{M}^{-1} \text{ s}^{-1}$	$k_{\text{escape}}$ $\mu\text{s}^{-1}$	$K_{\text{entry}}$ $\text{M}^{-1}$
	28	29	32	68	107			
AsHbC1D1	Val	Phe	Val	Val	Ser	$5.82 \pm 0.18$	$23.65 \pm 4.2$	0.24
AsHbC1D5	Val	Phe	Leu	Val	Val	$69.13 \pm 1.6$	$46.97 \pm 1$	1.47
						$5.56 \pm 0.05$		
SWMb <sup>a</sup>	Ile	Leu	Leu	Val	Ile	$34 \pm 7$	$6.3 \pm 1$	5.39

<sup>a</sup> The kinetic values are measured based on  $\text{O}_2$  as ligand, whereas those of the recombinant protein are measured based on CO.

of tetramer and monomeric structure in AsHbC1D1, while the AsHbC1D5 just appears in monomeric structure.

Continuous Wave EPR showed that the heme pocket structure of the AfHb is more closed than the reference myoglobin. Furthermore, the results of difference pulse voltammetry indicated a difference between the redox potential for recombinants and the native protein, emphasizing that the more the globin domains, the higher the redox potential.

In order to study the kinetics in more details, we calculated the association rate of CO binding together with geminate rebinding parameters in pseudo first order condition and the association and dissociation of  $\text{O}_2$  in double replacement condition. The results showed that the AsHbC1D1 has monophasic binding to CO, while the CO association to AsHbC1D1 and AfHb is biphasic. It is also indicated that AsHbC1D1 has a slightly higher oxygen affinity than AsHbC1D5. It shows that they have a diverse role in ligand binding. When studying the full native molecule using ligand binding kinetics, we are faced with a large heterogeneity due to the presence of at least 18 globin domains. However the overall measurements would support an average oxygen affinity. Taken into account the fact that *Artemia* live in circumstances of variations in  $\text{O}_2$  partial pressure due to difference in salinity, we would have expected a larger oxygen affinity. For example, it was reported that *Ascaris* Hb [41,42], due to its unique organization of the B10Tyr and E7Gln, has a lower  $\text{O}_2$  dissociation rate and hence a higher affinity for  $\text{O}_2$  to support life under low oxygen pressures. Hence, our results demonstrate that the AfHb in contrast to *Ascaris* Hb supplies the  $\text{O}_2$  requirement just by increasing the number of the ligand binding domains not by increasing the  $\text{O}_2$  affinity. Indeed, the affinity is similar to SWMb.

## Transparency documents

The Transparency documents associated with this article can be found, in online version.

## Acknowledgments

This work was supported by the Ministry of Science, Research and Technology of I. R. Iran and by the University of Antwerp.

## References

- G. Fermi, M.F. Perutz, B. Shaanan, R. Fourme, The crystal structure of human deoxyhaemoglobin at 1.74 Å resolution, *J. Mol. Biol.* 175 (1984) 159–174.
- J.C. Kendrew, R.E. Dickerson, B.E. Strandberg, R.G. Hart, D.R. Davies, D.C. Phillips, V.C. Shore, Structure of myoglobin: a three-dimensional Fourier synthesis at 2 Å resolution, *Nature* 185 (1960) 422–427.
- D. Bashford, C. Chothia, A.M. Lesk, Determinants of a protein fold. Unique features of the globin amino acid sequences, *J. Mol. Biol.* 196 (1987) 199–216.
- M. Bolognesi, D. Bordo, M. Rizzi, C. Tarricone, P. Ascenzi, Nonvertebrate hemoglobins: structural bases for reactivity, *Prog. Biophys. Mol. Biol.* 68 (1997) 29–68.
- S.N. Vinogradov, D. Hoogewijs, X. Bailly, R. Arredondo-Peter, J. Gough, S. Dewilde, L. Moens, J.R. Vanfleteren, A phylogenomic profile of globins, *BMC Evol. Biol.* 6 (2006) 31.
- S.N. Vinogradov, D. Hoogewijs, X. Bailly, R. Arredondo-Peter, M. Guertin, J. Gough, S. Dewilde, L. Moens, J.R. Vanfleteren, Three globin lineages belonging to two structural classes in genomes from the three kingdoms of life, *Proc. Natl. Acad. Sci. U. S. A.* 102 (2005) 11385–11389.
- R.E. Weber, S.N. Vinogradov, Nonvertebrate hemoglobins: functions and molecular adaptations, *Physiol. Rev.* 81 (2001) 569–628.
- S.N. Vinogradov, L. Moens, Diversity of globin function: enzymatic, transport, storage, and sensing, *J. Biol. Chem.* 283 (2008) 8773–8777.
- K.E. Banister, A.C. Campbell, *The Encyclopedia of Aquatic Life, Facts on File, Incorporated, Place Published, 1985.*
- A.M. Manning, C.N.A. Trotman, W.P. Tate, Evolution of a polymeric globin in the brine shrimp *Artemia*, *Nature* 348 (1990) 653–656.
- C.N.A. Trotman, A.M. Manning, L. Moens, W.P. Tate, The polymeric hemoglobin molecule of *Artemia*: interpretation of translated cDNA sequence of nine domains, *J. Biol. Chem.* 266 (1991) 13789–13795.
- C.J. Vandenberg, C.M. Matthews, C.N. Trotman, Variant subunit specificity in the quaternary structure of *Artemia* hemoglobin, *Mol. Biol. Evol.* 19 (2002) 1288–1291.
- C.N. Trotman, A.M. Manning, J.A. Bray, A.M. Jellie, L. Moens, W.P. Tate, Interdomain linkage in the polymeric hemoglobin molecule of *Artemia*, *J. Mol. Evol.* 38 (1994) 628–636.
- C.M. Matthews, C.J. Vandenberg, C.N. Trotman, Variable substitution rates of the 18 domain sequences in *Artemia* hemoglobin, *J. Mol. Evol.* 46 (1998) 729–733.

- 657 [15] D.T. Chyou, V.L. Rawle, C.N. Trotman, Quaternary structure of *Artemia* haemoglobin  
658 II: analysis of T and C polymer alignment and interpolymer interface, *BMC Struct.*  
659 *Biol.* 7 (2007) 26–40.
- 660 [16] G. Wolf, M. Van Pachtenbeke, L. Moens, M.L. Van Hauwaert, Oxygen binding charac-  
661 teristics of *Artemia* hemoglobin domains, *Comp. Biochem. Physiol. B: Comp.*  
662 *Biochem.* 76 (1983) 731–736.
- 663 [17] J. D'Hondt, L. Moens, J. Heip, A. D'Hondt, M. Kondo, Oxygen-binding characteristics of  
664 three extracellular haemoglobins of *Artemia salina*, *Biochem. J.* 171 (1978) 705–710.
- 665 [18] J. Heip, L. Moens, M. Joniau, M. Kondo, Ontogenetical studies on extracellular hemo-  
666 globins of *Artemia salina*, *Dev. Biol.* 64 (1978) 73–81.
- 667 [19] J. Heip, L. Moens, M. Kondo, Effect of concentrations of salt and oxygen on the syn-  
668 thesis of extracellular hemoglobins during development of *Artemia salina*, *Dev. Biol.*  
669 63 (1978) 247–251.
- 670 [20] V. Sugumar, N. Munuswamy, Physical, biochemical and functional characterization  
671 of haemoglobin from three strains of *Artemia*, *Comp. Biochem. Physiol. A Mol. Integr.*  
672 *Physiol.* 146 (2007) 291–298.
- 673 [21] D. Geelen, L. Moens, J. Heip, R. Hertsens, K. Donceel, J. Clauwaert, The structure of  
674 *Artemia* sp. haemoglobins—I. Isolation and characterization of oxygen binding do-  
675 mains obtained by limited tryptic digestion, *Int. J. Biochem.* 14 (1982) 991–1001.
- 676 [22] L. Moens, D. Geelen, M.L. Van Hauwaert, G. Wolf, R. Blust, R. Witters, R. Lontie, The  
677 structure of *Artemia* sp. haemoglobin. Cleavage of the native molecules into func-  
678 tional units by limited subtilisin digestion, *Biochem. J.* 223 (1984) 861–869.
- 679 [23] L. Moens, M. Kondo, Evidence for a dimeric form of *Artemia salina* extracellular he-  
680 moglobins with high-molecular-weight subunits, *Eur. J. Biochem.* 82 (1978) 65–72.
- 681 [24] D. Hoogewijs, E. Geuens, S. Dewilde, L. Moens, A. Vierstraete, S. Vinogradov, J.  
682 Vanfleteren, Genome-wide analysis of the globin gene family of *C. elegans*, *IUBMB*  
683 *Life* 56 (2004) 697–702.
- 684 [25] S. Dewilde, K. Mees, L. Kiger, C. Lechauve, M.C. Marden, A. Pesce, M. Bolognesi, L.  
685 Moens, Expression, purification, and crystallization of neuro- and cytoglobin,  
686 *Methods Enzymol.* 436 (2008) 341–357.
- 687 [26] S. Helbo, S. Dewilde, D.R. Williams, H. Berghmans, M. Berenbrink, A.R. Cossins, A.  
688 Fago, Functional differentiation of myoglobin isoforms in hypoxia-tolerant carp in-  
689 dicates tissue-specific protective roles, *Am. J. Physiol. Regul. Integr. Comp. Physiol.*  
690 302 (2012) R693–R701.
- 691 [27] R.E. Weber, A. Fago, A.L. Val, A. Bang, M.L. Van Hauwaert, S. Dewilde, F. Zal, L. Moens,  
692 Isohemoglobin differentiation in the bimodal-breathing amazon catfish  
693 *Hoplosternum littorale*, *J. Biol. Chem.* 275 (2000) 17297–17305.
- 694 [28] J. Uzan, S. Dewilde, T. Burmester, T. Hankeln, L. Moens, D. Hamdane, M.C. Marden, L.  
695 Kiger, Neuroglobin and other hexacoordinated hemoglobins show a weak tempera-  
696 ture dependence of oxygen binding, *Biophys. J.* 87 (2004) 1196–1204.
- 697 [29] S. Dewilde, A.I. Ioanitecu, L. Kiger, K. Gilany, M.C. Marden, S. Van Doorslaer, J.  
698 Vercruyse, A. Pesce, M. Nardini, M. Bolognesi, L. Moens, The hemoglobins of the  
699 trematodes *Fasciola hepatica* and *Paramphistomum epiclitum*: a molecular biological,  
physico-chemical, kinetic, and vaccination study, *Protein Sci. Publ. Protein Soc.* 17  
700 (2008) 1653–1662.
- [30] I. Birukou, R.L. Schweers, J.S. Olson, Distal histidine stabilizes bound O<sub>2</sub> and acts as a  
702 gate for ligand entry in both subunits of adult human hemoglobin, *J. Biol. Chem.* 285  
703 (2010) 8840–8854.
- [31] J.S. Olson, E.W. Foley, D.H. Mailliet, E.V. Paster, Measurement of rate constants  
705 for reactions of O<sub>2</sub>, CO, and NO with hemoglobin, *Methods Mol. Med.* 82 (2003)  
706 65–91.
- [32] J. Haladjian, P. Bianco, F. Nunzi, M. Bruschi, A permselective-membrane electrode for  
708 the electrochemical study of redox proteins. Application to cytochrome c552 from  
709 *Thiobacillus ferrooxidans*, *Anal. Chim. Acta* 289 (1994) 15–20.
- [33] É. Lojou, P. Bianco, Membrane electrodes can modulate the electrochemical re-  
711 sponse of redox proteins – direct electrochemistry of cytochrome c, *J. Electroanal.*  
712 *Chem.* 485 (2000) 71–80.
- [34] C.P. Scholes, K.M. Falkowski, S. Chen, J. Bank, Electron nuclear double resonance  
714 (ENDOR) of bis(imidazole) ligated low-spin ferric heme systems, *J. Am. Chem.*  
715 *Soc.* 108 (1986) 1660–1671.
- [35] E.E. Scott, Q.H. Gibson, J.S. Olson, Mapping the pathways for O<sub>2</sub> entry into and exit  
717 from myoglobin, *J. Biol. Chem.* 276 (2001) 5177–5188.
- [36] R.E. Cashion, M.E. Vayda, B.D. Sidell, Kinetic characterization of myoglobins from ver-  
719 tebrates with vastly different body temperatures, *Comp. Biochem. Physiol. B*  
720 *Biochem. Mol. Biol.* 117 (1997) 613–620.
- [37] E. Antonini, M. Brunori, Hemoglobin and Myoglobin in Their Reactions With Li-  
722 gands, North-Holland Pub. Co, 1971. (Place Published).
- [38] A. Pesce, M. Nardini, S. Dewilde, L. Capece, M.A. Marti, S. Congia, M.D. Salter, G.C.  
724 Blouin, D.A. Estrin, P. Ascenzi, L. Moens, M. Bolognesi, J.S. Olson, Ligand migration  
725 in the apolar tunnel of *Cerebratulus lacteus* mini-hemoglobin, *J. Biol. Chem.* 286  
726 (2011) 5347–5358.
- [39] G. Amiconi, E. Antonini, M. Brunori, H. Formanek, R. Huber, Functional properties of  
728 native and reconstituted hemoglobins from *Chironomus thummi thummi*, *Eur. J.*  
729 *Biochem. FEBS* 31 (1972) 52–58.
- [40] M.D. Salter, K. Nienhaus, G.U. Nienhaus, S. Dewilde, L. Moens, A. Pesce, M. Nardini,  
731 M. Bolognesi, J.S. Olson, The apolar channel in *Cerebratulus lacteus* hemoglobin is  
732 the route for O<sub>2</sub> entry and exit, *J. Biol. Chem.* 283 (2008) 35689–35702.
- [41] A. Pesce, S. Dewilde, L. Kiger, M. Milani, P. Ascenzi, M.C. Marden, M.L. Van Hauwaert,  
734 J. Vanfleteren, L. Moens, M. Bolognesi, Very high resolution structure of a trematode  
735 hemoglobin displaying a TyrB10–TyrE7 heme distal residue pair and high oxygen  
736 affinity, *J. Mol. Biol.* 309 (2001) 1153–1164.
- [42] E.S. Peterson, S. Huang, J. Wang, L.M. Miller, G. Vidugiris, A.P. Kloek, D.E. Goldberg,  
738 M.R. Chance, J.B. Wittenberg, J.M. Friedman, A comparison of functional and struc-  
739 tural consequences of the tyrosine B10 and glutamine E7 motifs in two invertebrate  
740 hemoglobins (*Ascaris suum* and *Lucina pectinata*), *Biochemistry* 36 (1997)  
741 13110–13121.